



DISTRIBUTION SYSTEM OPERATOR CERTIFICATION MANUAL

Kentucky Department of Environmental Protection

Division of Compliance Assistance

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Certification and Licensing Program

Mission

Promote responsible environmental stewardship.

Goal

Provide operators with the basic knowledge required to manage drinking water, wastewater and solid waste systems.

The Division of Compliance Assistance offers free compliance assistance. Our services are available to all individuals, communities and businesses regulated by the Kentucky Department for Environmental Protection. We want to help you succeed!

Hotline and Website for regulatory, technical or operational concerns

800-926-8111

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Other programs administered by the Division of Compliance Assistance:

Kentucky Excel Program

Kentucky Brownfield Program

Kentucky Environmental Compliance Assistance Program

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Chapter 1: THE DISTRIBUTION OPERATOR

Chapter 1 Objectives

1. Explain the three characteristics related to professionalism and drinking water distribution systems operators (*regulations, ethics, and professionalism*).

Why should I become a certified operator?

Wastewater and drinking water system operators are front-line environmental professionals who ensure the quality of Kentucky's water resources and protect the public's health. Only operators that are certified by the Kentucky Certification and Licensing Branch can be in responsible charge of a wastewater or drinking water system.

Working in the water and wastewater industry can be extremely rewarding as you will be providing a critical service to your community. It just might be one of the most important positions in the world since no one can live without water. It takes knowledgeable, conscientious people to deliver clean, potable water and to ensure that wastewater is treated and returned as clean water to the environment.

Certification Process

Certification is obtained by meeting minimum education and experience requirements, submitting the appropriate forms and fee and by passing the certification examination with at least a 70%. Regulations pertaining to the certification of drinking water and wastewater system operators are located in [401 KAR Chapter 11](#).

System Classifications

Class I	Water distribution system serving a population < 1,500
Class II	Water distribution system serving a population ≥ 1,500 but < 15,000
Class III	Water distribution system serving a population ≥ 15,000 but < 50,000
Class IV	Water distribution system serving a population ≥ 50,000

Regulatory Education and Experience

Class I	High School Diploma or GED <u>and</u> One (1) year of acceptable operation of a water distribution system shall be required.
Class II	High School Diploma or GED <u>and</u> Two (2) years of acceptable operation of a water distribution system shall be required with Six (6) months in a water distribution system serving a population $\geq 1,500$.
Class III	High School Diploma or GED <u>and</u> Three (3) years of acceptable operation of a water distribution system with One (1) year of that experience in a water distribution system serving a population $\geq 1,500$.
Class IV	Baccalaureate degree in engineering, science or equivalent is required and One (1) year of acceptable operation of a water distribution system serving a population $\geq 15,000$.

Operator in Training (OIT) Certifications

Operators in training certifications are available for each certification type. An individual can apply for an OIT license that is one level higher and of the same type as the certification that the individual currently holds. An individual may also apply for an entry level OIT certification. OITs must pass the appropriate operator certification exam and work under the responsible charge of a mentor. To apply for the exam, individuals must submit the following to the Certification and Licensing Branch:

- Education and Experience Documentation Form;
- Registration Form for Exams and Training;
- The appropriate fee; and
- A letter from the applicant's mentor. The letter from the mentor must include:
 - A commitment to oversee the applicant's work after the applicant becomes an OIT;
 - A commitment to mentor the applicant as long as the applicant is under the mentor's direct responsible charge;
 - Verification that the mentor is not currently mentoring any other OITs; and
 - Confirmation that the mentor holds a certification license that is equal to or greater than the certification level required to serve in primary responsibility of the facility where the mentor and prospective OIT works.

A Wastewater Treatment Class I-OIT who operates a wastewater treatment plant owned by the operator that serves only one residence may have primary responsibility for that plant. All other OITs may not be in responsible charge of a facility unless they hold an additional certification license that does not have an OIT designation.

Certification Renewal or Maintenance

Drinking water treatment, distribution and bottled water system certifications expire on June 30 of even-numbered years. Certifications shall remain valid until the expiration date, unless suspended, revoked or replaced by a higher classification certificate before that date. Certificates issued between Jan. 1 and **June 30** of a renewal year will be issued to include the next two-year renewal period. Failure to renew before July 1 of the renewal year will result in the expiration of certification and a late fee assessment. The certificate shall terminate if not renewed on or before December 31 of the year the certification expired. Certified operators with expired certificates shall not be in responsible charge of a drinking water or wastewater facility.

Certified operators (excluding OIT certificates) may renew their license(s) electronically through the cabinet Web site using the [E-Search](#) link or by submitting the Application for Certification Renewal and the appropriate fee to the Division of Compliance Assistance, Certification and Licensing Branch, 300 Fair Oaks Lane, Frankfort, KY 40601.

Certified operators who are designated an Operator in Training may renew a certification without examination if the operator has:

- Satisfied the continuing education requirements;
- Earned the required years of operational experience;
- Submitted an Education and Experience Documentation form verifying his or her experience;
- Submitted a letter of recommendation from a mentor; and
- Submitted a completed Application for Certification Renewal form and the renewal fee to the cabinet or has renewed the certification electronically on the cabinet's Web site.

Drinking water treatment, distribution and bottled water certified operators training hours shall be completed for each renewal during the two (2) year period immediately prior to the certificate expiration date. Certified operators holding a treatment, distribution and/or bottled water certificate shall complete the required number of cabinet-approved training hours for the highest certificate held in lieu of completing the required number of continuing education hours required for both certificates.

- ***Reminder -- Continuing education hours earned prior to certification shall not count toward certificate renewal.***

Operator Ethics – Standards of Professional Conduct for Certified Operators

In order to safeguard the life, health, and welfare of the public and the environment and to establish and maintain a high standard of integrity in the certified operator profession, standards of professional conduct apply to persons certified in accordance with 401 KAR Chapter 11. These standards state:

- (a) A certified operator shall, during the performance of operational duties, protect the safety, health, and welfare of the public and the environment;
- (b) A certified operator shall use reasonable care and judgment in the performance of operational duties;
- (c) If a certified operator's judgment is overruled by an employer under circumstances in which the safety, health, and welfare of the public or the environment are endangered, the certified operator shall inform the employer of the possible consequences;
- (d) A certified operator shall be objective, truthful, and complete in applications, reports, statements, and testimony provided to the cabinet; and
- (e) A certified operator shall ensure the integrity of the samples that the operator collects, prepares, or analyzes so that results shall be a true representation of water quality.

The full set of standards is located in 401 KAR 11:020.

Certified operators who violate the standards in 401 KAR 11:020 are subject to disciplinary actions which include but are not limited to:

- (a) Probation of the operator's certification for a specified period of time, not to exceed one (1) year;
- (b) Suspension of the operator's certification for a specified period of time, not to exceed four (4) years, during which the certification shall be considered void;
- (c) Revocation of the operator's certification;
- (d) Civil or criminal penalties; or
- (e) A combination of the disciplinary actions listed above.

Disciplinary actions are outlined in 401 KAR 11:050, Section 4.

All regulations related to the certification of wastewater and drinking water operators are located in 401 KAR Chapter 11. A copy of the regulations is located in the appendix of this manual and it is recommended that you become familiar with the regulations that govern your profession.

Key Definitions

Operator means a person involved in the operation of a wastewater treatment plant, wastewater collection system, drinking water treatment plant, or drinking water distribution system. (401 KAR 11:001)

Water Distribution System is defined as the portion of the water supply in which water is conveyed from the water plant or other supply point to the premises of a consumer or a system of piping and ancillary equipment, which is owned or operated by an established water system independent of the water supply system from which the potable water is purchased.

Ethics are defined as a code of morality or a system of moral principles governing the appropriate conduct for a person or group.

System Classifications are used in determining regulatory standards and compliance requirements for each drinking water system.

Public water system is a water system for the provision of water to the public for human consumption and the system has at least 15 service connections or regularly serves an average of at least 25 individuals a year.

Community water system is a public water system that serves at least 15 service connections used by year-round residents or regularly serves at least 25 year-round residents.

Non-community water system is a public water system that serves at least 15 service connections used by persons less than year-round or serves an average of at least 25 individuals at least 60 days per year but less than year-round.

Transient non-community water system is a non-community water system that does not regularly serve at least 25 of the same persons over 6 months per year. (Example – gas stations, parks, resorts, restaurants, etc. with their own water systems.)

Non-transient non-community water system is a non-community water system that does not regularly serve at least 25 of the same persons over 6 months per year. Example – schools, factories, hospitals, etc., with their own water system.

Semi-public water system is a water system made available for drinking or domestic use that serve more than three families but does not qualify as a public water system. Semi-public water systems are commercial facilities that serve food or drink to the public and

shall meet the requirements of 401 KAR 8:020. *Semi-public water systems are not required to have certified operators.*

Monthly Operator Report (MOR) is a system's monthly performance report for regulatory compliance. It must be submitted to the Division of Water in Frankfort no later than 10 days after the end of the month for which the report is filed.

KAR – Kentucky Administrative Regulations are the regulations set forth by the Commonwealth of Kentucky, some of which regulate public water systems (Chapter 8, 11).

KRS – Kentucky Revised Statutes are the laws that govern the Commonwealth of Kentucky.

SDWA – Safe Drinking Water Act was passed by the United States Congress in 1974 to protect public health by regulating the nation's public drinking water supply. The law was amended in 1986 and again in 1996 and requires many actions to protect drinking water and its sources: rivers, lakes, reservoirs, springs, and groundwater wells. This act authorizes the United States Environmental Protection Agency (**US EPA**) to set national health-based standards for drinking water to protect against both naturally occurring and manmade contaminants that may be found in drinking water. The federal operator certification requirement was established by this act.

AI # - Agency Interest Number is the number that the Kentucky Environmental Protection Cabinet and the Division of Compliance Assistance (operator certification program) uses to track all activities pertaining to an individual or group involving environmental or regulatory compliance activities.

AWOP – Area Wide Optimization Program is a national program whose goal is to maximize public health protection through optimization of existing conventional surface water treatment utilities. AWOP was developed by the EPA's Technical Support Center as a non-regulatory, voluntary approach to microbial control using optimization tools to focus on operational changes to improve the drinking water quality. Goals are set that go beyond the regulatory requirements for conventional surface water treatment plants for settled water, filtered water and disinfection.

List of Key Acronyms

AOC	Assimilable Organic Carbon – low molecular weight dissolved organic carbon (DOC) that is readily consumed by bacteria leading to growth.
BAC	Biologically Active Carbon

BAT	Best Available Treatment
CPE	Comprehensive Performance Evaluation
C-T	Concentration of disinfectant residual multiplied by contact time
CTAP	Kentucky DOW Comprehensive Technical Assistance Program
CWS	Community Water Supply
D/DBP	Disinfectant and Disinfection By - Product
D/DBPR	Disinfectant and Disinfection By - Product Rule
DBP	Disinfection By- Product
DOC	Dissolved Organic Carbon
DOW	Kentucky Division of Water
EPTDS	Entry Point To Distribution System
GW	Groundwater
GWR	Groundwater Rule
GWUDI	Groundwater Under Direct Influence
HAA₅	Haloacetic Acids (chloroacetic acid, dichloroacetic acid, trichloroacetic acid, bromoacetic acid, dibromoacetic acid)
HPC	Heterotrophic Plate Count (synonymous with Standard Plate Count)
LCR	Lead and Copper Rule
LTESWTR	Long Term Enhanced Surface Water Treatment Rule
LT2	Long Term 2 Enhanced Surface Water Treatment Rule
µg/L	Microgram per Liter
MIEX	Magnetic resin(trademark acronym for a resin specific for DOC removal)
NOM	Natural Organic Matter
ppb	parts per billion
PWS	Public Water System
QA	Quality Assurance
QC	Quality Control
SDWA	Safe Drinking Water Act (as amended in 1996)
SS	Sanitary Survey
TC	Total Coliform
TCR	Total Coliform Rule
THM	Trihalomethane
TNCWS	Transient Non-Community Water System
TOC	Total Organic Carbon

Treated Water	considered no later than the combined filter influent
TTHM	Total Trihalomethanes (chloroform, bromodichloromethane, dibromochloromethane, and bromoform)
USGS	United States Geological Survey
UV	Ultraviolet Light
UV₂₅₄	Ultraviolet wavelength at 254 nanometers
VA	Vulnerability Assessment
VOC	Volatile Organic Contaminant
WHPA	Wellhead Protection Area
WTP	Water Treatment Plant

Ethics

1. noun – the study of moral standards and how they affect conduct.
2. plural noun – a code of morality, a system of moral principles governing the appropriate conduct for a person or group.



Your job as a distribution system operator has a direct impact on the health of the people who consume your water. Failure to obtain a Bac-T sample from a problem area or worse yet, writing down false readings, ruins your reputation, puts your future credibility in jeopardy, puts your career choice at risk, could lead to incarceration and potentially could cause someone who ingests your water to become sick or die. Nationally, more and more certified operators are being prosecuted for unethical or dishonorable behavior in the workplace. Water treatment and distribution in some areas have become a practice of notating the “right” numbers. If the “right” numbers are turned into the regulatory agency, visits from these regulatory agencies would probably not be as frequent or involved as they would be if valid or honest numbers were turned in. Our job is to protect the consuming public and provide potable water to each and every one of them, not satisfy regulatory numbers. Take pride in your work and in your ethical behavior.

Chapter 1 Review

1. Operator certification became federally mandated in _____.
2. A public water system is defined as having _____ service connections or serves at least _____ people year round.
3. Is it possible to obtain a Distribution Class III – OIT certification if the applicant does not currently hold a valid Distribution Class II certification?
4. After June 30th of even-numbered years, an operator has _____ months to renew before their certification terminates.
5. Certified operators who violate the standards in 401 KAR 11:020 are subject to disciplinary actions which include but are not limited to _____; _____; _____; or _____.
6. The federal law that operator certification falls under is the _____.

Answers for Chapter 1 Review

1. 1996
2. 15, 25
3. No
4. 6
5. Probation of the operator's certification for a specified period of time, not to exceed one (1) year;
Suspension of the operator's certification for a specified period of time, not to exceed four (4) years, during which the certification shall be considered void;
Revocation of the operator's certification;
Civil or criminal penalties; or
A combination of the disciplinary actions listed above.
6. SDWA

Chapter 2: WATER QUALITY

Chapter 2 Objectives:

1. Explain pH, DBPs, HPE's, and Turbidity.
2. Understand the steps a distribution system operator must engage in to maintain or improve water quality.
3. Identify the importance of pH as it relates to water quality.
4. Understand chemical reactions.
5. Explain how halogenated disinfection byproducts are formed.
6. Define corrosion and scaling.
7. Explain the purpose of maximum contaminant level goals (MCLGs).

Water quality in the distribution system is initially determined by the effectiveness of the treatment operation that supplies that system. Actually improving the quality of the water after it enters the distribution system is usually not a viable endeavor. The water delivered to the consumer will be of a lesser quality than what it was when it entered the system. How much the water quality degrades is largely the responsibility of the distribution operator.

The challenge of the distribution operator is to **maintain** the quality of the water from the treatment facility to the consumer's tap. This is best accomplished by developing a strategy of developing and maintaining a distribution system that is clean, biologically and chemically stable, while protecting it from events and situations that will degrade the water quality to an unsuitable level.

This is NOT an easy task, and a practical working knowledge of system design, shortcomings, and practical solutions is necessary to maintain water quality to an acceptable level. This will make it necessary for the distribution operator to have a rudimentary knowledge of water chemistry, hydraulics, biology and water quality parameters.

Pathogens

Pathogens are disease causing organisms. Waterborne diseases are usually caused by the presence of **bacteria, viruses or protozoa**. Their origin could be a shortcoming in the treatment process or they could enter a distribution system through a cross-connection, line installation or repair, intrusion, biofilms or an opening to the outside environment. Poor operational and maintenance practices or excessive water age will allow these pathogens to reproduce and could cause serious and even fatal consequences to ensue. Disinfection residuals need to be maintained to control the growth of microorganisms in our distribution system, whether they are pathogenic or not. Large disinfectant demands reduce the residual disinfectant levels and should be investigated immediately. There are some pathogenic organisms that are extremely resistant to most current disinfection practices and allowed to flourish could bring about dire health concerns to our consuming public.

The challenge for distribution operators is to prevent the entry and to minimize the growth of the entities that can harm the health of the people who consume the water that is ingested from our distribution systems. Currently, coliform sampling is conducted to indicate the likely presence of harmful pathogenic organisms. A **Heterotrophic Plate Count** (a.k.a. a **Standard Plate Count**) can measure microbial **bacterial** growth, and used over time can be a viable and useful indicator of distribution water quality.

Common Harmful Pathogenic Organisms Found in Water

Etiological Agent	Community Water Systems ¹		Noncommunity Water Systems ²		Individual Water Systems ⁴		All Systems	
	Outbreaks	Cases	Outbreaks	Cases	Outbreaks	Cases	Outbreaks	Cases
<i>Giardia</i>	11	2,073	5	167	6	16	22	2,256
<i>Cryptosporidium</i> *	7	407,642	2	578	2	39	11	408,259
<i>Campylobacter</i>	1	172	3	66	1	102	5	340
<i>Salmonellae</i> , nontyphoid	2	749	0	0	1	84	3	833
<i>E. coli</i>	3	208	3	39	3	12	9	259
<i>E. coli</i> O157:H7/C <i>jeuni</i>	0	0	1	781	0	0	1	781
<i>Shigella</i>	1	83	5	484	2	38	8	605
<i>Plesiomonas shigelloides</i>	0	0	1	60	0	0	1	60
Non-O1 <i>V. cholerae</i>	1	11	0	0	0	0	1	11
Hepatitis A virus	0	0	1	46	1	10	2	56
Norwalk-like viruses	1	594	4	1806	0	0	3	2400
Small, round-structured virus	1	148	1	70	0	0	2	218
Chemical	18	522	0	0	7	9	25	531

Turbidity

Turbidity is suspended particles in water that may be organic or inorganic by nature. Turbidity levels can be a viable indicator of chemical or microbial growth in a distribution system. Turbidity can be the source of color, taste or odor complaints, as well as providing a shield from disinfection.

Turbidity levels are determined by turbidimeters that measure levels in NTU's (nephelometric turbidity units). NTU's are calculated by measuring the light reflected off of turbidity particles. Particles of different sizes, shapes and colors will reflect light differently. An MCL of 0.3 measured at the entry point of the distribution is required by regulation.

Simplified Diagram of a Pathogen Encapsulated by a Particulate

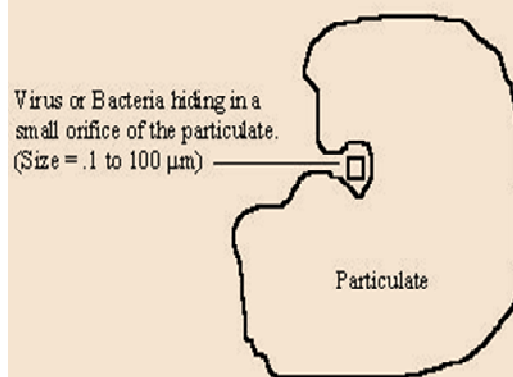


Figure 1 Courtesy of Eric Karch and David Loftis

Higher turbidity levels in the distribution system should be identified and corrected immediately because they could very well be an indication of water quality degradation.

Corrosion/Scaling

Corrosion is the deterioration of a substance due to its environment. In the distribution system, corrosion is the decomposition of metallic components due to the chemical and/or biological characteristics of the water that these components are always in contact with. Scaling is the deposition of calcium carbonate (**CaCO₃**) on distribution piping, valves, meters, etc. This deposition results from chemically unstable water, due primarily to the relative concentrations of pH, alkalinity, total dissolved solids, temperature and hardness.

Disinfection By-products

When chlorine reacts with organics, the chemical reaction forms halogenated disinfection by products. These by products have been found to be carcinogenic or cancer causing. Some of the by products that are regulated are Trihalomethanes, Haloacetic Acids and Chlorite, among others. The formation of these chemical compounds is directly proportionate to the amount of disinfectant, organic carbon and time, occurring concurrently. The most effective means to prevent the formation of these substances is to optimize the entire treatment and distribution processes. The key is the removal of natural organic matter (NOM) entering the distribution system, reduce the disinfectant demand in the system and minimize the residency time or water age in the system. More to the point, we need to practice good operation and control of our systems.

pH

A measure of the hydrogen ion concentration. To put it another way, pH describes the relative acidity of the water. It is measured on a scale from 0 (zero) to 14 with a reading of 7 being neutral. A change in the pH of one number (example: from 5 to 6) is actually a tenfold change in the relative acidity of the water. pH affects disinfection, corrosion control, the deposition of scale and just about every reaction, biological and chemical, that occurs in our water. A basic comprehension of pH can provide a better understanding of why problems occur in some areas but not others.

Taste and Odor

Taste and odor can be the result of improper or incomplete treatment or a biological or chemical phenomenon that occurs in the distribution system. It is frequently associated

with either the decomposition of organic matter or a host of other distribution system issues. Some taste and odor issues are directly attributable to inorganic contaminants, such as corrosion and scaling byproducts or chemical carryover from the treatment facility. Taste and odor issues are more noticeable in warmer weather and/or warmer water temperatures. All taste and odor issues should be reported to the treatment facility immediately or a small problem could accelerate into a large problem in a short period of time. If these issues are not addressed, they will only become worse with time. The best prevention of taste and odor problems is to keep them from occurring.

Hardness

Water that contains a high concentration of calcium or magnesium ions. These complaints should be reported to the treatment facility. Hard water makes the lathering of soap difficult, as well as its propensity to form a layer of scale on dishes, basins and fixtures. The deposition of scale on plumbing fixtures is calcium carbonate (CaCO₃).

Air

Air can restrict the flow or, in some instances, act just like a closed valve in the distribution system. Air in the form of dissolved oxygen, can intensely accelerate the rate of corrosion. Sources of air should be identified immediately to prevent unfavorable results. Air relief valves installed at the high points of a distribution system will alleviate this problem.

Maintaining Water Quality in Distribution Systems

- Proactive unidirectional flushing program
- Storage tank cleaning and maintenance program
- Minimization of residency time (water age)
- Limit dead ends and proactively flush the ones in place
- Disinfection of new and repaired installations
- Viable corrosion control program
- Cross-connection control programs
- Budgeted line replacement program
- Leak detection program
- Water quality monitoring program
- Communication with treatment personnel
- Water quality considerations in design and replacement upgrades
- Training
- Public relations

Complaint Troubleshooting Guide

AIR IN WATER/MILKY WATER	
Possible Causes	
Distribution System	
Shutdown of water mains	
Low pressure in water mains	
Temperature changes	
Cross connections	
Leaking pump glands	
Main breaks	
Fire hydrant use	
Malfunctioning air relief valves	
Premise plumbing	
Overheating of hot water systems	
Warming of cold water lines	
Cross connections	
Zinc	
Complaint Investigation	
Necessary Information	
Location	
Determine pressure zone of complaint	
When was the milky condition first noticed?	
Does the milky appearance dissipate when a glass of water stands for several minutes?	
Is the milky appearance present in both cold and hot water?	
Has the household plumbing recently been turned off?	
Field Investigation	
Check customer's water at premises	
Eliminate air by flushing at hydrant or blow-off	
If necessary, take samples to laboratory	
Check for faulty air relief valves	

Complaint Troubleshooting Guide cont.

DIRTY WATER/COLORED WATER/FOREIGN PARTICLES IN WATER	
Possible Causes	
Distribution Systems	
Treatment facilities	
Main breaks	
Dead ends	

	Cross connections
	New line installation, line repair or refurbishing
	Fire hydrant use
	Customer's service line disturbance
	Changes in pressure zones
	Pipe coatings
	Sand
	Corrosion/scale
	Premise Plumbing
	Hot water systems
	Cross connections
	Scale
	House plumbing system
	Plumbing repairs
	Complaint Investigation
	Necessary Information
	Location
	When was the condition first noticed?
	What is the water's appearance?
	What color is prevalent?
	Are both the hot and cold water affected?
	Is the condition affecting all faucets?
	What color, size and shape are the particles?
	Field Investigation
	Determine pressure zone of premises
	Eliminate complaint by:
	Flushing of hydrants and/or blow offs
	Flush service line
	Take a sample to laboratory if necessary

Complaint Troubleshooting Guide cont.

HARD WATER/SCALE/SPOTS ON GLASSWARE
Possible Causes
Distribution Systems
Water supply changes
Cross connections
Dead ends

Maximum Contaminate Levels (MCL)

The USEPA has established **Maximum Contaminant Levels** for regulated contaminants and pollutants. These levels represent the highest level of a contaminant allowed that would still allow the water to be safe to drink. For instance, the MCL for nitrate is 10.0 mg/l. Water that contains levels higher than 10.0 mg/l of **nitrate** can cause **methemoglobinemia** or **blue baby syndrome**. Public notification must take place if these regulated levels are exceeded. These MCL's are broken down into two categories. Primary MCL's are health-related or, in other words, can cause health problems if water is consumed with levels that exceed the MCL. The second category is secondary MCL's. These contaminants are aesthetic and have no long-term derogatory health issues associated with them.

The EPA has also established Maximum Contaminant Level Goals (MCLG) that represent desirable levels of contaminants that may well be MCL's in the future, dependent upon technology and the costs associated with the removal or reductions of the aforementioned contaminants.

PRIMARY		SECONDARY	
nitrate	10.0 mg/l	iron	0.3 mg/l
lead *	0.05 mg/l	manganese	0.05 mg/l
copper *	1.3 mg/l	sulfates	250 mg/l
fluoride	4.0 mg/l	fluoride	2.0 mg/l

* denotes action levels

ACTION LEVEL for LEAD is .015 mg/l

ACTION LEVEL for COPPER is 1.3 mg/l

There is NO MCL for chlorine, but there is a Maximum Residual Disinfectant Level (MRDL) that is 4.0 mg/l. *Acute Tier 1 Violations require public notification within 24 hours.*

Chapter 2 Review

1. A primary MCL refers to _____ related contaminants.
2. pH refers to the _____ ion concentration of water.
3. Taste and odor issues as well as chemical reactions are exacerbated in _____ water.
4. _____ is an electrochemical process that can cause leaching and other health related problems.
5. DBPs are caused by the reaction of _____ and _____.
6. Hardness is caused by high concentrations of _____ or _____ in the water.

Answers for Chapter 2 Review

1. Health
2. Hydrogen
3. Warmer
4. Corrosion
5. Chlorine, organics
6. Calcium, magnesium ions

Chapter 3: STORAGE

Chapter 3 Objectives

1. Describe the state regulations related to drinking water storage, operation, and maintenance.
2. Explain the difference between the two types of maximum contaminant levels (MCLs).
3. Describe the three types of storage and the associated design features.
4. Explain why excessive water age is one of the most important factors related to water quality deterioration.
5. Explain the two ways to address water age issues.
6. Identify the chemical, biological, and physical issues related to water quality problems.
7. Demonstrate the ability to calculate conversions, area, and volume related to water storage.

Traditionally, finished water storage facilities have been designed to equalize water demands, reduce pressure fluctuations in the distribution system and provide reserves for firefighting, power outages and other emergencies. Some storage facilities have been operated to provide adequate pressure and have been kept full in order to be prepared for emergency conditions. Additionally, some storage facilities have been designed such that the high water level is below the hydraulic grade line of the system, making water turnover extremely difficult. If the hydraulic grade line of the system drops significantly, very old water may enter the system. If tanks are kept full yet are underutilized, the stored water ages and water quality is frequently compromised.

Purpose of Storage: To meet all consumer demands

Storage Quantity: Ensure an adequate supply for fire protection, peak usage periods, distribution maintenance, plant shutdown, major leaks, pressure maintenance, or any other consumer needs.

Quality: Provide pathogen-free water that is also free from inorganic precipitates, biological activity, corrosion byproducts, taste, odor, color, turbidity, disinfection-by-products and any other factor that is detrimental to the health or aesthetic properties of the water.

Under-storage: limits the ability to get the necessary quantity of water to the consumer when needed. Can allow pressure drops during peak usage if sufficient quantities aren't maintained.

Over-storage: can lead to water quality degradation due to the water age or lack of turnover.

Ideally, the operator needs to maintain the minimal amount of water necessary to meet demand on a daily basis in order to prevent water quality degradation and residual chlorine depletion due to a lack of turnover.

Types of Storage

Underground
Ground level
Elevated storage

Design Features

Storage tanks should be:

- Covered to prevent contamination

- Vented to relieve vacuum conditions

- Screened to prevent entry of birds, rodents, etc.

- Screened and flappered overflows that extend to or below ground level and be directed away from the tank foundation, businesses, homes, etc.

- Protected against corrosion and its by-products.

- Built and maintained to ensure structural integrity

Operation and Maintenance

Inspection: coatings (paint), security, safety, corrosion protection, stratification, vents, overflows, telemetry, valve operation, etc. Inspection can be accomplished by draining the facility, diving, remote sensing or float down.

Monitoring: total coliform, disinfectant residuals, pH, temperature, DBP's, bacteria concentration (heterotrophic or standard plate counts), corrosion, scale, nitrification (especially in systems using chloramines), taste, odor, iron, manganese, turbidity, gases, etc.

Cleaning and Disinfection: Draining and cleaning of tanks should be performed when necessary and should be considered on an annual basis. Painting and coatings should be evaluated at least every five to ten years.

Storage Tank Disinfection

Kentucky Administrative Regulations concerning Disinfection of New and Repaired Water Lines:

(1) New Construction Projects and Line Extensions

- (a) Disinfection of water lines. A water distribution system, including storage distribution tanks, or all extensions to existing systems, shall be thoroughly disinfected before being placed in service.
- (b) A water distribution system shall disinfect with chlorine or chlorine compounds, in amounts as to produce a concentration of at **least fifty (50) ppm** and a residual of at least **twenty-five (25) ppm** at the end of **twenty-four (24) hours**, and the disinfection shall be followed by a thorough flushing.
- (c) Other methods and testing procedures that provide an equivalent level of protection may be used if the cabinet grants prior written approval.

(401KAR 8:150, SECTION 4)

Continuous Method

1. Drain and clean tank.
2. As tank is refilling, inject chlorine into the flow to produce 50 ppm.
3. Hold for 24 hours and then measure the free residual (must be > 25 ppm).
4. Drain and dechlorinate the chlorinated water in the tank.
5. Refill with treated water and measure total coliforms and chlorine residual.
6. Return to service *
7. Record and maintain records for a minimum of five (5) years.

Spray/Swab Method

1. Drain and clean tank
2. Fill tank with a measured volume (i.e. 1 ft deep)
3. Dose pre-measured volume at 250 ppm
4. Swab or spray interior walls and ceiling with the chlorine solution.
5. Repeat process no sooner than 1 hour later
6. Allow an additional 30 minutes contact time
7. Drain and dechlorinate the chlorinated water
8. Refill with treated water and measure total coliforms and chlorine residual
9. Return to service *
10. Record and maintain records for a minimum of five years after fulfilling bacteriological sample requirements

** If a new installation is involved, the continuous method must be used and negative coliform results must be obtained before being placed into service.*

Operational Issues

Calibrate and maintain the accuracy of the telemetry and valving systems.

Drawdown tanks as far as feasible before refilling so water turnover is achieved.

Design the tank with a smaller diameter riser to increase velocity and increase mixing.

Contemplate retrofitting older designs with separate inlets and outlets to induce water circulation and mixing.

Water Age

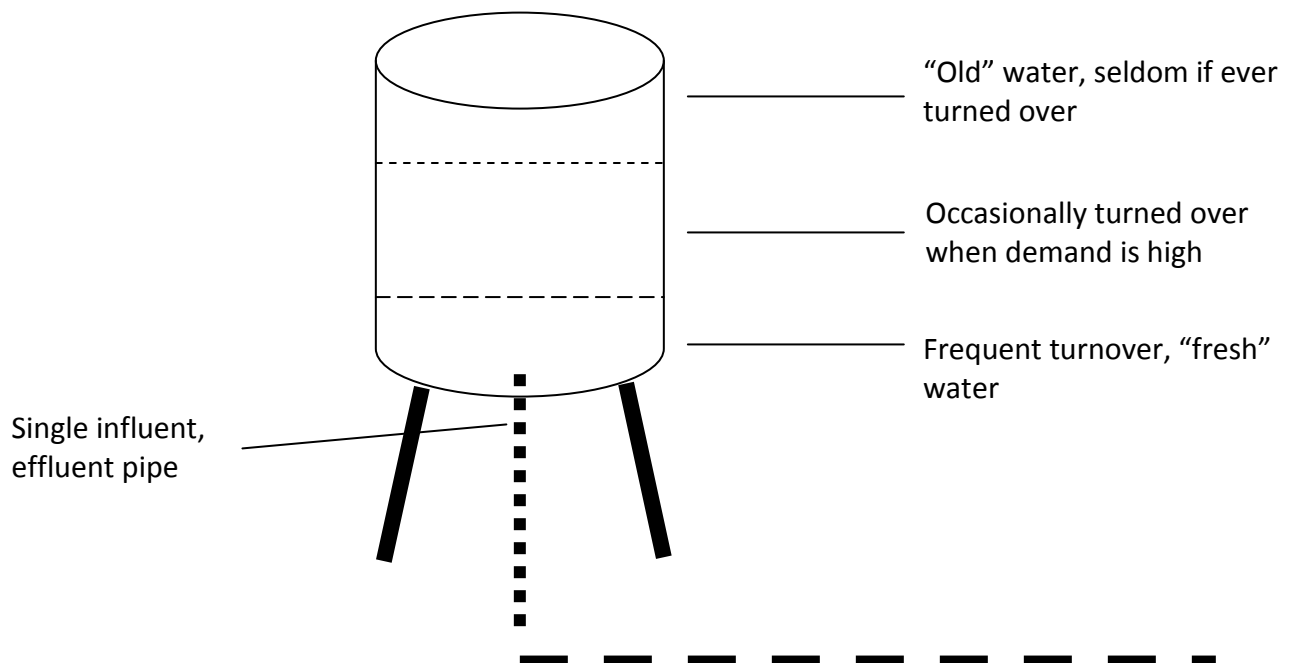
Traditionally many water storage facilities were built with a common inlet and outlet. Water quality problems associated with water storage can be classified as chemical, microbiological or physical.

Excessive water age is probably the most important factor related to water quality deterioration. Long detention times can be conducive to microbial growth and chemical changes. Excessive water age is usually caused by 1) under utilization and 2) short circuiting within the storage facility. Poor mixing (including stratification) can make water quality problems worse by creating zones within the water storage facility where the water age significantly exceeds the average water age throughout your system.

Summary of Water Quality Problems Associated with Finished Water Storage Facilities

CHEMICAL ISSUES	BIOLOGICAL ISSUES	PHYSICAL ISSUES
Disinfectant Decay	Microbial Regrowth*	Corrosion
Chemical Contaminants*	Nitrification*	Temperature/Stratification
DBP Formation*	Pathogen Contamination*	Sediment*
Taste and Odors	Taste and Odors	

* *Water quality issues with direct potential health impact*



Guidelines on Water Turnover Rate

SOURCE	GUIDELINE	COMMENTS
Georgia Environmental Protection Division	Daily turnover goal equals 50% of storage facility volume; minimum desired turnover equals 30% of storage facility volume.	As part of this project, state regulators were interviewed by telephone.
Virginia Dept of Health, Water Supply Engineering Division, Richmond, Va	Complete turnover recommended every 72 hours	As part of this project, state regulators were interviewed by telephone.
Ohio EPA	Required daily turnover of 20%; recommended daily turnover of 25%	Code of state regulations; turnover should occur in one continuous period rather than periodic water level drops throughout the day
Baur and Eisenbart 1988	Maximum 5-7 day turnover	German source, cement based internal surface
Braid 1994	50% reduction of water depth during a 24-hour cycle	Scottish source
Houlmann 1992	Maximum 1-3 day turnover	Swiss source

Information taken from Kirmeyer et al. 1999

Summary

- Microbiological, chemical, and physical water quality problems can occur in finished water facilities due to poor mixing or under – utilization.
- Poor mixing can result from poor design and/or operational practices.
- Health-related water quality issues associated with water age or lack of water turnover:
 - DBP formation
 - Nitrification
 - Pathogen contamination
 - Increases VOC/SOC concentrations
 - Loss of disinfectant residual

Area and Volume

Area: a two-dimensional measurement (length X width). Area is a measurement of the amount of space on the surface on an object. Since the square is the basis by which these measurements are made, the units used to express this surface space are in square feet (ft²).

$$\text{feet X feet} = \text{square feet (ft}^2\text{)}$$

Volume: a three-dimensional measurement (length X width X height). Volume is used to measure the holding capacity of an object. The basis of measuring this capacity is the cube expressed as cubic feet (ft³).

$$\text{feet X feet X feet} = \text{cubic feet (ft}^3\text{)}$$

<u>OBJECT</u>	<u>AREA(ft²)</u>	<u>VOLUME (ft³)</u>
Rectangle	Length' x Width'	Length' x Width' x Height'
Circle	.785 x D' x D'	
Triangle	½ (Base' x Altitude')	
Cylinder		.785 x D' x D' x Length'
Sphere		.5236 x D' x D' x D'
Diameter (D) = 2 x radius Circumference = 3.14 x D Perimeter = Sum of the Sides		

Volume Examples

- The cross-sectional area of a 12 inch pipe is _____?
 $.785 \times 1 \text{ ft} \times 1 \text{ ft} = .785 \text{ ft}^2$
- The volume of a standpipe that is 20 feet in diameter and 60 feet in height is _____?
 $.785 \times 20 \text{ ft} \times 20 \text{ ft} \times 60 \text{ ft} = 18,840 \text{ ft}^3$

Conversions

In our industry, we don't usually use measurements in cubic feet. In order to convert cubic feet to gallons, we multiply the number of cubic feet by 7.48 gallons. To use the previous problem as an example we would multiply 18,840 ft³ by 7.48 gallons.

$$18,840 \text{ ft}^3 \times 7.48 \text{ gallons} = 140,923.2 \text{ gallons}$$

1 psi	= 2.31 ft. of head
1 ft. of head	= .433 psi
1 cuft of water	= 7.48 gallons
1 cuft of water	= 62.4 lbs.
1 gallon	= 8.34 lbs.
1 gallon	= 3,785 ml
1 Liter	= 1,000 ml
1 Liter	= 1,000 grams
1 mg/L	= 8.34 lbs/MG
1 ppm	= 1 mg/L
1 ml	= 1 gram
1 pound	= 453.6 grams
1 pound	= 7,000 grains
1 kilogram	= 1,000 grams
1 cuft/sec	= 448.8 gpm
1 MGD	= 1.55 cuft/sec
1 MGD	= 694.5 gpm
1 HP	= 33,000 ft.lbs./min
1 HP	= .746 kilowatt
1 mile	= 5,280 feet

We use conversions every day. Conversions are a change from one measuring or calculating system to another. Sometimes, we must convert certain measurements in order to determine dosages, volumes, velocities, etc. The key to using the conversion portion of the formula sheet (above) is to find the two entities you need to solve the problem and then decide whether to multiply or divide.

Conversion Examples

If you are given cuft/sec (ft³/sec) and you need gallons per minute (gpm), there is only one line under the conversions header that gives you both forms of measurement you need. What you know is on the left side and what you don't know is on the right side, so you would multiply to obtain the answer.

1. Convert 15 cuft/sec to gpm → $15 \text{ cuft/sec} \times 448.8 \text{ gpm} = 6732 \text{ gpm}$

If you are given gpm and need to convert them to MGD there is only one line that has both gpm and MGD. What we are given is on the right side and what we need to know is on the left so we would divide.

2. Convert 2500 gpm to MGD → $2500 \text{ gpm} / 694.5 \text{ gpm} = 3.599 \text{ or } 3.6 \text{ MGD}$

Math Review

If we need to find the area (a two-dimensional measurement) in a rectangle, we simply multiply length x width.

<u>OBJECT</u>	<u>AREA(ft²)</u>	<u>VOLUME (ft³)</u>
Rectangle	Length' × Width'	Length' × Width' × Height'
Circle	.785 × D' × D'	
Triangle	½ (Base' × Altitude')	
Cylinder		.785 × D' × D' × Length'
Sphere		.5236 × D' × D' × D'
Diameter (D) = 2 × radius Circumference = 3.14 × D Perimeter = Sum of the Sides		

If we need the cross – sectional area of a circle, we multiply .785 x diameter in feet x diameter in feet, which gives us an answer in **square feet (ft²)**.

2-d Area Examples

1. What would the area be of a basin that measures 6 feet wide by 10 feet long?

Area = Length in feet X Width in feet

Area = 6 ft X 10 ft

Area = 60 square feet (ft²)

2. What would the cross- sectional area of a 24-inch circle be?

Area = .785 X diameter in feet X diameter in feet

Area = .785 X 2 ft X 2 ft

Area = 3.14 ft²

To find volume (the size of a three – dimensional space enclosed within or occupied by an object) in a rectangle, we account for the third measurement which would be height or depth.

<u>OBJECT</u>	<u>AREA(ft²)</u>	<u>VOLUME (ft³)</u>
Rectangle	Length' × Width'	Length' × Width' × Height'
Circle	.785 × D' × D'	
Triangle	½ (Base' × Altitude')	
Cylinder		.785 × D' × D' × Length'
Sphere		.5236 × D' × D' × D'
Diameter (D) = 2 × radius	Circumference = 3.14 × D	Perimeter = Sum of the Sides

To find the volume of a pipe or cylinder, we multiply .785 x diameter in feet x diameter in feet x height or depth in feet.

3-d Area Examples

1. A sedimentation basin measures 18 feet wide by 26 feet long by 12 feet in depth. What is the volume of this basin?

Volume = Length (in feet) X Width (in feet) X Depth or Height (in feet)

Volume = 26 ft X 18 ft X 12 ft

Volume = 5,616 cubic feet (ft³)

2. How many cubic feet of water would a 24-inch pipeline that measures 1500 feet in length contain?

Volume = .785 X diameter (in feet) X diameter (in feet) X length (in feet)

Volume = .785 X 2 ft X 2 ft X 1500 ft

Volume = 4,710 cubic feet (ft³)

Demand

A means that many states use to figure out the demand for a given area is as follows:

- § Area to be served: 15,000 acres
- § Number of customers per acre: 3.5 taps
- § Peak hourly rate usage: 1.71 gpm

$$15,000 \text{ acres} \times 3.5 \text{ taps per acre} = 52,500 \text{ taps}$$

$$52,500 \text{ taps} \times 1.71 \text{ gpm} \times 1440 \text{ minutes} = 129.3 \text{ MGD peak flow rate or for 24 hr basis}$$

$$\frac{1}{2} \text{ the peak flow or } 64.7 \text{ MGD}$$

$$30 \% \text{ needed to be stored } (.30)$$

$$64.7 \text{ MGD} \times 30 \% = 19.4 \text{ MGD}$$

To figure how much fire fighting water should be stored is accomplished by this formula:

- § Area to be served: 15,000 acres
- § Estimated population per acre: 12 persons
- § Fire storage reserve per capita: 60 gallons

$$15,000 \text{ acres} \times 12 \text{ persons} = 180,000 \text{ pop.}$$

$$180,000 \text{ pop.} \times 60 \text{ gal} = 10,800,000 \text{ gallons}$$

$$\text{Domestic: } 19.4 \text{ MG}$$

$$\text{Fire: } 10.8 \text{ MG}$$

$$\text{Total minimum storage required} = 30.2 \text{ MG}$$

Chapter 3 Review

1. Chemical reactions occur more quickly in _____water.
2. Tank _____is an important consideration in reducing water age.
3. A lack of water turnover causes a rapid depletion of _____ residuals.
4. A storage facility should turn the water completely over every _____ hours.
5. Every time a tank is taken out of service for repair or cleaning, it must be disinfected at a dosage of _____ ppm, held for _____ hours and maintain a residual of _____ ppm.
6. Monkeys Eyebrow's 4800 residents use on average 190,000 gallons of water a day. What would the average daily water demand per resident be for the month of June?
 - a. 40 gallons
 - b. 175 gallons
 - c. 1187.5 gallons
 - d. 467.6 gallons

Answers for Chapter 3 Review

1. Warmer
2. Turnover
3. Chlorine
4. 72
5. 50, 24, 25
6. a

Chapter 4: DISINFECTION

Chapter 4 Objectives

1. Explain the purpose of disinfection.
2. Explain the relation between dosage demand and residual.
3. Identify the chemicals and/or processes used to disinfect the water.
4. Identify the factors that affect the disinfection process.
5. Demonstrate competence with several disinfection calculations.
6. Describe the process for pipeline disinfection.
7. Explain the regulations associated with drinking water disinfection.
8. Explain how to properly disinfect a leak site.
9. Explain how to keep a repair or replacement of pipe as contamination free as possible.
10. Explain what regulatory parameters are involved when repairing a line.
11. Explain how to assure that the disinfectant is reaching all areas of the entire pipe.

Purpose of Disinfection

- To kill pathogens
- To provide a residual safeguard

Disinfection is the selective destruction or inactivation of pathogenic organisms. Water, as well as being the universal solvent, also provides a favorable environment for the growth of many biological organisms, some of which are capable of causing disease. The primary purpose of disinfection is to kill or inactivate pathogens in the water.

Pathogens are disease-causing organisms. Many of the largest disease outbreaks in history can be attributed to waterborne pathogens. Because of their propensity to kill large numbers of the population, these and all other pathogens should be controlled in the treatment process. In fact, the processes of coagulation, flocculation, sedimentation and filtration were developed to control pathogenic contamination in drinking water. Pathogens, which are found in either the forms of bacteria, viruses, or protozoa, are not “removed” by disinfection. Some are killed, some are temporarily inactivated, and some are not affected at all by chemical disinfectants or biocides. Because of this fact, it is mandatory to optimize our disinfection processes.

The secondary purpose of disinfection is to provide a residual safeguard. After regulatory standards are met at the treatment process, it will degrade as it moves through the distribution system. The degree of water quality degradation is affected by its chemical and biological composition, as well as the physical condition of the distribution system it travels through. Proper management, which includes flushing, cleaning, pressure maintenance, backflow prevention, monitoring and line replacement programs is necessary to prevent substantial water quality degradation in the distribution system.

We maintain disinfectant residuals to provide a safeguard against pathogenic contamination that could develop in the distribution system. The regulated **minimum disinfectant residual requirements** are **0.2 mg/l for chlorine** and **0.5 mg/l for chloramines**. The maximum residual disinfectant level (**MRDL**) allowable in a distribution system is **4.0 mg/l for chlorine and chloramine** and **0.8 mg/l for chlorine dioxide**.

Water quality degradation can be indirectly determined by calculating the disinfectant demand:

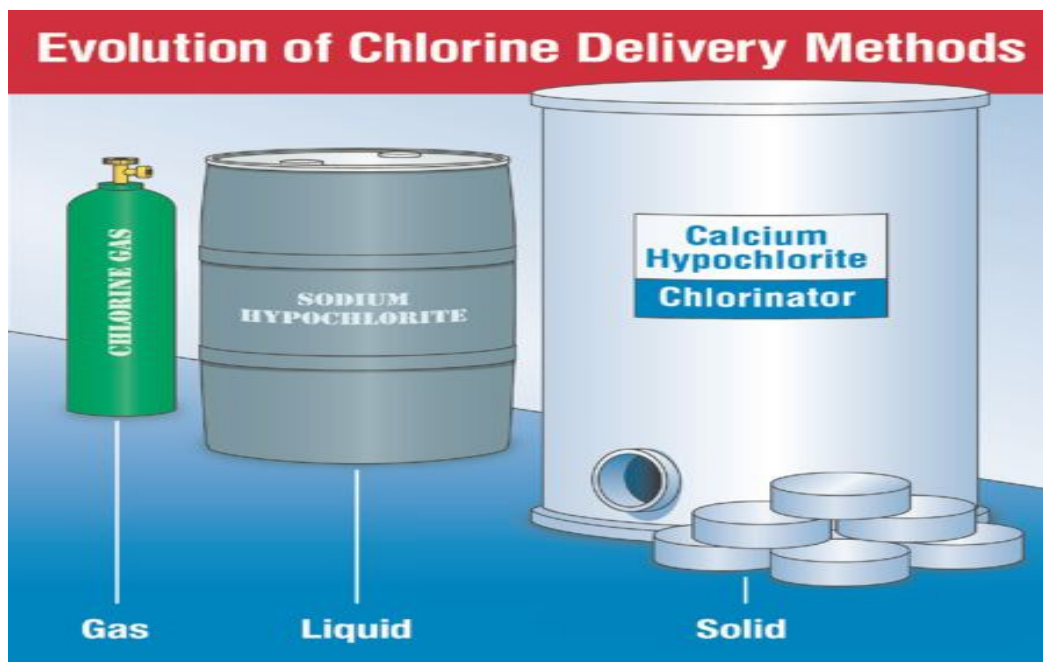
$$\text{DEMAND} = \text{DOSAGE} - \text{RESIDUAL}$$

When demand increases in the distribution system, the cause of this increase should be determined and rectified immediately. Increasing the disinfectant feed will only cover up any problems in the system, and it will make the development of halogenated disinfection by products (carcinogens) much more problematic. It can also lead to compromised corrosion control (lead, copper and other metal leaching; red or black water; leaks ;pressure issues), as well as increased operating costs.

Types of Disinfectants

Chlorine

Chlorine is the most widely used disinfectant. Chlorine is effective at pathogenic inactivation as well as providing a residual safeguard in the distribution system. Chlorine is also a viable oxidizer that allows for the precipitation of substances, including iron and manganese prior to settling or filtration. Chlorine volatilizes gases, such as hydrogen sulfide, methane and benzene, and will also reduce organic related taste and odor and color causing compounds. When combined with organics for a significant length of time, the formation of trihalomethanes and haloacetic acids can be an issue. THM's and HAA5's are suspected to be carcinogenic, and their levels are federally regulated.



Chlorine Gas

Chlorine is 100 % available gas that is fed out of pressurized tanks or cylinders that actually contain both liquid and gas. The liquid-to-gas concentration is temperature-dependent with higher temperatures yielding a higher concentration of liquid. The atmospheric expansion ratio from liquid to gas is 460:1, and a fusible plug is placed in the bottom of the cylinders and designed to melt between 154° - 165° Fahrenheit in order to exhaust the chlorine as a gas. In the event of a leak on the bottom of a cylinder, it should be rolled over to prevent an escape of the liquid. Chlorine is greenish-yellow in color, and it is 2.5 times heavier than air. It is not flammable, but will readily support combustion. Chlorine gas is corrosive when combined with moisture and usually lowers the pH of water.

Whenever chlorine is fed at a rate of ≥ 40 lbs. a day in 150 lb. cylinders, two or more cylinders must be connected together to prevent freezing of the feed system. Though small leaks can be detected with a 10% ammonium hydroxide solution and will show up with the appearance of a white cloud, leak repair kits, as well as SCBA equipment, should be kept just outside of the chlorine room to allow for quick access. The chlorine feed system is contained in a separate room that opens to the outdoors. It must be designed with an exhaust fan and vent located near the floor that activates prior to entry and is capable of exchanging the atmosphere in the chlorine room within one minute. A window for visual inspection and a door equipped with panic hardware are also necessities pertinent to chlorine rooms. Evacuation and emergency plans, as well as up-to-date operator training, are essential for the safe operation of these rooms. Inhalation of chlorine gas can be fatal and repeated brief contact can permanently damage the lungs.

Calcium Hypochlorite

65 – 70 % available chlorine is manufactured as a powder as well as in granular form and a tablet. Calcium hypochlorite has a tendency to elevate the pH of water and is a weaker disinfectant than chlorine gas. It has a maximum shelf life of two years and degrades rapidly at that point, especially at higher temperatures. When in contact with oxidizers or organic compounds, a high flammability factor must be accounted for. Calcium hypochlorite should be stored in a clean, cool, dry areaway from other chemicals. It is used frequently used in distribution systems to disinfect new or repaired pipelines and storage facilities, as well in treatment facilities because of safety concerns about chlorine gas.

Sodium Hypochlorite

More commonly referred to as bleach, this is 5.25 – 12 % available chlorine. Just as with calcium hypochlorite, sodium hypochlorite has a tendency to raise the pH of water and is a weaker disinfectant than both chlorine gas and calcium hypochlorite. Just as with all chlorine products, an acid is produced when in contact with water and is corrosive. The shelf life of sodium hypochlorite is 60 – 90 days and higher concentrations degrade more rapidly, so greater volumes of weaker solutions should be fed. This “bleach” is used for distribution system disinfection, as well as a primary disinfectant in some treatment facilities. Sodium hypochlorite should be stored in a cool, dark location because heat, metals, and time will cause it to degrade.

Chloramines

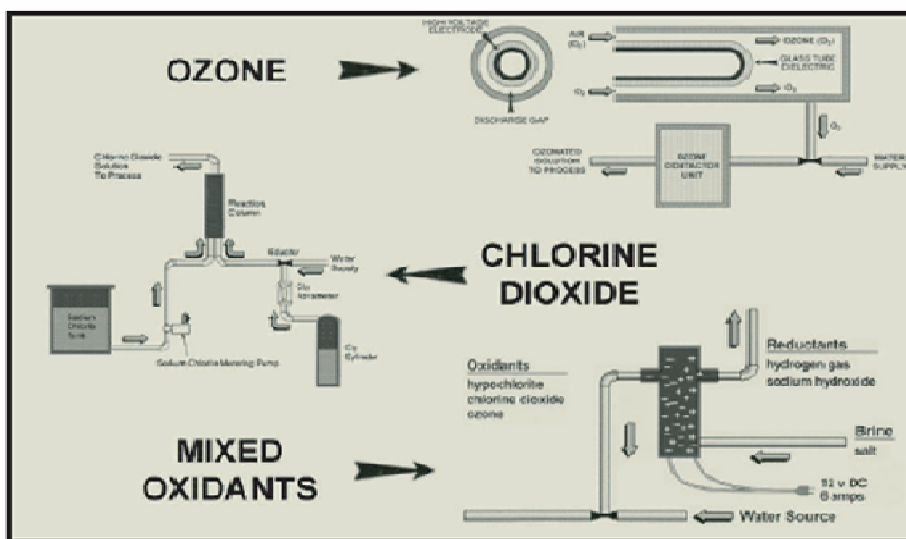
Chloramines are formed when water containing ammonia is chlorinated or when ammonia is added to water containing chlorine. It is an effective bactericide that produces fewer halogenated disinfection by – products, and it is generated on site. Chloramine is a weak disinfectant, and it is much less effective against viruses or protozoa than free chlorine. Chloramine lasts longer in a distribution system than chlorine, by some accounts up to 27 days. Used appropriately as a secondary disinfectant, it prevents bacterial regrowth in a distribution system.

Chlorine Dioxide

This is a pungent, reddish – yellow gas that is an effective disinfectant against waterborne microbes, including *Cryptosporidium*. It is an effective oxidizer with iron, sulfur compounds and certain taste and odor issues. It dissipates quickly as it reacts as it converts back to chlorine. Chlorine dioxide must be generated on – site and can be formed by combining gaseous chlorine with a 25% dosage of sodium chlorite. A stronger disinfectant than chlorine, it produces fewer halogenated by-products than does chlorine, but a major by-product is chlorite.

Ozone (Ozonation)

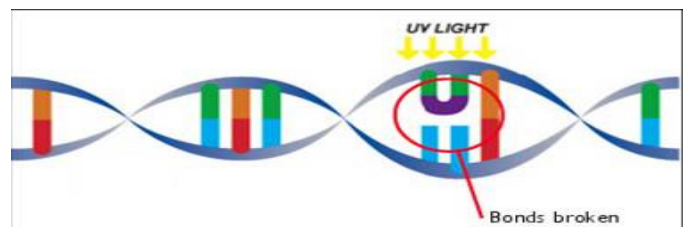
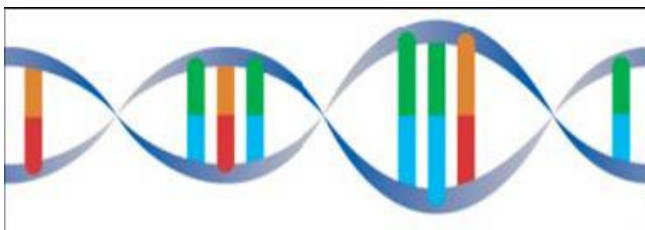
Ozone is formed when a third atom is added to an oxygen molecule. It is produced by passing oxygen through low amperage current. Ozone requires a much shorter contact time than chlorine and does not produce halogenated by-products unless the bromide ion is present. Ozone is a viable disinfectant against Giardia and Cryptosporidium. Ozone gas is unstable and does not maintain an adequate residual in water. Ozone is the strongest disinfectant available for the treatment of potable water.



6.4 Figure 1: Alternative disinfectants

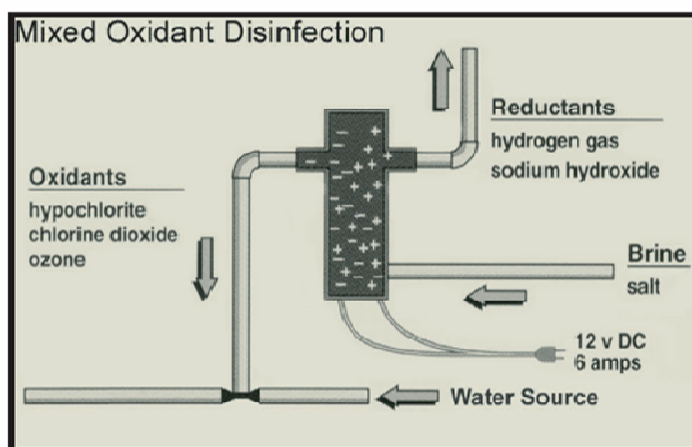
Ultraviolet Disinfection (UV)

Ultraviolet radiation is generated by a special lamp that, if used correctly, penetrates the cell wall of an organism and disrupts the cell's genetic material and does not allow the cell to reproduce. UV produces no known toxic by-products and requires short contact times to accomplish its task. It does not provide a residual disinfectant.



MIOX (Mixed Oxidants)

Miox is a dilute chlorine-based liquid disinfectant generated through electrolysis of salt water. It is a weaker, but safer, disinfectant than gaseous chlorine. In some instances, reduced DBP formations have been reported. It is gaining popularity in some facilities because of the cost and the ease of production of MIOX.



6.4.2 Figure 1: Mixed oxidant disinfection

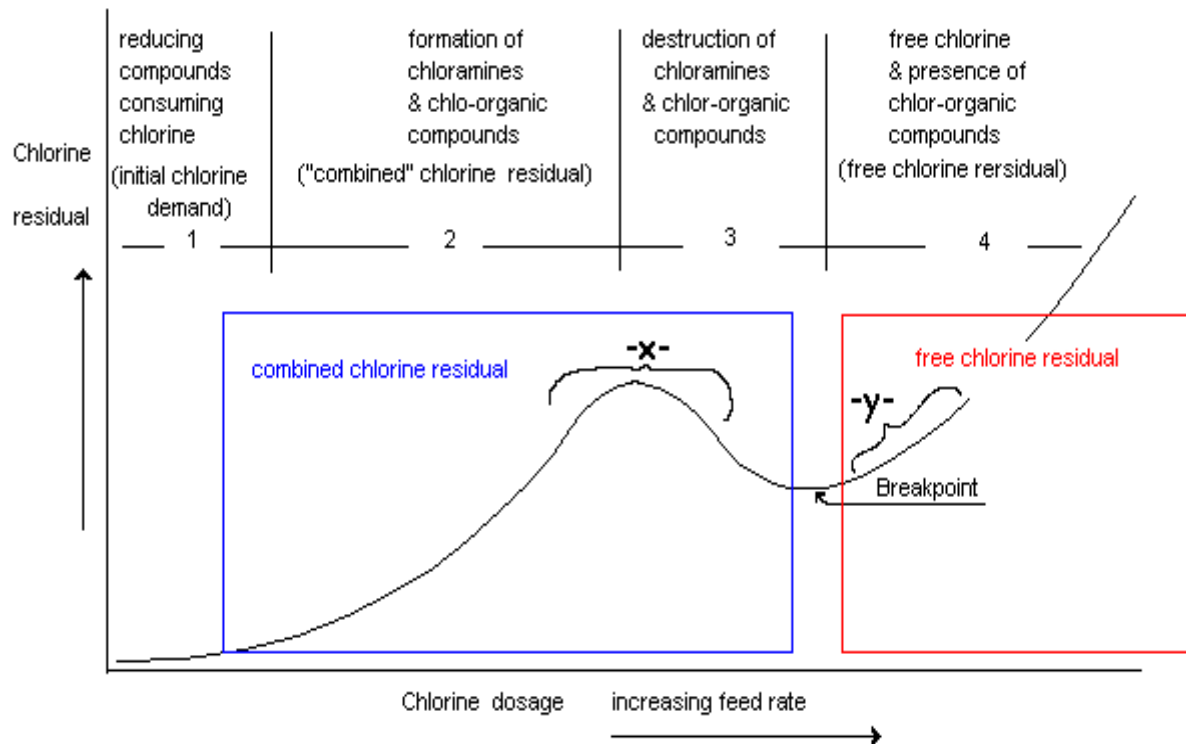
Factors that Affect the Disinfection Process

Dosage (Concentration of Disinfectant)

The more disinfectant added to the water, the more effective the disinfection process is if all other factors are equal. If longer contact times are used, similar results can be achieved with reduced dosages. If shorter contact times are used, greater dosages are required.

Contact Time

The longer the contact time, the more effective the disinfection process will be. The disinfectant must physically come in contact with the pathogens or contaminants that we desire to disrupt. So the longer the disinfectant is allowed to react with the water, the more effective it will be as a disinfectant. Inadequate contact between chlorine and water will not allow breakpoint chlorination to occur, which will greatly reduce the viability of the disinfection process and the control of pathogens.



Interfering Substances

Chlorine disinfectants must physically contact the pathogens they react with so turbidity and other substances can provide a barrier or shield that may protect the pathogen from effectual disinfection. Some substances are actually food sources for pathogens.

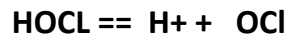
Temperature

The warmer the water, the faster the chemical reactions take place. In the warmer months when the water is warmer, breakpoint chlorination can be more quickly achieved. But, because the dissipation of chlorine is driven by chemical reaction, free residuals will be harder to maintain in the summer months as opposed to the winter months.

When chlorine is added to water, the following chemical reactions take place.

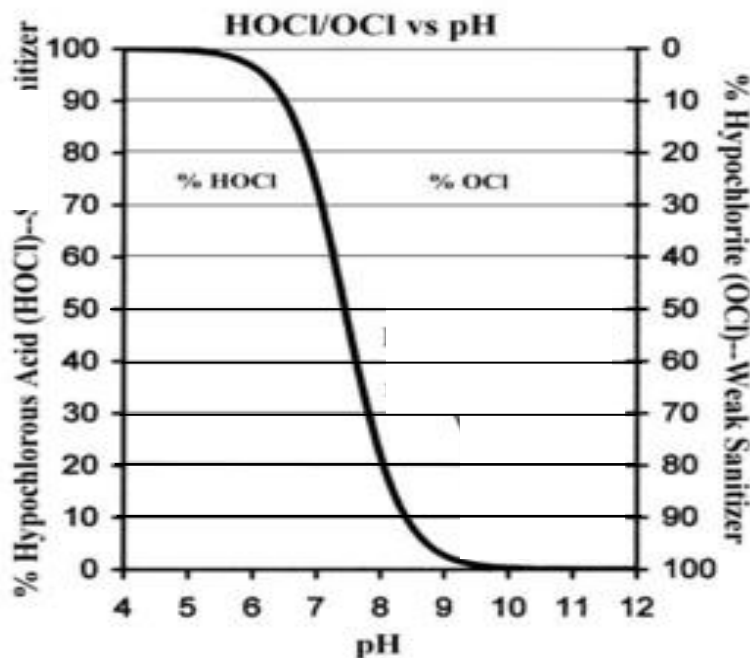


(Chlorine + water \rightleftharpoons Hypochlorous Acid + Hydrochloric Acid)



(Hypochlorous Acid \rightleftharpoons Hydrogen Ion + Hypochlorite Ion)

The Hypochlorous Acid and the Hypochlorite Ion are the compounds that inactivate or kill pathogens. Hypochlorous Acid is 100 X more effective than the Hypochlorite Ion. Both occur simultaneously in water with the concentrations of each compound being pH dependent.



Point of Application

The more thorough the mixing process, the quicker the chemical reaction will take place and the less contact time needed for effective disinfection. If we apply all of the chlorine in one end of a long pipeline before filling it with water, then when we do fill up the pipe with water all of the disinfectant will wash toward the other end of the pipeline. Adding a proportional amount to each pipe joint just before slowly filling the pipeline will provide a more thorough and effective means of disinfection.

Disinfection Monitoring & Control

Disinfection Feed and Dosage Determination

The amount of disinfectant applied to the water is dependent upon the **demand** and the desired amount of **residual** to be carried into the distribution system. The total of **demand + residual** makes up the **dosage** required. Breakpoint chlorination should be practiced at all times to ensure that all demand is met and a free residual is established after adequate contact time.

If using the *slug method* of disinfection, the operator will add an arbitrary amount of disinfectant and then mathematically determine if sufficient amounts of the disinfectant were added to achieve correct dosage requirements. For this type of application, the following formula is used:

$$\text{ppm (or mg/l)} = \frac{\text{lbs. of chemical} \times \% \text{ purity}}{8.34 \times \text{MG}}$$

Although the slug method works, it wastes chemical and therefore, money. A more feasible method is to first determine what dosage is required and then calculate the amount of chemical needed to obtain the desired results. The *continuous flow method*, which feeds the correct dosage of chemical into a flow of water, uses the following formula:

$$\text{lbs of chemical/day} = \frac{\text{ppm (mg/l)} \times 8.34 \times \text{MGD}}{\% \text{ purity}}$$

When disinfecting a volume of water (no flow condition), such as when using the *spray or swab method*, the formula used would be:

$$\text{lbs of chemical} = \frac{\text{ppm (mg/l)} \times 8.34 \times \text{MG}}{\% \text{ purity}}$$

When mixing a chlorine solution using calcium or sodium hypochlorite, the following formula can be used to determine the strength of solution:

$$\text{strength of solution} = \frac{\text{lbs of chlorine}}{\text{lbs of solution}}$$

To express the strength of solution as a percentage of chlorine in the solution, one would multiply the strength of solution by 100. The strength of solution can be multiplied by the pounds of solution fed to determine the pounds of chlorine fed. The pounds of solution fed can be determined by multiplying the gallons of solution fed by 8.34 (lbs).

For example, if 5 gallons of bleach (5.25% available) is mixed with 45 gallons of water to produce 50 gallons of solution, the strength of solution would be calculated as follows:

$$\frac{5 \text{ gal. of bleach} \times 8.34 \text{ lbs/gal} \times .0525}{50 \text{ gal. of solution} \times 8.34 \text{ lbs/gal}} = \frac{2.189 \text{ lbs of chlorine}}{417 \text{ lbs of solution}} = .005$$

.005 is the decimal equivalent of .5% (.005 X 100)

If 40 gallons of the solution was fed: $40 \text{ gal} \times 0.005 = 0.2 \text{ lbs of chlorine fed}$

If the chemical being mixed into the solution is a liquid, then the final weight must be multiplied by the specific gravity of the liquid chemical to get the actual weight of the chemical being fed. This is because not all liquids weigh the same as water (8.34 lbs/gal).

If the 0.2 lbs of chlorine were fed into 5,000 gallons of water, the dosage could be calculated as follows:

$$\text{DOSAGE (ppm or mg/l)} = \frac{0.2 \text{ lbs of chlorine}}{8.34 \times .005 \text{ MG}} = 4.79 \text{ or } 4.8 \text{ ppm (mg/l)}$$

Bacteriological sampling is used to determine potential contamination problems allowing the system to notify the public to boil their water while system personnel can locate and rectify the cause of the problem. Just because bacteriological samples are negative for total coliforms does not mean the water is not contaminated. Heterotrophic (standard) Plate Counts are very useful monitoring tools for the disinfection process.

Pipeline Disinfection

Operators should ensure that proper disinfection methods are followed to provide the maximum public health protection when installing and repairing pipelines that are intended for potable water service.

Inspection

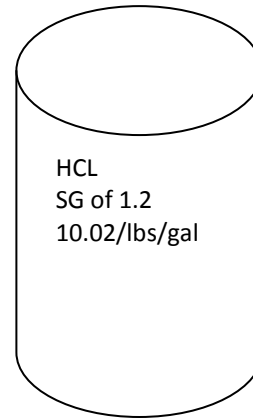
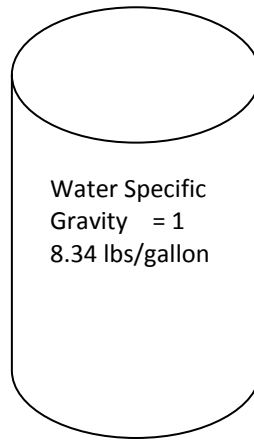
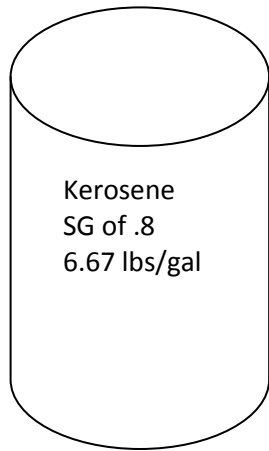
Pipes and fittings should be examined for any damage with particular attention to joints. Any damage to the pipe ends or gaskets may result in leakage. The pipe sections should be clean and free of blemishes on the interior. Before placing a pipe in the trench it should be thoroughly inspected for contamination.

Sanitary Construction Methods

Sanitary procedures should be adopted whenever work is to be done to a pipeline. It may seem unnecessary or time prohibitive to perform these sanitary practices, but experience has shown that the steps actually save time because they reduce the bacteriological test failure rates.

Keeping Pipes Clean and Dry

Keeping the pipe clean during installation is the single most important factor that results in successful disinfection. The interior of pipes must be protected during installation. Openings should be sealed with watertight plugs or coverings when the construction is halted and personnel are not present to ensure that contamination does not enter the pipe.



The new tank measures 42 feet in diameter and 95 feet tall. How many gallons of 11% bleach will be necessary to disinfect this structure if the bleach has a specific gravity of 1.4?

$$\text{Lbs.} = \text{ppm} \times 8.34 \times \text{MG}$$

$$\text{Lbs.} = 50 \text{ ppm} \times 8.34 \times \frac{(.785 \times 42 \text{ ft} \times 42 \text{ ft} \times 95 \text{ ft} \times 7.48 \text{ gallons})}{1,000,000}$$

$$\text{Lbs.} = 50 \text{ ppm} \times 8.34 \times .98 \text{ MG}$$

$$\text{Lbs. of 100 \% chlorine} = 408.66$$

$$\text{But we are using 11\% bleach so } \frac{408.66}{.11}$$

$$\text{Lbs. of 11\% bleach} = 3715.1 \text{ (3715.0909)}$$

Now, each gallon of bleach weighs 1.3 times more than water so the bleach weighs (8.34 X 1.3) 10.8 lbs.

$$\frac{3715.1 \text{ lbs of 11\% bleach}}{10.8 \text{ lbs/gal}}$$

$$\text{Gallons of bleach} = 344 \text{ (343.99) gallons}$$

401KAR 8:150**Section 4. Disinfection of New and Repaired Water Lines.****(1) New construction projects and line extensions.**

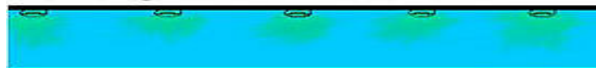
- (a) Disinfection of water lines. A water distribution system, including storage distribution tanks, or all extensions to existing systems, shall be thoroughly disinfected before being placed in service.
- (b) A water distribution system shall disinfect with chlorine or chlorine compounds, in amounts as to produce a concentration of at least fifty (50) ppm and a residual of at least twenty-five (25) ppm at the end of twenty-four (24) hours, and the disinfection shall be followed by a thorough flushing.
- (c) Other methods and testing procedures that provide an equivalent level of protection may be used if the cabinet grants prior written approval.
- (d) A new water distribution line shall not be placed into service until bacteriological samples taken at the points specified in paragraph (f) of this subsection are examined and are shown to be negative following disinfection.
- (e) A water distribution system shall submit to the cabinet results of bacteriological samples for each new construction project, replacement, or extension to existing systems, after the disinfection and flushing.
- (f) A sample shall be taken in the newly constructed line at each of the following points:
 - 1) Within 1200 feet downstream of each connection point between the existing and new lines;
 - 2) One (1) mile intervals; and
 - 3) Each dead end, without omitting any branch.
- (g) A new or routine replacement line shall not be placed in service until negative laboratory results are obtained on the bacteriological analyses.
- (h) Sample bottles shall be clearly identified as “special” construction tests and the results submitted to the cabinet shall be clearly marked as “special” samples.
- (i) With the routine monthly compliance bacteriological samples are to be used to lift a boil water advisory. Samples to be used to lift a boil water advisory shall be submitted to the cabinet as soon as results are known.

If using tablets of calcium hypochlorite, be sure to use food-grade glue when attaching them to pipes. Attach the tablets to the top of the pipe so the disinfectant can disperse throughout the pipe when it is filled.

1. The tablets are attached to the top of the pipe.



2. The tablets begin to dissolve, releasing chlorine into the water.



3. Chlorine fills the pipe and kills microorganisms.



Pipe Disinfection & Repair

Operators need to follow some important steps to ensure the public health when installing, replacing or repairing water lines.

Inspection

Pipes and fittings should be examined for any damage, with particular attention paid to the joints. Any damage to this area could result in leakage. The pipe sections used for potable water should be clean and inspected before placement in a ditch.

Sanitary Construction Methods

Sanitary procedures should be adopted to achieve satisfactory bacteriological quality. Time spent in this regard may seem excessive or unnecessary, but experience has shown that the steps actually save time because they reduce the failure rate of bacteriological tests, which would require additional disinfection or, in some cases, replacement.

Keep Pipe Clean and Dry

The interior of pipelines and fittings need to be protected from contamination during installation. Pipelines in storage should be covered or sealed to prevent contamination from entering the pipe while waiting to be used. Openings should be sealed with a watertight plug or other suitable plugs when construction is halted and personnel are not present.



Disinfection

State mandated disinfection parameters must be followed before a new or repaired line is put into service.

(2) Line repair due to breaks or ruptures.

- (a) The system shall thoroughly flush the break area and maintain at least a minimum disinfectant residual, pursuant to Section 1 (1) of this administrative regulation.
- (b) The system may leave the line in service or return the line to service before receiving bacteriological results, and may forego a boil water advisory of:

- 1) Pressure is maintained;
- 2) The break area is thoroughly flushed; and
- 3) At least the minimum disinfectant residual, pursuant to Section 1(1) of this administrative regulation is maintained.

- (c) The system shall take at least two (2) bacteriological tests, one (1) located before, or just upstream of, the break or rupture, and one (1) located behind, or just downstream of, the break or rupture, as close to the break or rupture as practical. Additional samples may be required, if necessary to be representative of the area affected by the break. Sample bottles shall be clearly identified as "special" tests and the results submitted to the cabinet shall be clearly marked as "special" samples.
- (d) Records of results shall be submitted to the cabinet with routine monthly compliance samples unless the samples are required to lift a boil water advisory, and shall be maintained for one (1) year. Samples needed to remove a boil water advisory shall be submitted to the cabinet as soon as the results are known.
- (e) A water system shall notify the cabinet immediately if the pressure drops below twenty (20) pounds per square inch in the distribution system surrounding the break, or a break or rupture occurs that requires more than eight (8) hours to repair.
- (f) The system shall issue a boil water advisory if the cabinet determines that a boil water advisory to protect the public health.
- (g) Reports pursuant to 401 KAR 8:020, Section 2(7)(c) are not required for a loss of pressure, break, or rupture occurring in service lines serving only one (1) single family residence.
- (h) Community and non-transient non-community public water systems shall maintain a log of all breaks or ruptures, which shall include the:
 - 1) Date and location of the break or rupture;
 - 2) Time it was discovered;
 - 3) Population affected;
 - 4) Length of time required to repair the break or rupture;
 - 5) Date and time disinfectant residuals are detected; and
 - 6) Date and time bacteriological samples are taken
- (i) The log shall be available for inspection by the cabinet.

Chapter 4 Math Problems to Review

1. Getz water system disinfects their water at 2.5 ppm sodium hypochlorite. Through tests, they have determined that the chlorine demand in Getz is 1.0ppm. What is Getz's disinfectant residual?
2. Petticoat Junction has determined that the chlorine demand in their system is 1.1 mg/l. At the far reaches of their distribution system, the chlorine residual is 0.8 mg/l. What is Petticoat Junction's chlorine dosage?
3. Barkerville treats their water with a 5.25 % available solution of sodium hypochlorite at a rate of 2.0 ppm. The treatment facility treats 2.5 MGD and their distribution system contains four elevated storage facilities. Barkerville has 528 miles of distribution piping and has discovered that the residual at the far reaches of their system is 0.4 ppm. What is Barkerville's chlorine demand?
4. How many pounds of 70 % available calcium hypochlorite would be needed to treat 720,000 gallons at a rate of 500 gpm to the desired dosage of 2.1 mg/l?
5. Would the regulations be met if we dosed a new 10" PVC waterline that is 7,500 feet in length if we glued in 10 (ten) 1 lb. (65% available) HTH tablets?
6. How many gallons of 9% available sodium hypochlorite will be needed to treat 90,000 gallons of water and obtain a dosage of 3.1 mg/l? The sodium hypochlorite has a specific gravity of 1.4.

7. If we add 2 gallons of bleach with a specific gravity of 1.5 to 200 gallons of water, what would be the percent strength of this solution?
8. The second shift treatment operator in Mt. Airy used 8.5 lbs. of chlorine during his 8-hour shift. Mt Airy treats 4.5 MGD. What dosage did the second shift operator obtain during his shift?
9. If Hooterville doses its water at the treatment facility with 12.5 pounds of gaseous chlorine and the residual in the distribution system is 0.8 mg/l. What would the demand be if they treat 1 MGD?
10. If the concentration of the disinfectant is decreased then the _____ must be increased.
11. The strongest disinfectant for potable water disinfection is _____.
12. If your disinfectant residual is at least _____% of the total chlorine, then breakpoint has been achieved.

Answers for Chapter 4 Review

1. ppm (dosage) – 1.0 ppm (demand) = 1.5 ppm residual
2. 1.1 ppm (demand) + 0.8 ppm (residual) = 1.9 ppm (mg/L) dosage
3. 2.0 ppm (dosage) – 0.4 ppm (residual) = 1.6 ppm (mg/L) demand

4. $\text{lbs} = \frac{\text{ppm} \times 8.34 \times \text{MG}}{\% \text{ purity}}$
 $\text{lbs} = \frac{2.1 \text{ ppm} \times 8.34 \times .72 \text{ MG}}{.70 \text{ purity}}$

$$\frac{\text{lbs} = 12.61}{.70} \quad \text{lbs} = 18$$

5. $\text{ppm} = \frac{\text{lbs of chemicals}}{8.34 \times \text{MG}}$
 $\text{ppm} = \frac{10 \text{ lbs of chemical}}{8.34 \times (.785 \times .83 \times .83 \times 7500 \text{ ft} \times 7.48 \text{ gal})}$
 $\text{ppm} = \frac{10 \text{ lbs of HTH} \times .65}{8.34 \times .03 \text{ MG}}$
 $\text{ppm} = \frac{6.5 \text{ lbs}}{.25}$
 $\text{ppm} = 26.00$

6. $\text{lbs} = \frac{\text{ppm} \times 8.34 \times \text{MG}}{\% \text{ purity}}$
 $\text{lbs} = \frac{3.1 \text{ ppm} \times 8.34 \times 0.09 \text{ MG}}{0.09}$
 $\text{lbs} = \frac{2.33}{0.09}$
 $\text{lbs} = 25.9$
 $\text{gallons} = 25.9 \text{ lbs} \div 11.7 \text{ lbs/gal} (8.34 \text{ lbs} \times 1.4 \text{ SG})$
 $\text{gallons} = 2.2$

7. $SOS = \frac{\text{lbs of chemical}}{\text{lbs of solution}}$

$SOS = \frac{25 \text{ lbs of chemical}}{1693 \text{ lbs of solution}}$

$SOS = 0.014$ $SOS = 1.4\%$

8. $\text{ppm} = \text{lbs of chemical} \div (8.34 \times \text{MG})$

$\text{ppm} = 8.5 \div (8.34 \times 1.5 \text{ MG})$

$\text{ppm} = .68$

9. $\text{ppm} = \frac{\text{lbs of chemical}}{8.34 \times 1 \text{ MG}}$

$\text{ppm} = \frac{12.5 \text{ lbs}}{8.34}$

$\text{ppm} = 1.5$ (1.4988) dosage

$1.5 \text{ ppm (dosage)} - 0.8 \text{ ppm (residual)} = 0.7 \text{ ppm DEMAND}$

10. Contact time

11. Ozone

12. 85

Chapter 5: BACTERIOLOGICAL SAMPLING

Chapter 5 Objectives

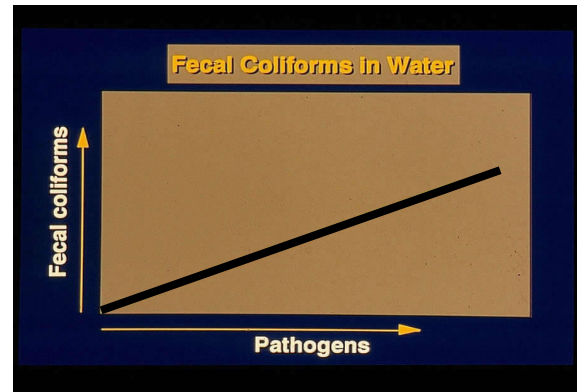
1. Identify the purpose of sampling.
2. Define a pathogen.
3. Describe the relationship between pathogens and coli forms.
4. Describe requirements relative to drinking water sampling.
5. Describe and explain proper drinking water sample collection techniques.
6. Describe the importance of representative sampling.
7. Define the relevance and necessity of check samples.
8. Denote disinfection test methods.
9. Define regulatory parameters for proper record keeping.

Purpose of Bacteriological Monitoring

Bacteriological analysis is used for the detection of pathogens in potable water. Pathogens are disease-causing organisms.

Coliforms, found in the intestinal tract of warm-blooded animals, are used as indicator organisms. They are easy to grow and identify and the presence/absence tests are relatively inexpensive. If coliforms are present, there is a great likelihood that pathogens are also present.

Sodium thiosulfate is the dechlorinating tablet found in Bac-T sample bottles.



Fecal or E. coli contamination dictates a regulatory mandated boil water and public notice.

Sample Requirements

All systems require a minimum of 2 samples. The size of the population served determines the required number of samples.

- If < 40 samples taken, all must be negative for compliance.
- If > 40 samples taken, may have 5 % positive without a violation.

HPC's

Heterotrophic Plate Count (HPC) measures the total number of bacteria in a sample. This may also be called a standard plate count. An operator may use an HPC sample as a residual compliance sample if needed.

Collection

Collect at least 100 milliliters of water in a sterile container. Sample bottles contain sodium thiosulfate that is used to dechlorinate the sample. Compliance samples should NOT be collected from fire hydrants, water fountains, hoses or taps with swivel faucets. Do NOT use leaky faucets or faucets that sputter or do not have a steady flow. Allow the water to run in a pencil-sized stream for 2 – 3 minutes or until temperature is constant. *THE MAJORITY OF SAMPLER ERROR IS THE RESULT OF POOR SAMPLING TECHNIQUE.*

Representative Sampling

Samples should represent the quality of the water throughout the entire system. A negative sample only means that at the time the sample was taken at the location from which it was taken, the water was satisfactory.

Do NOT avoid suspected areas of poor water quality dead ends and low-use areas. Update sampling plans as the distribution system extends or recedes and include residential, commercial, industrial and educational sites.

Check Sampling

Within 24 hours of a positive sample notification, a system must collect 3 check samples: 1 from the original sample location and 1 within 5 service connections upstream and one within 5 service connections downstream.

Check samples are still considered compliance samples. If the system collects fewer than 5 samples a month for compliance, the month after a check sample incident the system MUST submit 5 compliance samples.

Special samples (line extensions, new lines, line breaks, etc) do NOT count toward compliance. BARF's (bacteriological analysis report forms) must still be submitted with special samples even though they are NOT counted toward compliance.

Methods

Multiple Tube Fermentation	ColiBlue 24
Membrane Filtration	Colitag
MMO – MUG Most Probable Number	Colilert 18
MI agar	Colilert
Coliscan	Fluorocult
ONPG-MUG	Colisure

Recordkeeping

Always fill out Chain of Custody documents for samples. BARF's do NOT take the place of a Chain of Custody document. Bacteriological records must be maintained for 5 years.

Sample labels, Chain of Custody documents and BARF's should include the following:

- Collection date and time
- Type of disinfection
- Location code
- Name, address and PWSID
- Chlorine residual
- Signature of sampler

Chapter 5 Review

1. What temperature should the sample achieve?
2. Check samples must number ____ samples and be performed within ____ hours of a positive sample notification.
3. The number of samples taken is dependent upon ____.
4. ____ can be used in place of a compliance sample if minimum residual levels are not satisfied.
5. What needs to be on a chain of custody?
 - a) PWSID
 - b) Samplers name
 - c) Where the sample was taken
 - d) All of the above

Answers for Chapter 5 Review

1. The temperature of the water in the main.
2. 3, 24
3. Population
4. HPC's
5. d

Chapter 6: PIPES & SERVICES

Chapter 6 Objectives

1. Identify important infrastructure issues related to distribution systems.
2. Differentiate between the different types of pipes and the variables to consider when selecting a pipe.
3. Describe the importance and need of detailed management plans and distribution system maps as they relate to operational concerns and pipe installation.
4. Describe the regulatory requirements related to mandated inclusions.
5. Explain proper pipe installation techniques, bedding practices, and thrust blocking techniques.
6. Differentiate between the types of distribution systems (tree, loop, grid, dead end lines) and describe the shortcomings and strengths of each type of system.
7. Describe the process of pipe disinfection and repair related to pressure and leak testing of new and repaired water lines.
8. Describe the benefits of unidirectional flushing compared to conventional flushing.
9. Explain the sanitary practices related to pipe repair.
10. Calculate math computations related to area and volume of water distribution systems (sedimentation basins, storage tanks, pipes, etc.) and related to basic disinfection practices.
11. Explain the regulations and tests associated with improving water quality.
12. Identify the characteristics and pathogens related to biofilms.
13. Differentiate between repair and rehabilitation options of water distribution pipes (e.g., pipe bursting, slip lining, pigging, pipe jacking, etc.)
14. Explain the factors related to the depth and width of trench excavation and identify basic trench safety practices.

Type of Pipe and Their Uses

- Pipe material selection along with the appropriate size should match the intended use of the pipe.

Some variables to consider when selecting the correct pipe are:

- Flow requirements
- Pressure requirements
- Corrosive characteristics
- Ease of locating
- Strength
- Effective life span – durability
- Ease of repairs

MATERIAL PROPERTY	DI	PVC	HDPE
Tensile Strength	60,000 psi	7,000 psi	3,200 psi
Compressive strength	48,000 psi	9,000 psi	1,600 psi
Yield strength	42,000 psi	14,500 psi	5,000 psi
Ring bending stress	48,000 psi	None specified	None specified
Impact strength	17.5 ft-lbs/in	0.75 ft-lbs/in	3.5 ft-lbs/in
Density	441 lbs/ft ³	88.6 lbs/ft ³	59.6 lbs/ft ³
Modulus of elasticity	24,000,000 psi	400,000 psi	111,000 psi
Temperature range	< 150° F	< 140° F	-50 to 140° F under pressure
Thermal expansion	0.07" per 10° F per 100'	.33" per 10°F per 100'	1" per 10°F per 100'
Corrosion resistance (int)	Good-w/cement lining	Excellent	Excellent
Corrosion resistance (ext)	Good- w/polywrap	Excellent	Excellent
UV resistance	Excellent	Gradual strength decline	Yes – w/carbon black
Abrasion resistance	Excellent	Good	Good
Cyclic resistance	Excellent	Fair	Good
Permeation resistance	Yes	No-solvents & petroleum	No – solvents & petroleum
Scale & growth resistance	Good	Excellent	Excellent

Planning

Detailed plans of the project should be prepared by the project engineer in conjunction with input of the distribution operators. Planning for future growth and development will save money in the future. These plans should include grade, depth, alignment and soil type specifications. Plans must be submitted to the appropriate agencies for approval before construction begins. Right of way and access issues should be resolved before beginning construction. Familiarizing your work plans with other utilities in the area (phone, gas, cable, etc.) will aid in reduced problems once construction begins. Reduce your liability by always having utility lines marked before construction begins. In Kentucky the number to call for line marking is 811.

Data Requirements for a Detailed Management Plan

PHYSICAL	Exist	New	PERFORMANCE	Exist	New	COMMERCIAL/ SERVICE	Exist	New
Installation/yr	Y	Y	Complaint frequency	A	Y	Critical customer	Y	Y
Diameter	Y	Y	Type of complaint	A	Y	Affect on Community	Y	Y
Material	Y	Y	Break frequency	A	Y	No. of People Served	A	A
Length	Y	Y	Type of break	A	Y	Length of shutdown	A	A
Location	Y	Y	Reason for break	A	Y	Coordination w/others	A	A
Interior lining	A	Y	Service (hydraulic) adequacy	Y	Y			
Exterior protection	A	Y	Fire Flow Adequacy	Y	Y			
Joint	A	Y						
Wall thickness	A	Y						
Soil conditions	A	A						
Internal condition	A							
External condition	A							

Y = yes, in all cases A = as needed, or as available

Pipe Installation

The proper installation of pipe will minimize maintenance problems and aid in desirable water quality and therefore, public health. Improper installation can result in water loss, water quality degradation, and loss of service, loss of consumer confidence, water leaks and increased monetary demands.

OPERATIONAL/PHYSICAL	Applies to	Options	CHEMICAL	Applies to	Options
Manufacturing defects	M,P,C	No	Internal corrosion	M,C	Yes
Improper design/installation	M,P,C	No	External corrosion-soil	M,C	Yes
Geologic instability	M,P,C	No	External corrosion-other	M,C	Yes
Higher operating pressures	M,P,C	Yes	Leadite corrosion	M	Yes
Hydraulic transients	M,P,C	Yes	Leadite expansion	M	Yes
Changes in water temp	M	Yes	Material incompatibilities	M	Yes
Excessive external loads	M,P,C	No	Gasket deterioration	M,P,C	Yes
Damage from digging	M,P,C	No	Material fatigue	P	No

M = Metallic (ductile iron and/or cast iron) P = (PVC or HDPE) C = Concrete (RCP or PCCP)

Comparison of Distribution-Size Pipe Materials – Operational Concerns

OPERATIONAL CONCERN	DI	PVC	HDPE
Ease of installation	subjective	subjective	subjective
Can be direct tapped	yes	yes	no
Need for special installation equipment	no	no	yes
Need for special bedding for typical installations	no	yes	no
Need for joint restraint	yes	yes	no
Ability to locate underground	excellent	poor – needs tracer wire	poor – needs tracer wire
Applicable for above-ground installations	yes	yes, with UV protection	yes
Applicable for aqueous installations	yes	yes	Yes-potential for flattening is high
Anticipated service life	100 years	50-100 years	50 years

Distribution system maps

All systems are required by regulation to have *up to date* maps of their distribution systems. Certain items are required to be listed on your map.

Public and semipublic water supplies – general provisions:

(12) Maps

(a) Within twelve months of November 15, 1990, public and semipublic water systems shall have on the premises, or conveniently located to the premises, and an up-to-date map of the distribution system. The map shall, at a minimum, show:

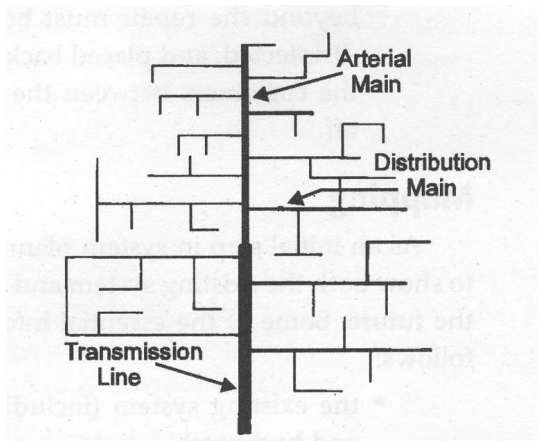
1. Line size
2. Cutoff valves
3. Fire hydrants
4. Flush hydrants
5. Tanks
6. Booster pumps
7. Chlorination stations
8. Connection to emergency or alternative sources
9. Wholesale customer master meters; and
10. Type of piping material in the distribution system and its location

(b) If a public water system, due to age, improper documentation, lost documentation or other valid reason is not able to comply with this requirement, the system may petition the cabinet to modify this requirement to the extent that compliance is not feasible. The petition for modification shall state specifically what portion of this requirement is not practical and why.

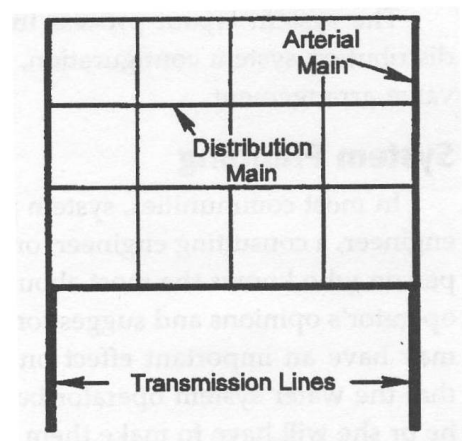
(401 KAR 8:020)

When designing a distribution system, the elimination of dead-end lines is very important. Dead-ends promote the degradation of water quality and reduced or nonexistent chlorine residual.

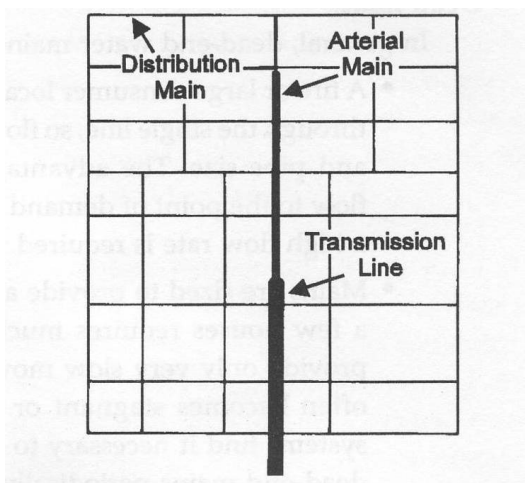
TREE (BRANCHING) SYSTEM with DEAD ENDS



LOOP SYSTEM



GRID SYSTEM



Obviously, the grid system is superior to the tree system as water is always moving and replenishing itself. The elimination of dead end lines or the frequent or routine flushing of them should be a priority. There are many types of flushing devices available for this purpose, from homemade to solar powered.

Excavation

Many factors dictate the depth and width of a trench excavation. Economics, pipe size, available equipment, location of excavation, climatic conditions, pipe load, soil characteristics and groundwater conditions have an impact on the size of your trench.

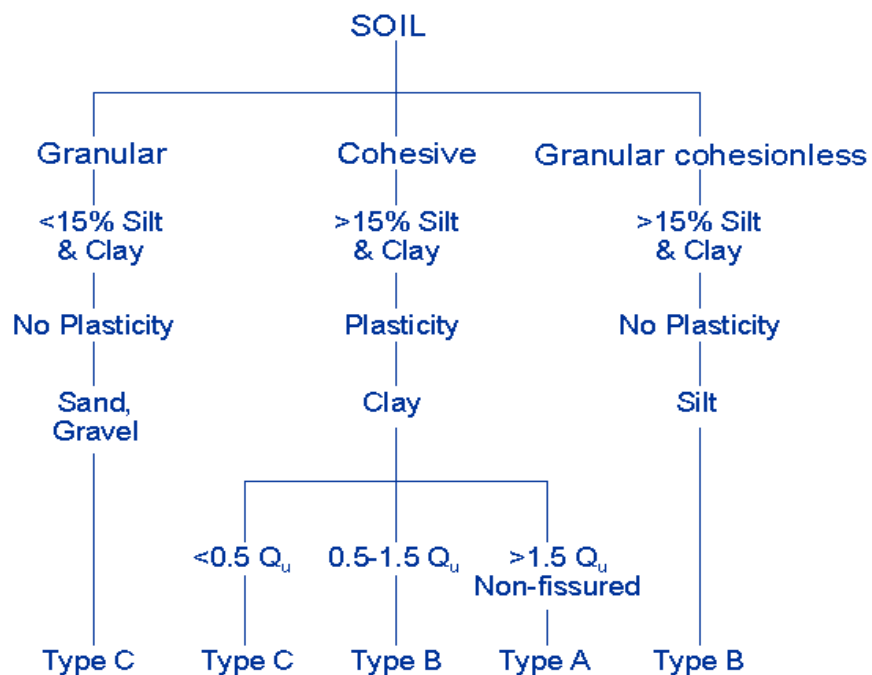
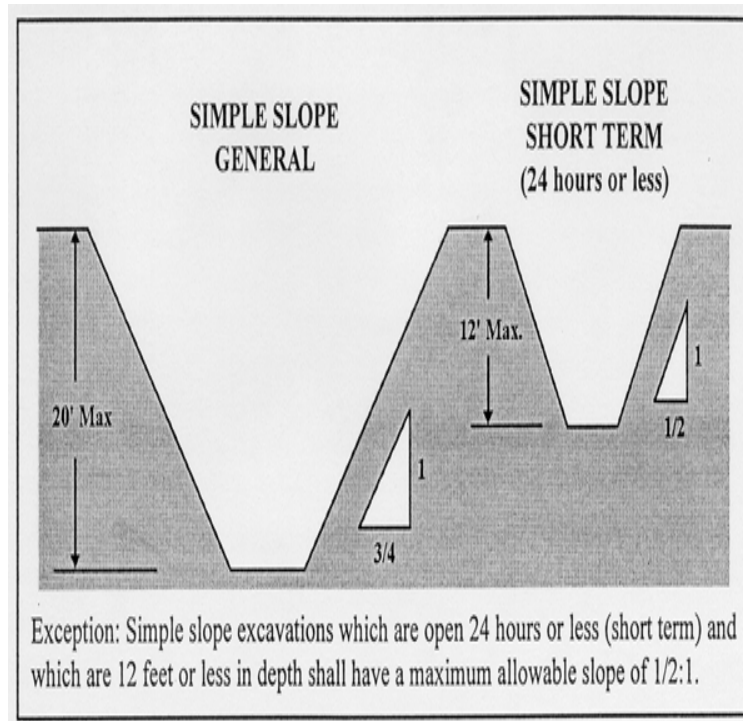
Open cut excavations include clean up that can turn into an expensive endeavor. The trench should be as shallow and narrow as is practical. The trench depth must satisfy the need to prevent freezing in cold weather (3 - 4 feet). If sufficient depth is not obtainable, the pipe should be insulated. The proper bedding material should be used to surround the pipe, particularly in rock trenches to prevent the pipe movement on the rock and causing leaks. Generally, the trench width should be no more than 1 to 2 feet greater than the pipe diameter.

Trench Safety

OSHA regulations should be followed to prevent injury or death whenever an excavation is undertaken. Spoil piles must be kept a minimum of two feet from the edge of a trench so as not to increase the weight on a trench wall and increase the likelihood of a trench collapse or cave-in. Ladders for entry and exit must be placed no more than 25 feet apart so workers can safely enter and leave the trench.

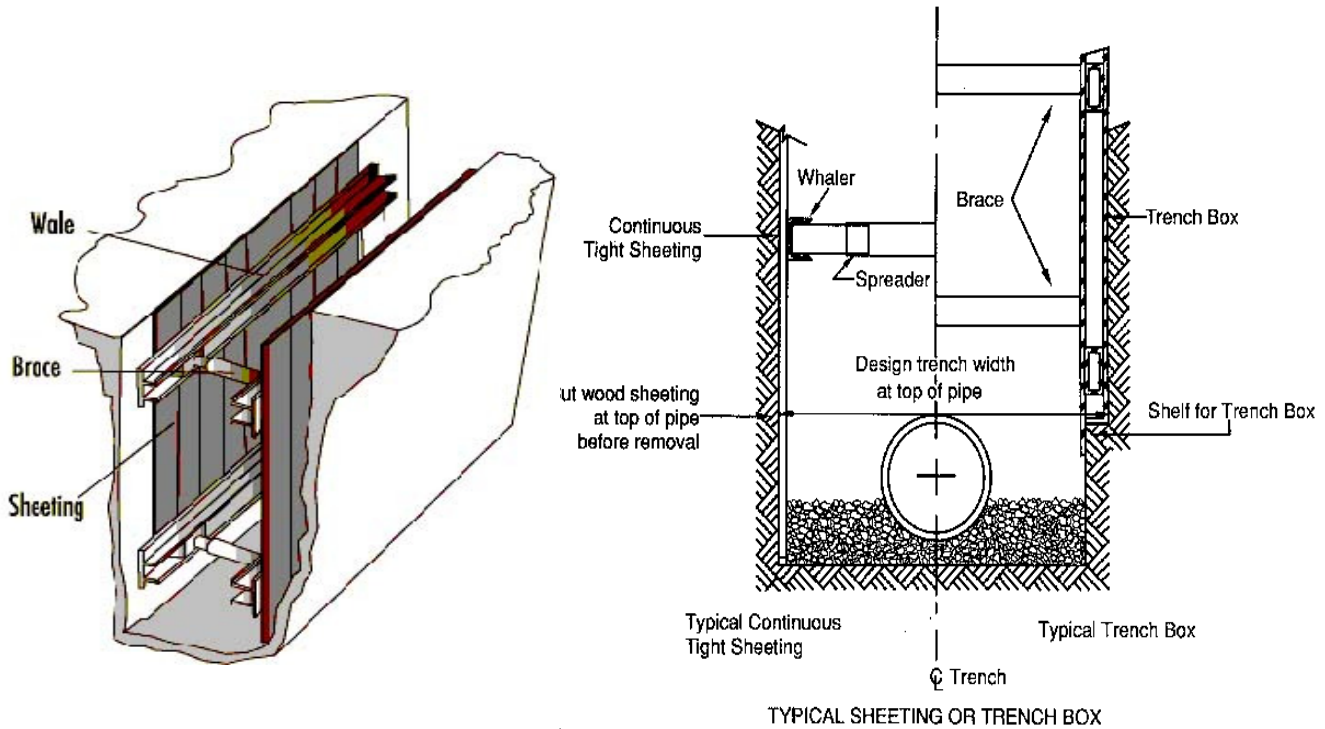
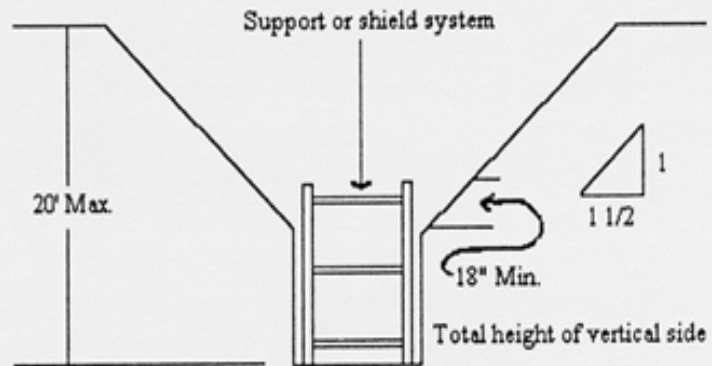


Wherever possible, trenches should be stepped or sloped (grade of slope dependent upon soil type) to cut down on the likelihood of a cave-in.



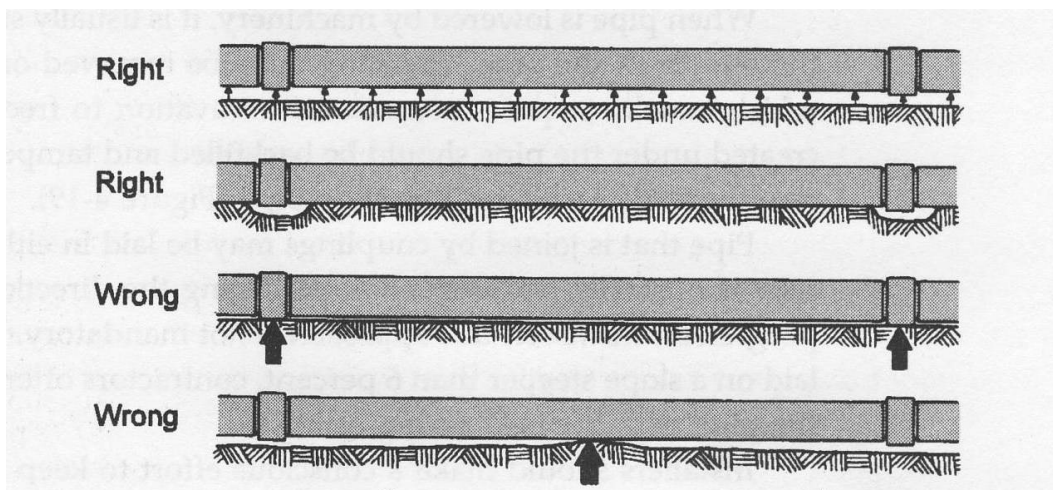
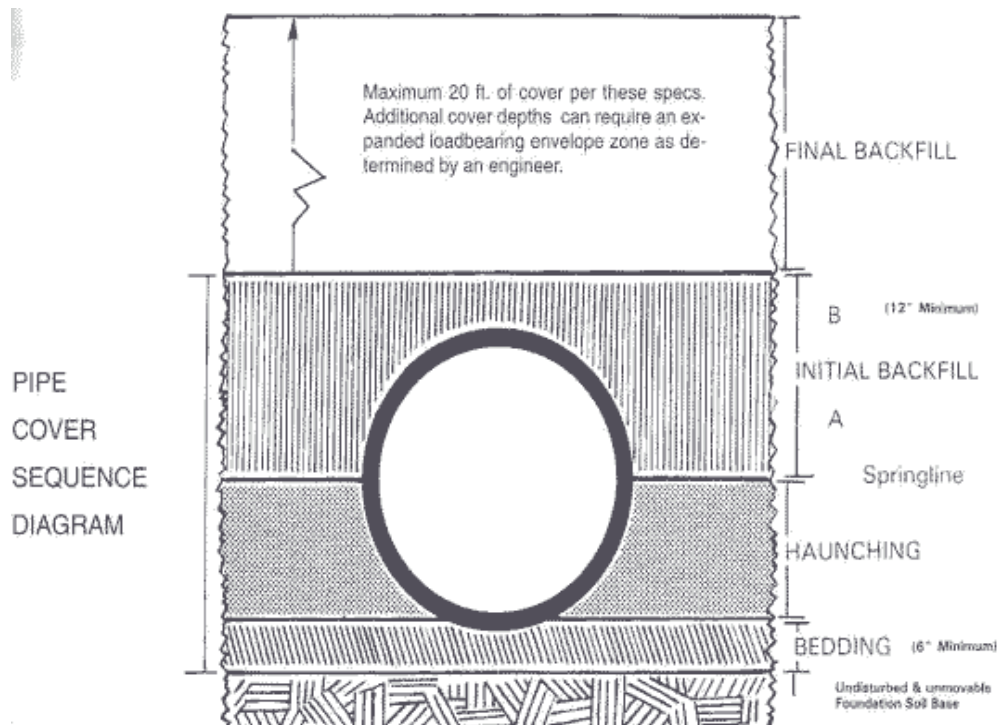
If sloping is not possible, shoring or trench boxes should be used to protect the employees installing or repairing the pipe.

VERTICAL SIDED LOWER PORTION



Bedding

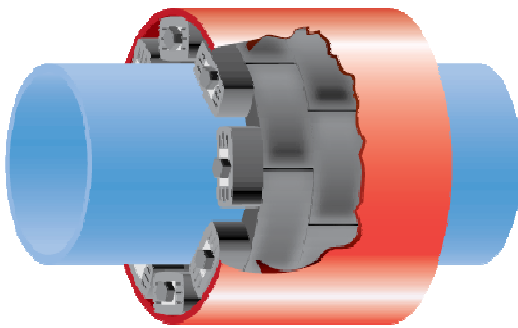
This is usually considered to be the material in which the pipe is partially or fully embedded. It increases the load bearing capability of the pipe. The trench bottom should be correctly leveled and compacted to ensure continuous, firm support throughout its length. The bedding material should be well graded material, uniform in size and not containing rocks or other materials that cannot be graded uniformly.



Pipe Laying

Water mains should not be placed in the same trench with sewer lines. If the sewer pipe is leaking, as they often do, a water main could be surrounded by contaminated soil. If a break were to ensue in that water main, a serious health hazard could be in the making. Waterlines should be laid at least 2 feet above and 2 feet laterally from the sewer line. If the water and sewer lines are at the same elevation, they should be at least 10 feet apart. Whenever a water line crosses a sewer line, the water line must be at least 3 feet above the sewer line and should not contain any pipe joints, except pressure joints on either side of the pipe crossing.

When pipelines are installed under roads, railroads, etc., the pipe should be installed in a way that allows for the accessible repair or replacement of that pipe. Casing or casement pipe is required to protect the water line from loads and stress.



Thrust Blocking

Water under pressure and in motion can exert tremendous forces inside a pipeline. The forces can appear as water hammer, and strains on every fitting and appurtenance associated with the distribution system. A bend, tee, plug, hydrant, valve or any other fitting that changes the direction or velocity of water should be blocked or restrained in some manner. Thrust blocks are masses of concrete that are cast in place between the fitting that needs to be restrained and the undisturbed soil in the trench.

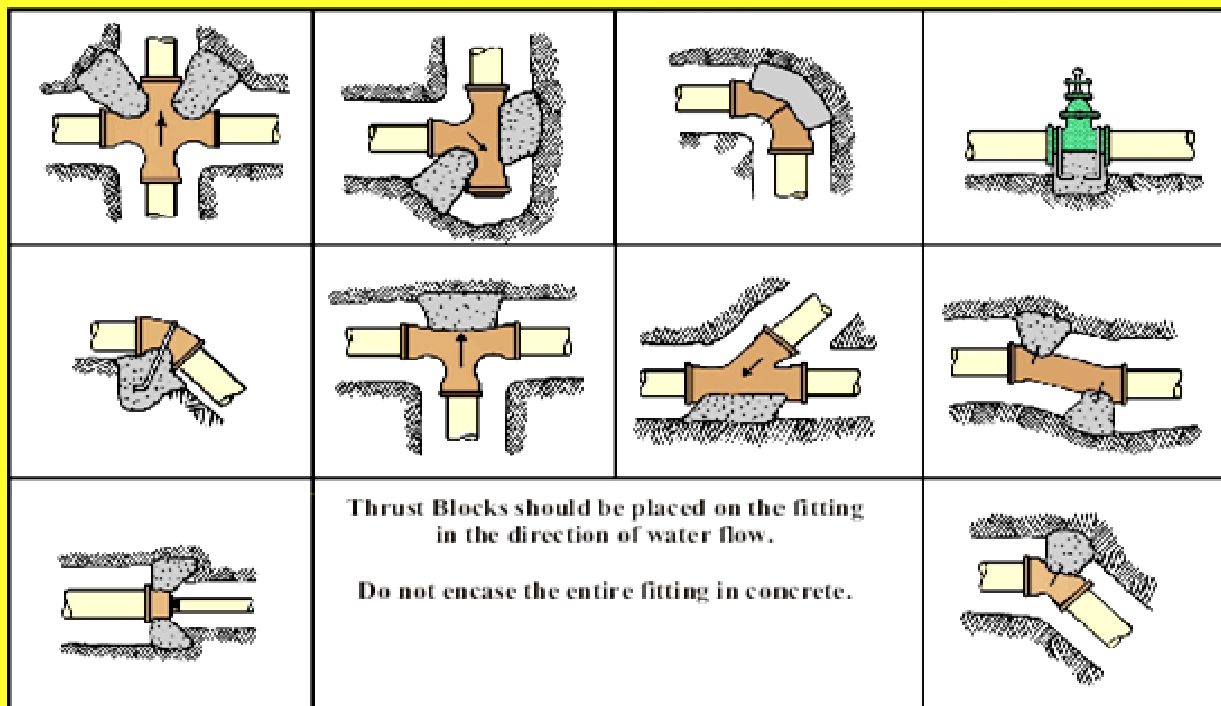
For example: If the water in a completely full 12 inch pipe that is 500 feet in length is required to stop or change direction, there is over 24,485 lbs of water moving in the pipe.

$$(.785 \times 1' \times 1' \times 500' \times 7.48 \text{ gallons} \times 8.34 \text{ lbs.})$$

*****Do NOT cover the entire pipe or fitting with concrete as this will lead to extra hours of labor if the fitting requires maintenance.*****

Thrust Blocking

TYPICAL THRUST BLOCK DETAILS FOR PVC BELL & RING PIPE



Thrust blocking should be used on all gasket joint pipe and fittings since there are great thrusts developed whenever there are changes in direction in the pipeline, dead ends, pipe sizes and valves. These thrusts can blow the joints apart creating big messes.

The thrust block should be placed against undisturbed, fully compacted earth. It should have sufficient area on the soil side to restrain the thrust. The fitting should be placed over a large enough area to keep stresses on the fitting at a minimum.

The thrust block should be designed knowing the intensity of the thrusts developed in the pipe and the load bearing strength of the soil.

Using the tables on the following page you can calculate the size of the thrust block for various pipe sizes.

Example: 4" pipe with a maximum pressure of 100psi at a 90 degree elbow in clay soil.

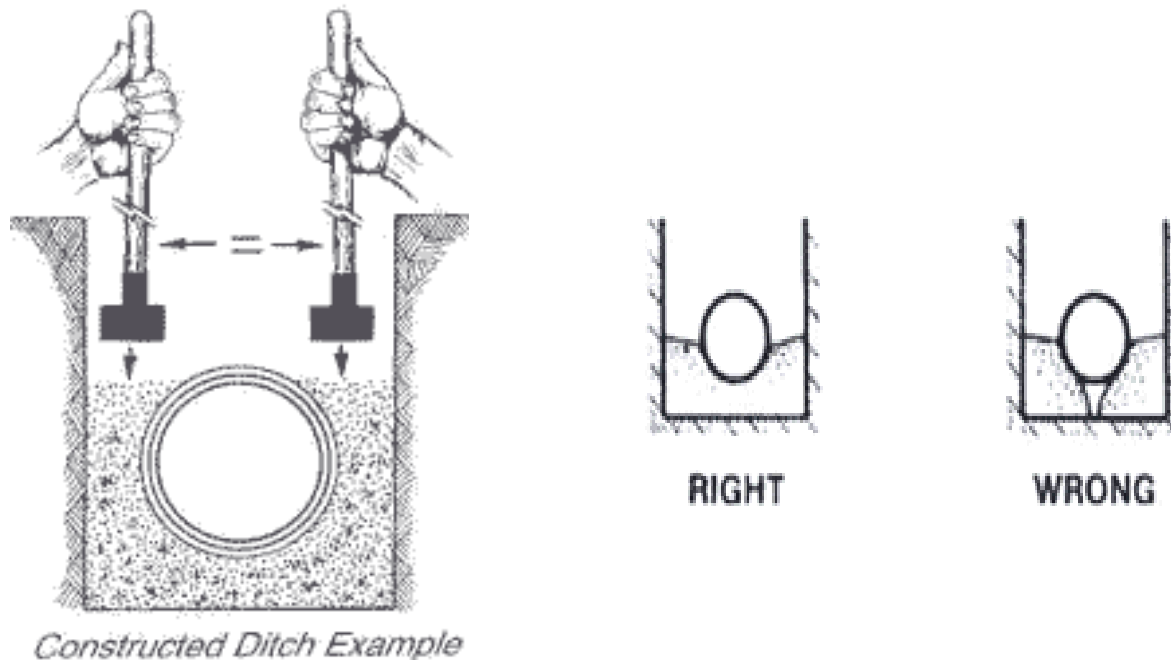
Thrust developed = 1,800 lbs. Load bearing strength of clay = 500 lbs.

$1,800 / 500 = 3.6$ square feet thrust block

Recent technological developments such as thrust restraints, hydrant restraints, pipe restraint rings, etc., can aid as an addition to thrust blocking or sometimes an alternative, particularly where the soil is disturbed or cannot be tamped at the time of excavation.

Backfilling

The purpose of backfilling is to provide support for the pipe, provide lateral stability between the pipe and the trench walls and to form a cushion around the pipe to protect it from potential damage. The backfill material should be clean and dry and small enough to fill voids and effectively protect the pipe. The material should contain enough material to be compacted properly to viably protect the pipe. Larger rocks, debris and frozen material should not be used to backfill. Fill the trench in stages and compact as each layer is installed. Tamping can be achieved by hand, mechanically or by saturation.



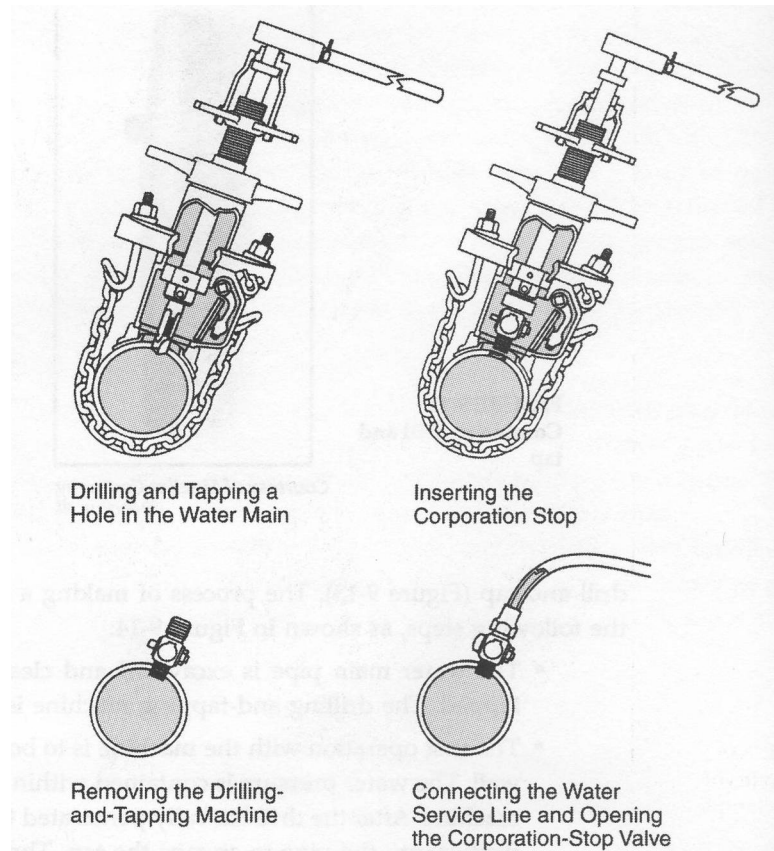
Pressure and Leak Testing

Testing for leaks and pressure should be accomplished before finish grading is done. The standard for pressure testing is 1.5 times the pipe's normal working pressure or 150 psi, whichever is greater, and held for a minimum of 24 hours. If your pipe is installed by an outside contractor, this testing before they leave can alleviate a great deal of exasperating and expensive work just to put the pipeline in service.

Tapping

Tapping, by definition, is the cutting in of threads on the inside of a drilled or cut hole. In the water business, tapping is the connecting of one water line to another.

Many utilities now use saddles for tapping because of the widespread use of plastic pipe. Connections can be made either when the main is empty (dry tap) or when the pipe is full and under pressure (wet tap). Fortunately, most tapping done today is wet tapping. This method is preferred because it lessens the risk of contamination and allows the connection to be made to the customer without service interruption to other customers.



The fitting used to connect small diameter service lines to water mains is called a corporation stop. It is usually a brass ball valve that is threaded on both ends and screwed directly into a water main or a tapping saddle.



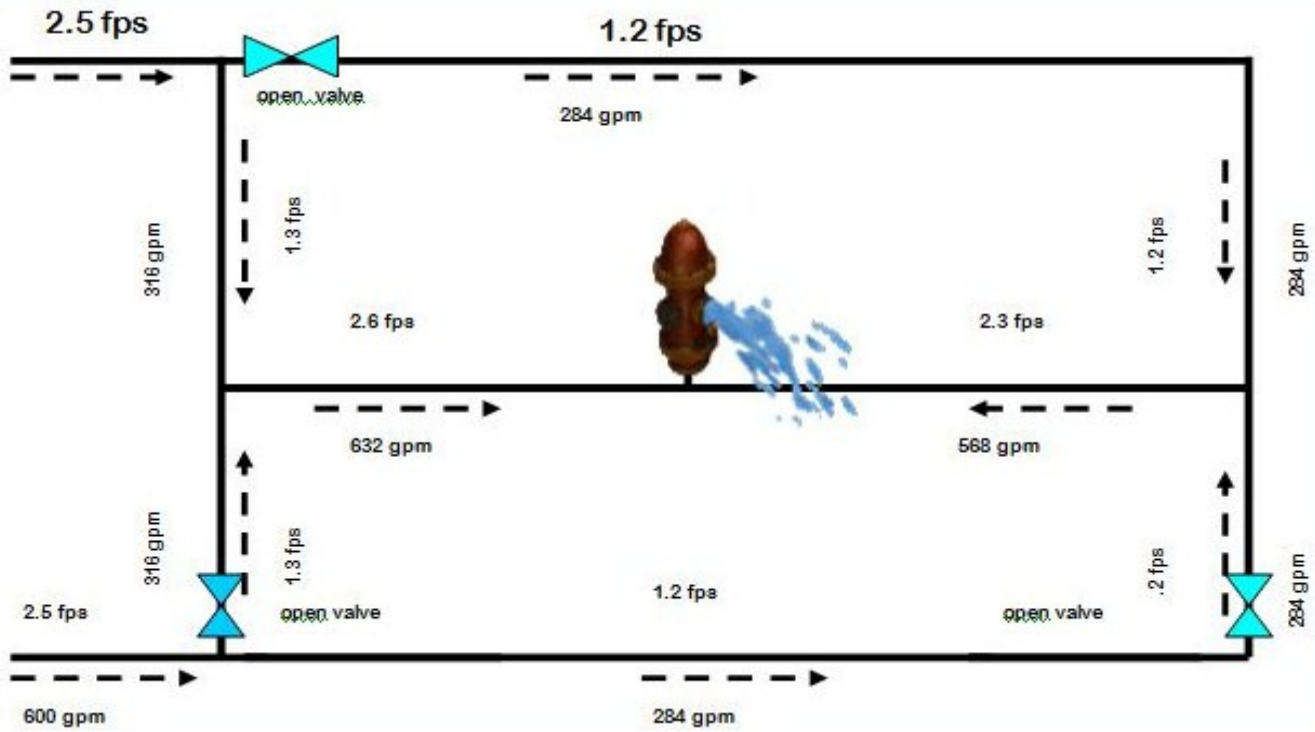
Larger taps are made by shutting off the main and installing a tee fitting or by tapping into an existing water main. The larger taps require more expensive tapping machines, as well as operational expertise, and many utilities farm them out to contractors.

Flushing and Cleaning

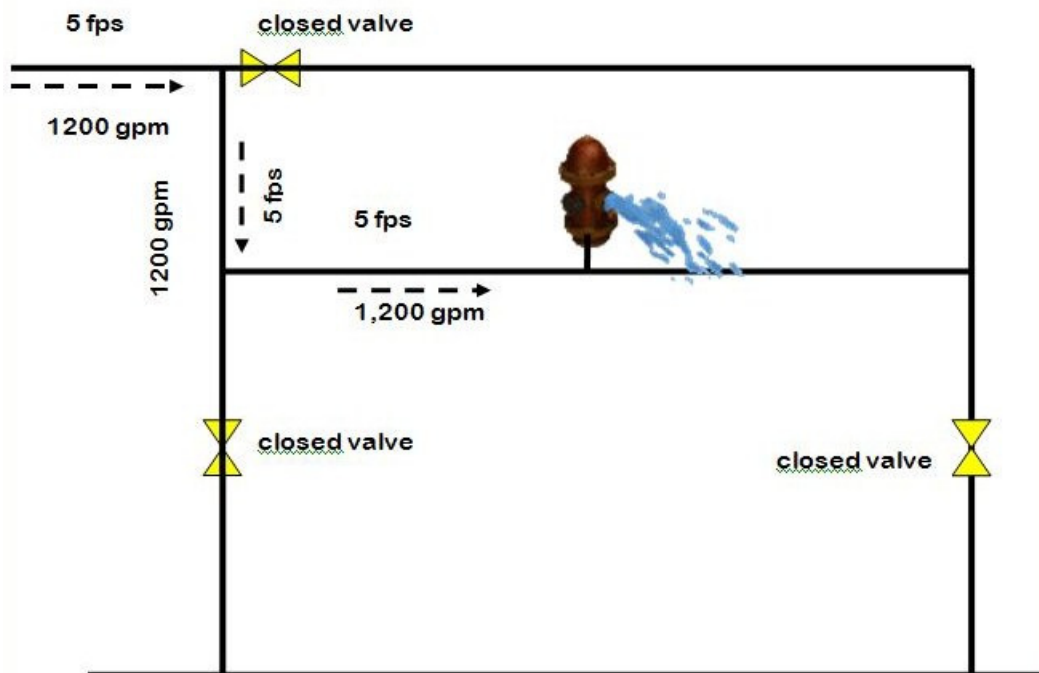
The flushing of water lines is absolutely necessary to maintain water quality in distribution systems. Flushing will generally occur for one of three purposes:

- 1) *To flush out debris or chlorinated water from newly installed or repaired lines.* After installation or repair of a water line, flush and pressure test, disinfect, dechlorinate and flush again before returning or putting the water line into service.
- 2) *To address localized problems such as taste, odor, color or coliform contamination.* When customer complaints or contamination monitoring necessitate a reactive form of flushing. Reactionary flushing means that the problem exists before any corrective actions are taken.
- 3) *To systematically and routinely clean the distribution system piping.* Planned, engineered, systematic flushing can achieve a degree of cleaning the distribution system dependent upon the condition of the pipes and the viability of your flushing program. This type of flushing is proactive and preventative and can, if done correctly, scour your distribution piping. Velocities must be a minimum of 2.5 feet per second (fps) with 5-6 fps being the optimum velocities necessary to fully flush your distribution system. Proactive **unidirectional flushing** will reduce customer complaints, improve water quality and prolong the life of your distribution system.

THE "OLD" WAY



UNIDIRECTIONAL



Mechanical cleaning could be necessary if tuberculation deposits are prolific or if biofilms, iron bacteria, or slime growths abound in large quantities. Try flushing initially, but if this proves inadequate, it may become necessary to use mechanical devices such as pigs or scrapers, to sufficiently clean the line. In addition to cleaning the lines, an increase in the carrying capacity of the lines will reduce pumping and energy costs as well as increase operating pressures close to where they were when the main was installed.

Cleaning can restore lines to close to their original viability but frequently this is a temporary reprieve from the problems associated with old or neglected mains. A change in water chemistry or the additions of corrosion inhibitors are necessary to prevent a reoccurrence of tuberculation problems.

A line replacement program should be planned for and implemented. These programs call for the replacement of a certain percentage or dollar amount of distribution system piping annually. This will also prevent the overall deterioration of the distribution system and spread the cost of replacement over a longer period of time than having to replace a line suddenly due to its failure.

In the United States, the average pipe age ranges from 21 – 91 years. The current replacement rate in the US is only 0.5% annually. Some of the oldest sections of pipe are over 125 years old. Distribution piping will not last indefinitely and water quality is usually adversely affected by the condition of these older pipes.

PIPE DIAMETER	CROSS-SECTIONAL AREA (SQUARE FEET)	GPM @ 2.5 FPS	GPM @ 5 FPS
2"	0.022	25	50
4"	0.087	98	196
6"	0.196	220	440
8"	0.348	390	780
10"	0.545	612	1224
12"	0.785	881	1762
16"	1.395	1565	3130
18"	1.766	1981	3962
24"	3.14	3523	7046
30"	4.91	5505	11010
36"	7.065	8078	16156

Biofilms

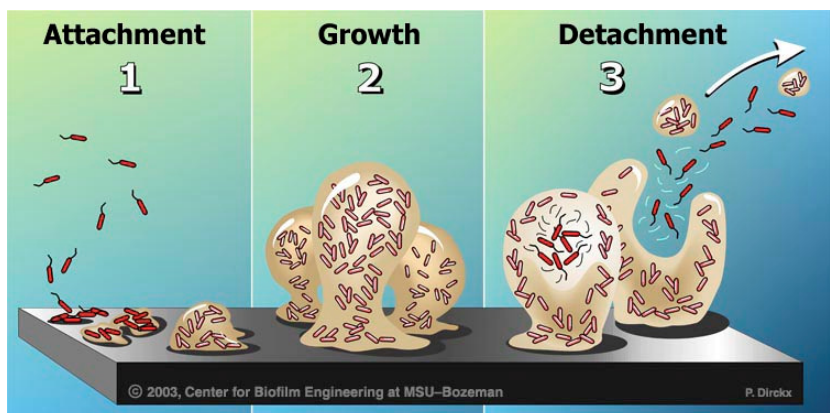
A thin layer of cells of a microorganism such as bacterium or fungus held to the surface of pipes by a material these microorganisms produce. These entities also produce and surround themselves with a gelatinous substance which makes them resistant to biocides. The most effective and cost-effective way to remove biofilms is flushing with velocities in the 5 fps range.

Factors Causing Biofilm Growth

- Presence of microbial nutrients in the water
- Characteristics of pipe walls, such as roughness or “C” factor
- Microbial and chemical quality of the water entering the distribution system
- Water temperature and pH
- Low chlorine levels in the water
- Velocity of the water

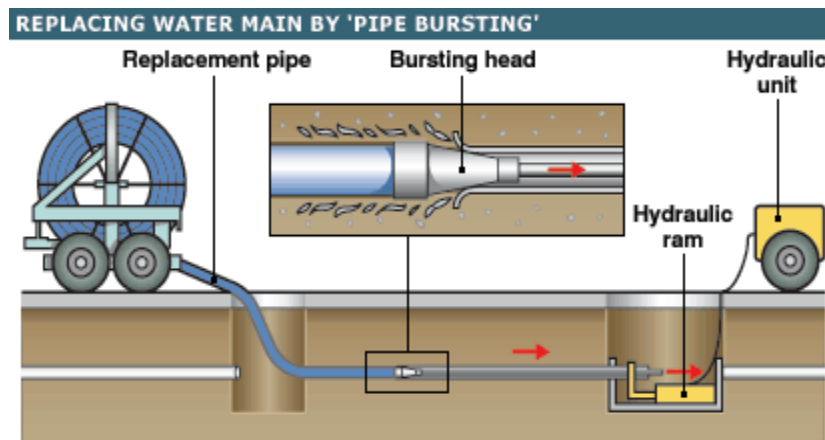
Characteristics of Biofilms

- Form on surfaces in aquatic systems
- “Resistant” to biocides
- Composed of organic and inorganic substances
- Monitoring is difficult
- Release organisms into bulk fluid (in our case, water)
- Pathogens Found in Biofilm
 - E. Coli
 - Legionella pneumophila
 - Pseudomonas
 - Anthrobacter
 - Sarcina
 - Micrococcus
 - Klebsiella
 - Enterobacter

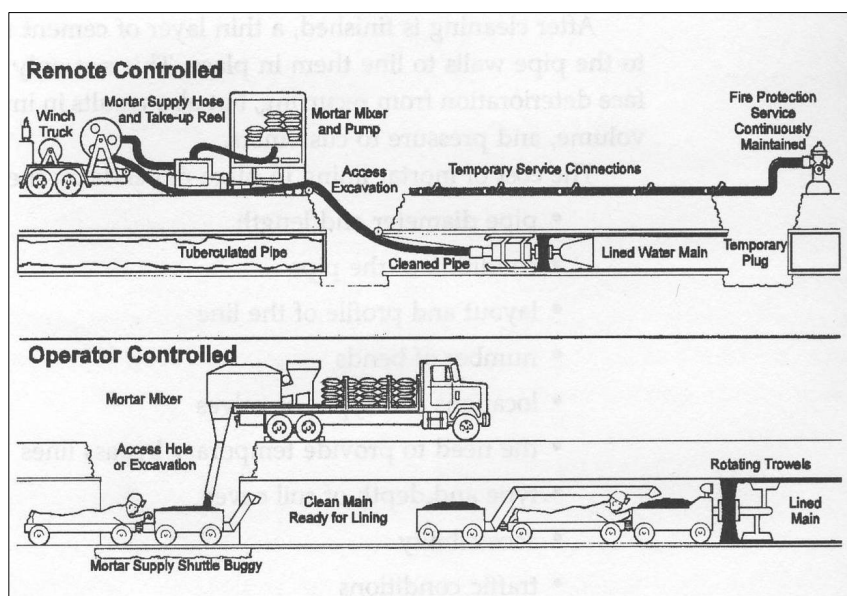


Trenchless Technologies

1) Pipe Bursting – a method where the existing pipe is “split apart” and left in the ground and the new pipe is pulled into place right behind the head that splits or bursts the existing pipe. HDPE pipe or a special type of PVC is butt welded or fused together to allow for the long pipe runs. A pipe larger than the original pipe can be pulled into place. Runs of 1500 feet or more are possible with this method.



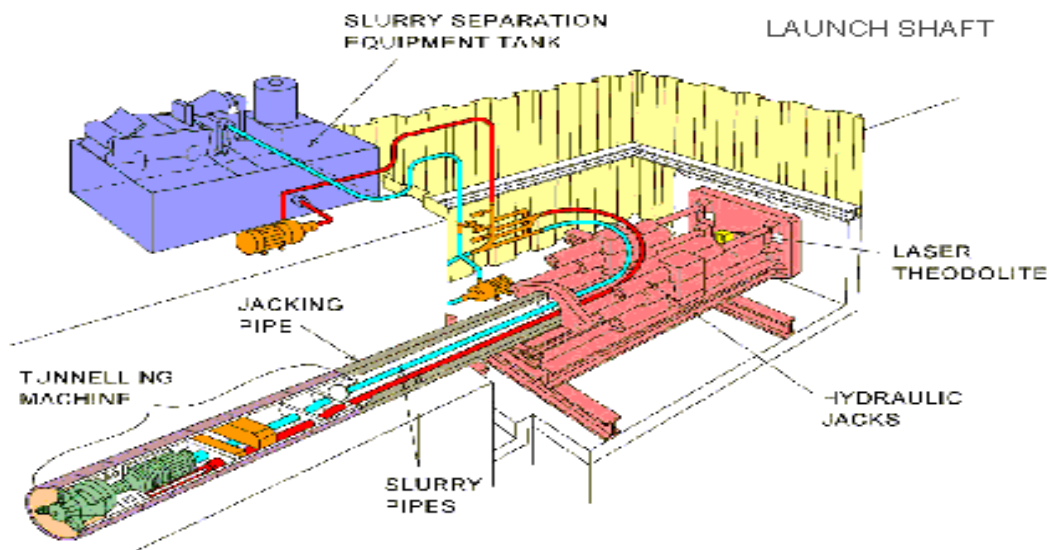
2) Slip Lining – a method whereby a replacement pipe is pulled into an existing pipe. This method is bound by the size of the existing pipe. Again, runs of 1500 feet are not uncommon.



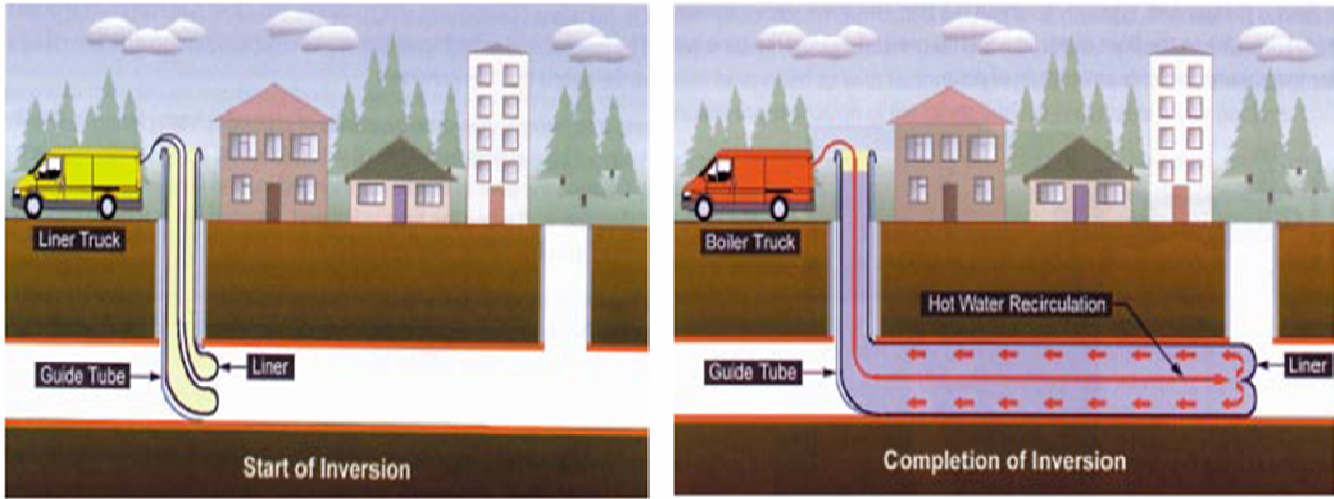
3) Pigging and/or mechanical cleaning – a mechanical means of removing mild tuberculation, scaling or biofilms from an existing pipe. Pigs are run through the existing lines by water pressure or mechanical scraper blades or rods are pulled through the existing pipe to clean it.



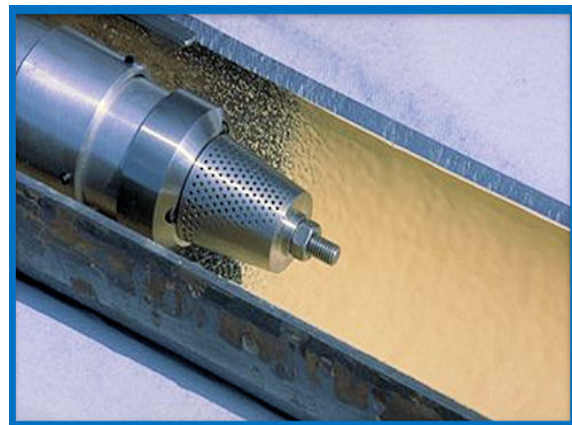
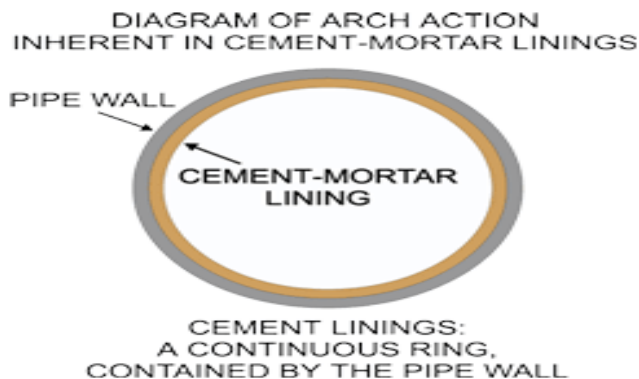
4) Pipe Jacking or microtunneling – a technique for installing underground pipes where hydraulic jacks are used with specially designed pipe behind a shield at the same time as excavation is taking place within the shield. Water is added at the point of excavation to form “slurry” to soften the ground for the excavating and recycled through lines to be separated in the slurry separation tank and reused for excavating.



6) Cured in Place Sliplining - a method whereby a liner is pushed or pulled through an existing pipe and “cured” or heated with steam, hot air, hot water, or UV light and relines the existing pipe. The pipe is usually folded and allowed to unfold or sometimes forced to unfold by pumping air through them.



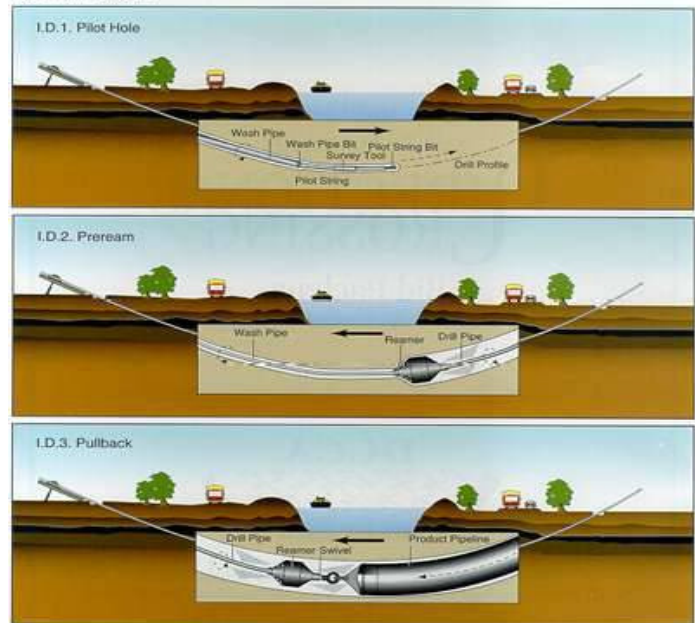
7) Cement lining – A high speed cement lining machine is pulled through an existing pipe followed by a trowel. The existing pipe is now relined with cement and its C- Factor returned to a high level. This method, since it relines an existing pipe, is bound by the size of the original pipe.



8) Horizontal Directional Drilling –

A machine that uses rods of varying lengths and diameters attached to a boring head. These rods can be directed to follow a predetermined route to follow the depth and routes desired and not disturb other utilities in the ground. Runs of 3000 feet are not uncommon. The original route is bored and then the product pipe is pulled back to the point of origin.

Figure 1. Technique



9) Moles - Pneumatically powered cylindrical tools laid in an excavated pit that bore a hole through rock and dirt toward a second excavated pit. Usually used for service connections.



There are more methods used for “trenchless” excavations, but these are among the most frequently used. With more and more utilities being laid underground, the viability of open cut trenching is getting harder all the time. One of these methods could be a favorable alternative to the traditional open- cut method of laying pipe and are extremely useful when attempting to rehabilitate a pipeline.

Leak Detection

Leak detection programs are a necessary task in a distribution system. The costs of treating water and never receiving any income from the water is foolish. Leak rates should not exceed 15%. All services should be metered and accounted for each month.

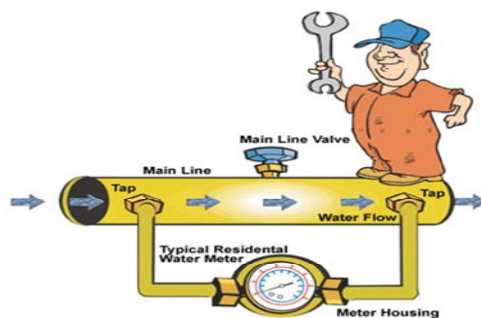
Leak Detection Methods

LISTEN

1. Pick valves that are suspected to be in the general area of your leak
2. Follow the directional flow of the water
3. Pick two valves in series and shut the second valve (#2) off and then go back to the first valve (#1) and shut it off. Wait ten minutes.
4. 4) Crack open valve #1 and put the valve wrench on the valve and listen.
5. 5) Since water under pressure moving through a small space makes a whistling or screeching sound. If there is no sound or if the sound suddenly stops then there is no leak between those valves.
6. 6) If the noise continues then the water is going somewhere, probably from your leak

Hydrant Pressure

- 1) Remove the hydrant cap and slowly flush line to remove any sediment.
- 2) Replace cap with a cap containing a pressure gauge.
- 3) Secure all hydrant caps and charge the hydrant.
- 4) Read the line pressure.
- 5) Slowly close the two valves on either side of the hydrant starting with the downstream valve (#2).
- 6) If both valves are tightly closed then the line pressure should hold.
- 7) If the pressure drops off quickly, water is leaving the main somewhere.

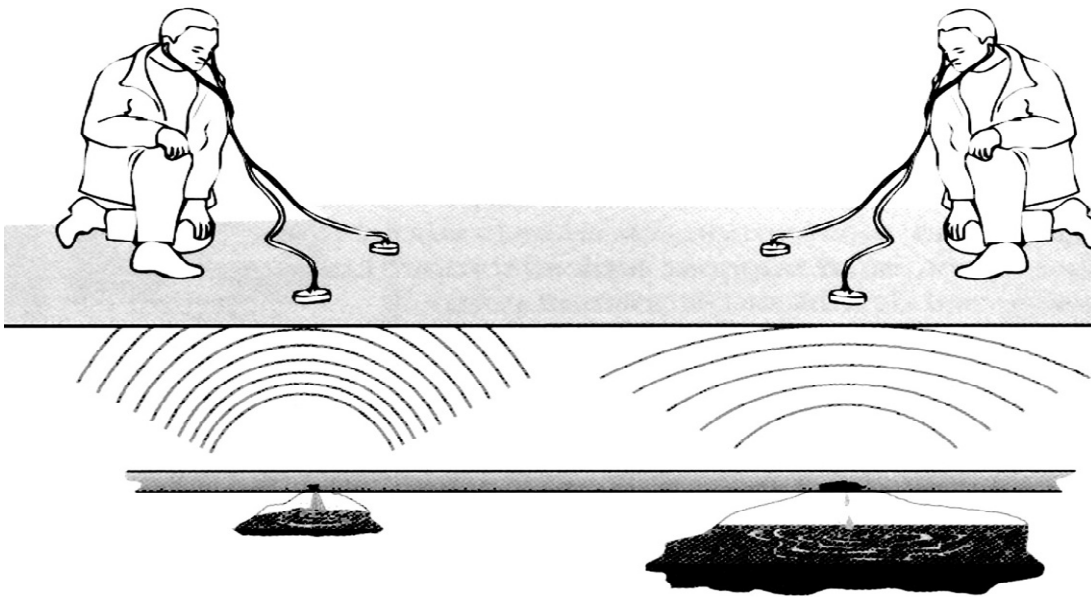


Permanent meter setting. (particularly viable on stream crossing, railroads, etc.)

- 1) Install a tap on both the upstream and downstream side of the valve on stream crossings, railroads, etc.
- 2) Install a meter between the two taps
- 3) If water is moving through the meter, a leak is present.

DETECTION DEVICES

Geophones – consist of two specially designed heavy brass plates connected to earpieces that resemble a doctor's stethoscope.



Portable electronic leak detectors – these use electronic sound amplification and go from basic models with just an on/off switch to ones that can tune out background noise.



Leak correlators – based on sound amplification, with a twist. A pair of small portable amplification and transmitting devices , usually magnetic, are placed in contact with a meter, valve, hydrant, exposed main etc., and spaced at varying distances dependent on the composition of the line. The signal from each device is transmitted to the correlator. If line size, location of the line and pipe material is known the correlator screen can show where the leak is located. Usually within three feet or so.



Chapter 6 Review

1. List three methods of trenchless technologies.
2. Unaccounted for water should be at _____ or less.
3. Anytime water changes directions or stops, what should be added to the pipeline?
4. _____ are microorganisms that attach themselves to pipe walls.
5. _____ flushing is an engineered, more viable form of flushing.

Answers for Chapter 6 Review

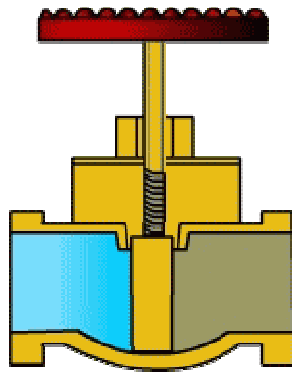
1. Pipe Bursting, CIP, Slip Lining, Epoxy Lining, and Cement Lining
2. 15%
3. Thrust blocks
4. Biofilms
5. Unidirectional

Chapter 7: VALVES

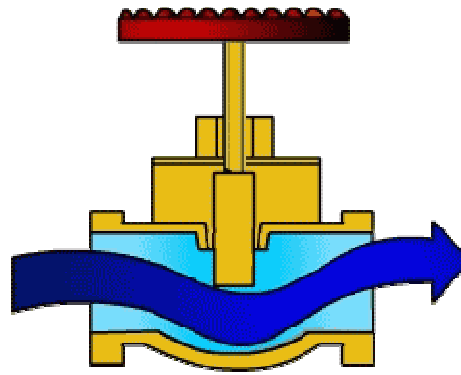
Chapter 7 Objectives

1. Describe the functions as well as the strengths and shortcomings related to the different types of valves (gate, butterfly, globe, plug, ball, air/vacuum release, pressure reducing, altitude, and check).
2. Explain the importance and purpose of a valve maintenance program.

Gate Valves



Gate Valve Closed

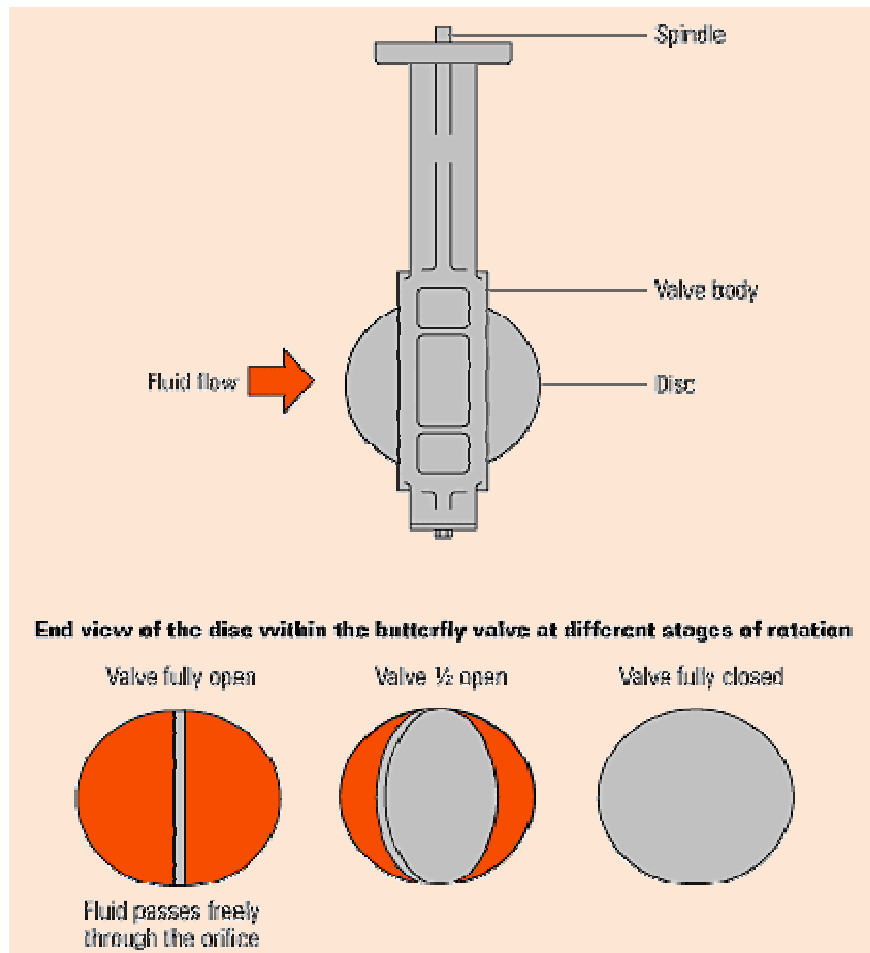


Gate Valve Opened

- Operated in a fully open or fully closed position only
- Used for isolating a part of the distribution system or a piece of equipment (i.e. fire hydrant, pump etc.)
- NOT used for throttling
- Very low headloss and turbulence in the fully open position
- Not the best choice if needed to operate frequently
- Thrust blocking and/or mechanical restraint must be used if installed at a dead end or water direction change
- Types – solid wedge, double disk, solid split gate
- Maintained by exercising a full cycle operation on a regular schedule – at least annually
- Valve card notations – number of turns, direction of open, location, type, size, date of installation and maintenance, inspection and/or repair.

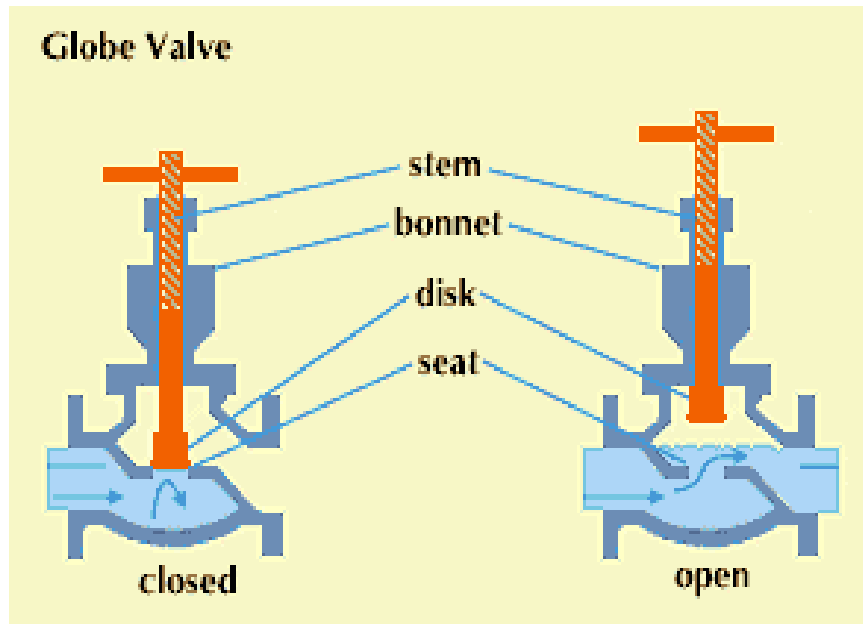
- Leakage repair – O-rings, seals, gaskets, or mechanical joint gaskets

Butterfly Valves



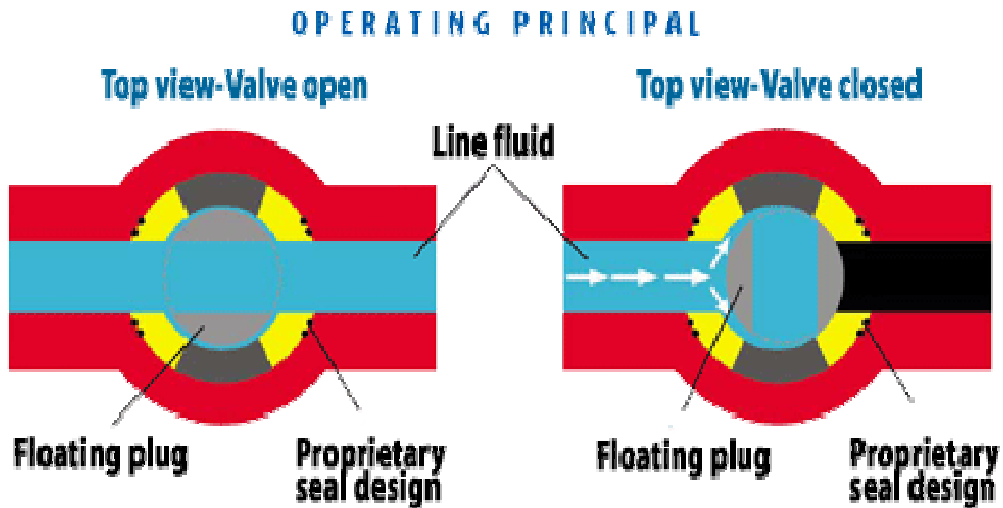
- Primarily used for isolation, but can be used for throttling or even as a check valve.
- Wears better than gate valves
- Shorter, lighter, and cheaper than gate valves in most cases
- Higher headloss than gate valve and prevents the relining or mechanical cleaning of pipes.
- Install in a closed condition
- Minimal maintenance requirements, but removal is necessary for repair

Globe Valves

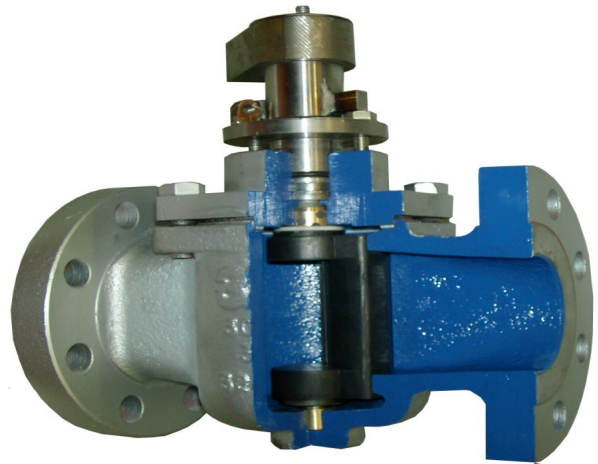


- An extremely versatile valve
- Used primarily for pressure
- Highest headloss of all valves operating in a fully open position
- Can be used to control pressure, throttling (flow or altitude) and also isolation in small flow or line size applications
- Design aids in disc wear so maintenance is seldom necessary, but removal is compulsory for repair.
- Design also excludes mechanical pipe cleaning or relining without valve removal

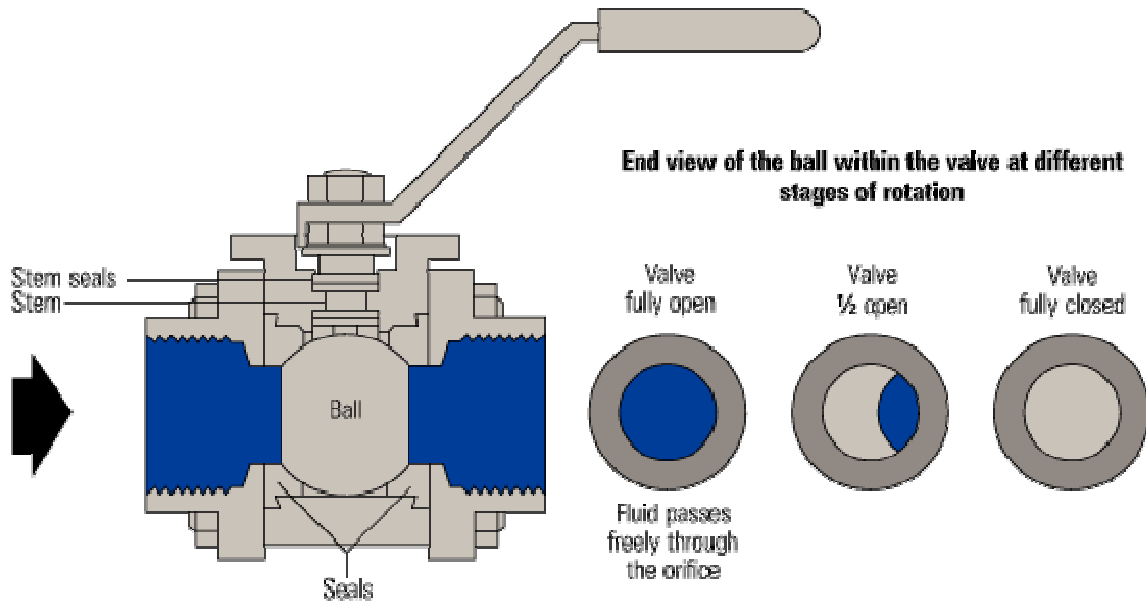
Plug Valves



- Used for isolation where frequent operation is required
- Not recommended for throttling unless especially designed as such
- Design minimizes closing surge and allows for longer valve life than a gate valve
- Generally heavier and more expensive than a gate valve
- Low headloss, dependable, easy operation

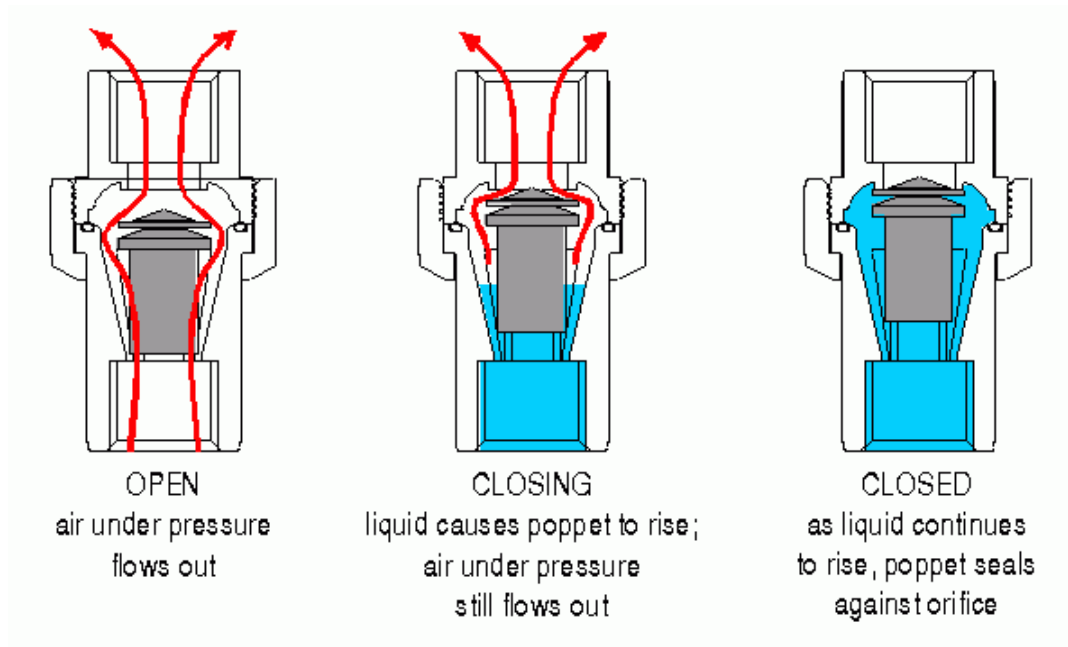


Ball Valves

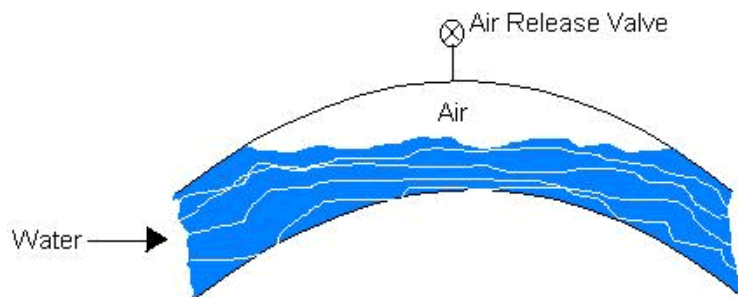


- Really an improved version of the plug valve
- Same characteristics as a plug valve except much lower turning torque required and no metal to metal contact, which extends the life of the valve
- We primarily use these in our service applications (corporation stop)

Air/Vacuum Release Valves

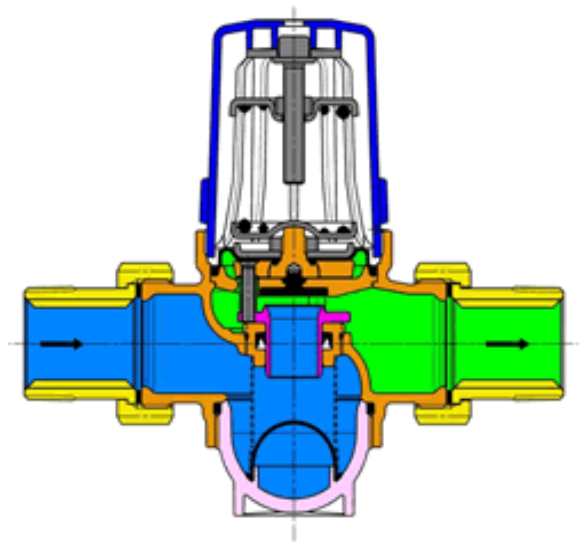
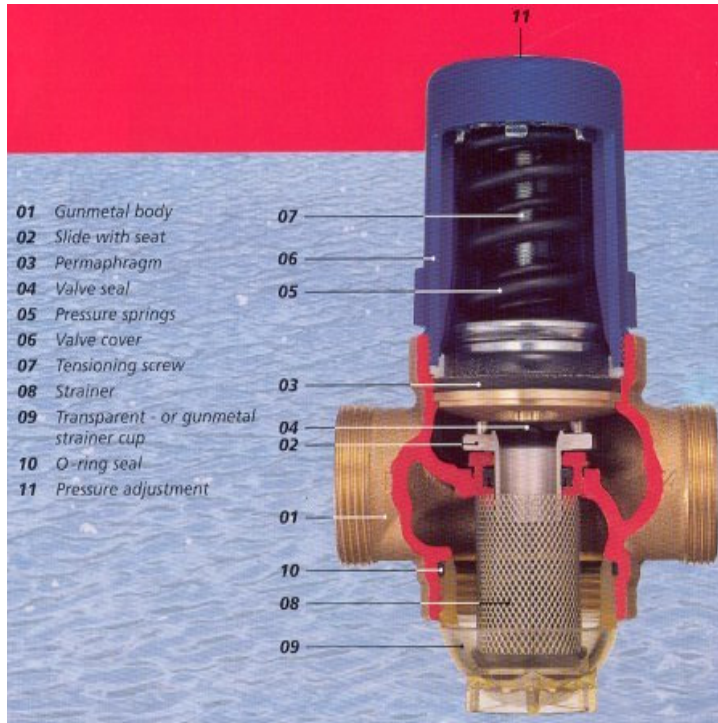


- Air release valves allow the removal of air from water lines due to turbulence or temperature variances which could collect and act as a partially or fully closed valve.
- Vacuum release valves relieve the occurrence of negative pressure due to vacuum conditions that could collapse our pipes.
- Vacuum release valves allow air to enter the pipeline while draining.
- Both air and vacuum relief valves should be installed at high points in the distribution system and atop the mains.



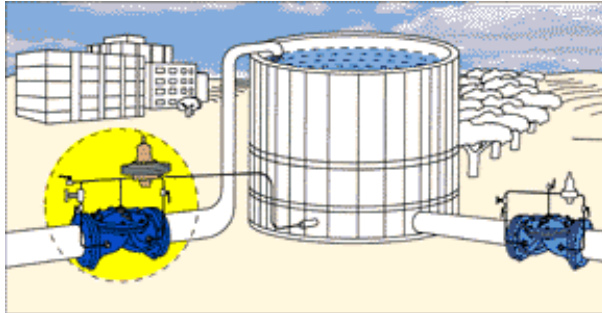
(They may be installed as combination valves or singularly in tandem.)

Pressure Reducing Valves



- Usually globe type valves with a spring loaded diaphragm
- PRV's automatically maintain constant outlet pressure regardless of changes in the flow rate or upstream pressure.
- These valves will hold a partially open position in response to the settings of the spring-loaded diaphragm.
- Pressure reducing valves will throttle in this open position relative to the balance of opposing forces.
- The use of two pressure reducing valves in sequence can be used to handle a wide range of flow rates.

Altitude Valves



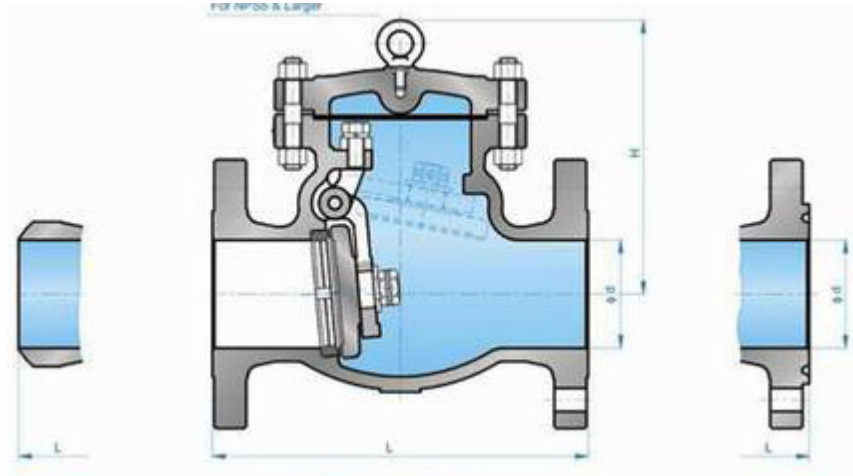
Controls tank levels below the hydraulic grade line

Four types:

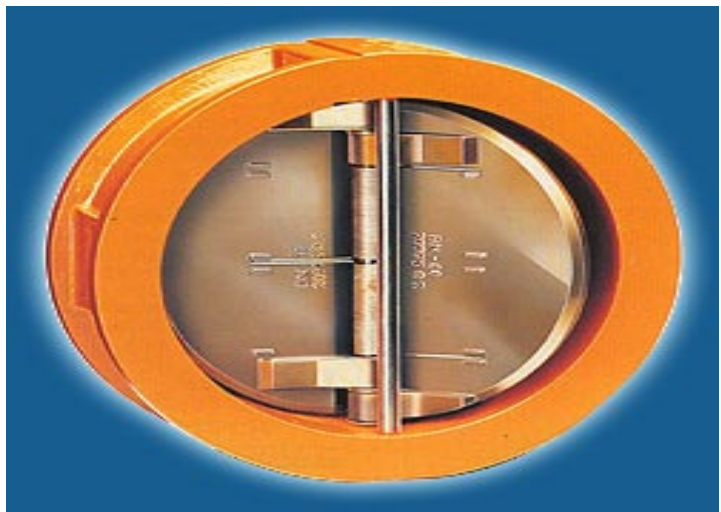
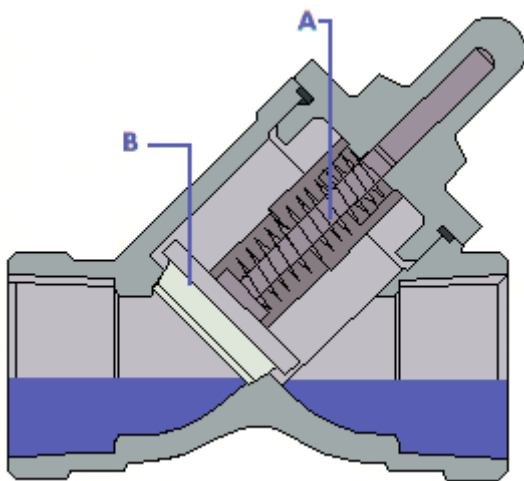
1. Double Acting – closes when the level in the tank reaches high water level; opens for return flow out when line pressure falls below static reservoir head
2. Double Acting with Differential Control – works like double acting valve, except it can be set to activate under preset pressure parameters
3. Single Acting – water discharges from the tank through a separate line. Valve actuates to refill tank when the water level falls 4 – 6 inches below the high water level.
4. Single Acting with Differential Control – works similarly to a single acting, except the valve actuation level can be preset anywhere from 1 – 15 feet.



Check Valves



- Prevent the reverse flow of water or prevent the water from traveling in an unintended direction
- Many controls are available to actuate the valve, such as swing gate (illustrated above) or spring loaded (illustrated below).



- Used to maintain head pressure on the effluent side of a pump or maintain the integrity of some meters

Chapter 7 Review

1. Explain the purposes of each type of valve and what application each is designed for.
2. Where more than one type of valve could be used in a particular application, why would you choose the valve you do?
3. Which type of frequently used valves are much more maintenance intensive than the others?
4. What valve is used to attach water services to water mains?
5. Which valve can be adjusted?

Answers for Chapter 7 Review

1. Gate – Isolation
 - Butterfly – Isolation and throttling
 - Globe – Throttling, pressure, altitude
 - Plug – Frequent isolation
 - Ball – Service connections
 - Air/vacuum – To release air from lines and to relieve vacuum conditions in a pipeline and to drain a pipeline.
 - PRVs – Maintain pressure
 - Check valves – allows the movement of water in only one direction
2. Applicability
3. Butterflies
4. Corporation stop
5. PRV

Chapter 8: HYDRANTS

Chapter 8 Objectives

1. Differentiate between the two major types of hydrants (wet barrel and dry barrel).
2. Describe the hydrant components (bonnet, upper barrel, lower barrel, and shoe).
3. Describe the proper order of installation for a hydrant.
4. Identify the purpose of flushing.
5. Describe how to distinguish a fire hydrant from other types.
6. Identify the purpose of blow off hydrants.
7. Identify how to obtain flow measurements.
8. Describe proper maintenance procedures for hydrants.

Two classifications of hydrants made in the United States are the wet barrel and the dry barrel hydrant. Keep in mind that the MINIMUM SIZE LINE USED FOR A FIRE HYDRANT TO SATISFY FIRE FLOW AND PRESSURE REQUIREMENTS IS 6 INCHES.

Wet barrel hydrants contain the operating valve in the nozzle section of the hydrant.

- Operated by rotating the operating stem on each of the outlets, with each discharge port being independently valved
- Used in warm weather climates
- Their main advantage is the ease in which a second fire truck can be connected
- Not used in Kentucky because of variations in temperature.



Dry barrel hydrants have the operating valve installed below ground level.

- These hydrants are equipped with special drain valves or weep holes that allow the above ground portion of the hydrant to automatically drain when the hydrant is not in use.
- Used in colder climates where there is a likelihood of freezing temperatures.
- If equipped with a traffic kit, the hydrant will not usually lose water if struck.

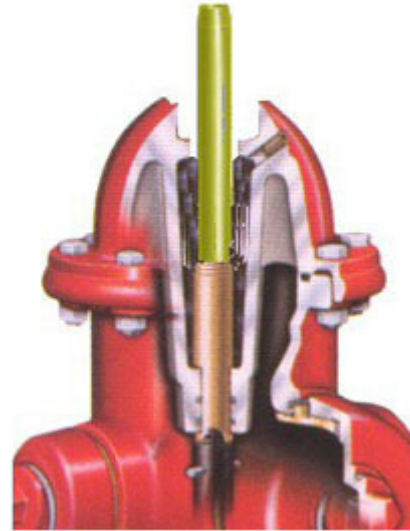


*****FIRE HYDRANTS SHOULD ALWAYS BE OPENED AND CLOSED SLOWLY.*****

Hydrant Components

Bonnet

The top of the hydrant contains the bonnet. The bonnet protects the packing and the operating nut mechanism.

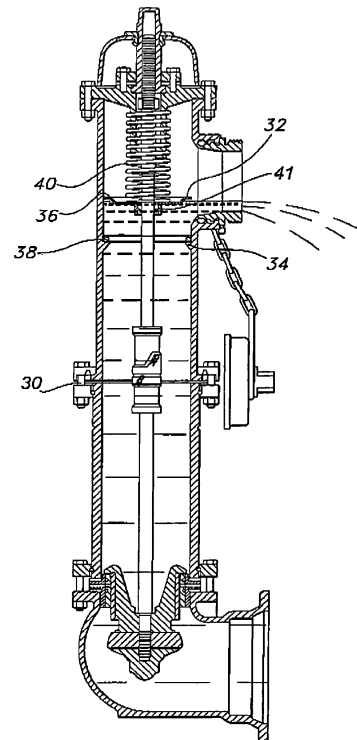


Upper Barrel

The upper barrel contains the outlet nozzles and the “O” rings that prevent water from entering the bonnet of the hydrant. Common nozzle sizes include 2½ inch hose nozzles and usually four or six inch pumper nozzles.

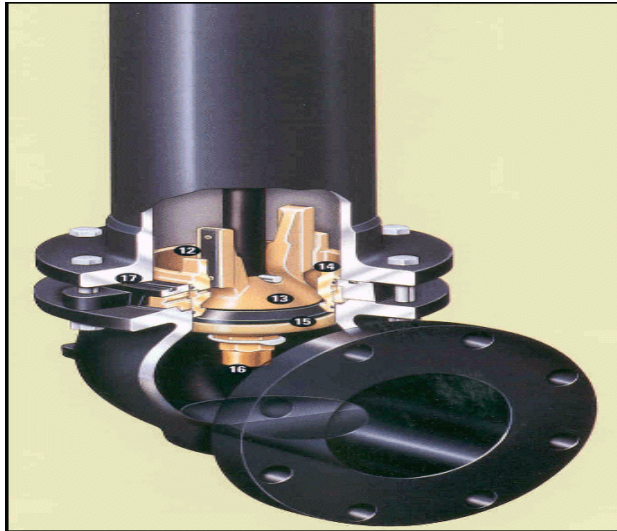
Lower Barrel

These sections determine the height or bury depth of the hydrant. The bury or bury depth is the distance from the bottom of the inlet connection to a point four inches below the flange that connects the lower barrel to the upper barrel.

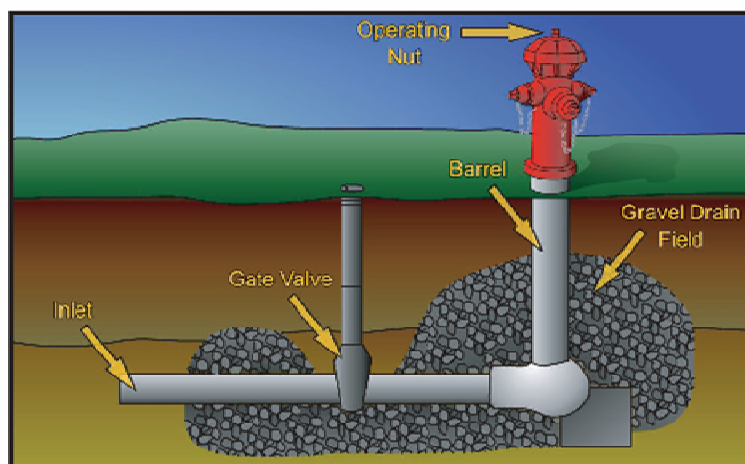


Shoe

The shoe is where the inlet connection is made. The shoe also changes the direction of the flow from horizontal to vertical. On most compression hydrants, the main valve assembly is located either in the shoe or directly on top of it. The size of the main valve is referred to as the main valve opening (MVO). This is the inside diameter of the valve seat and sizes that range from four through six inches are common.



The shoe also contains the drain valve assembly and/or weep holes, used to drain the hydrant when it is closed.

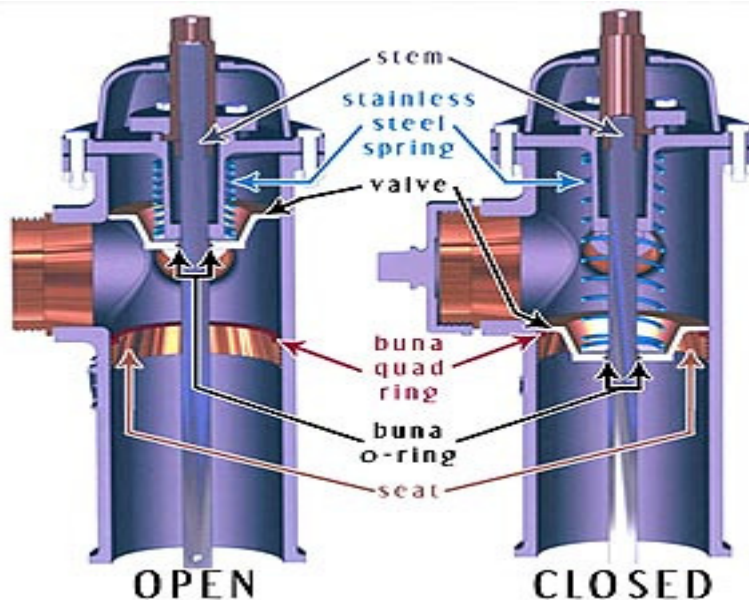


7.1.3 Figure 1: The basic fire hydrant

Proper installation order on a fire hydrant is main, gate valve, pipe, and thrust block. Fire hydrants should be inspected and maintained at least annually.

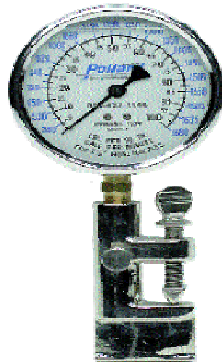
Compression Type Hydrants

In compression hydrants, the main valve moves jointly on a vertical axis against a seat located in the hydrant base. The valve moves against the seat to close and away from the seat to open. A vertical stem moves up or down when the operating nut is rotated. The valve may be located below the seat and opens against the pressure or above the seat and opens with the pressure.



Pitot Tubes, Diffusers, and Dechlorinators

Pitot tubes can be utilized in the distribution system to obtain flow measurements in gallons per minute and pressure (PSI) readings from fire hydrants.



Diffusers redirect the flow from the hydrant and somewhat dissipate the force or intensity of the flow in order to reduce potential property damage and hazards to motorists during flushing. Some diffusers are equipped with pressure and flow gauges and some have dechlorinating capabilities.



Always dechlorinate the water when flushing. Chlorine flowing into streams will damage the ecosystem of that stream. Sodium thiosulfate is commonly used to remove chlorine from water.

Blow Offs

Blow off valves are used to drain lines, remove accumulated sediment via flushing and resolve customer complaints. Blow offs are used at dead ends where fire hydrants are not needed. All blow offs should be flushed regularly to maintain water quality.



Chapter 8 Review

1. Hydrants should be exercised and lubricated on a _____ basis at a minimum.
2. One component that can prevent the successful opening or closing of a hydrant is the _____.
3. Water released from hydrants and blow-offs needs to be, by regulation, _____ before the water hits the ground.
4. What is located in the show of the hydrant that allows water to drain from the hydrant?

Answers for Chapter 8 Review

1. Twice a year (bi-annual or semi-annual)
2. Stem, operating nut
3. 0.001 PPM
4. Weep holes

Chapter 9: METERS

Chapter 9 Objectives

1. Identify the purpose of meters.
2. Describe issues to consider when choosing a meter.
3. Identify different types of meters and the advantages and disadvantages of each type.
4. Identify proper placement and installation techniques.
5. Describe the benefits of Automated Meter Reading.
6. Determine the amount of storage necessary to functionally operate a distribution system.
7. Describe the suggested acceptable rate of water loss.

Water Loss

- The United States possesses a 34,000,000,000-gallon-a-day supply.
- 6,000,000,000 gallons of water per day is water loss or non-accounted for public use.
- There are approximately 53,000 water systems in the United States.
- The average water loss is 113,207,540 gallons per system.
- At a rate of \$3.00 per thousand gallons, the average revenue loss per system equates to \$339,621 per system.

Because so many systems have ignored or under-maintained their distribution systems, these figures can only increase.

Purpose

1. To measure and display the amount of water passing through it.
2. To prevent wasted water by reducing consumption.
3. To enable accurate billing for the volume of water used by each customer
4. To measure the volume of water produced in relation to the volume of water used so water losses can be estimated and tracked.

When choosing a meter always take into account:

The ability to measure and register anticipated flow levels.

The ability to meet required capacity with minimal headloss.

The durability and ruggedness of the meter.

Is that meter supplier a known and reputable manufacturer?

Is the meter easy to maintain and repair?

Is there a sufficient quantity of spare parts and competent technical services?

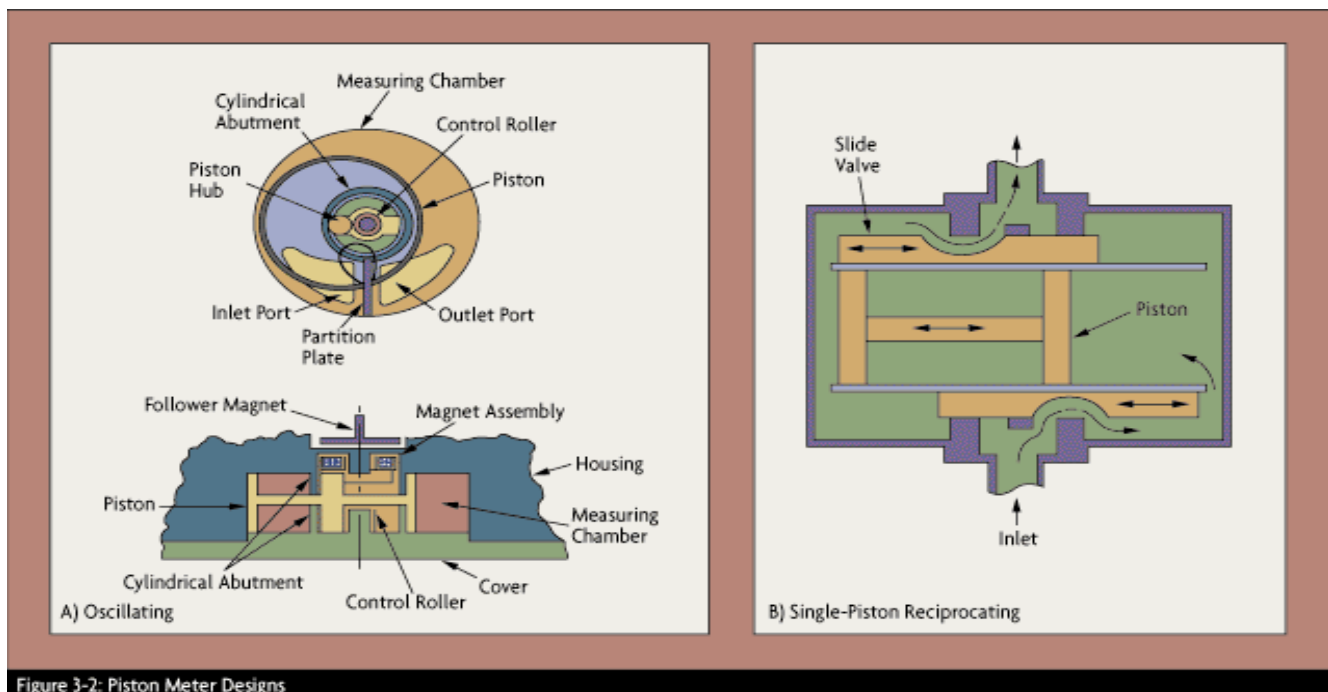
Types of Meters

Meters can usually be classified as small-flow meters, large-flow meters and combination large- and small-flow meters.

SMALL-FLOW METERS

Displacement Type

Piston: Measures the number of times the chamber (of a known volume) is filled and emptied. A piston or disc goes through a particular cycle of motion, and its movement is transferred to a register either notated in gallons or cubic feet per second.



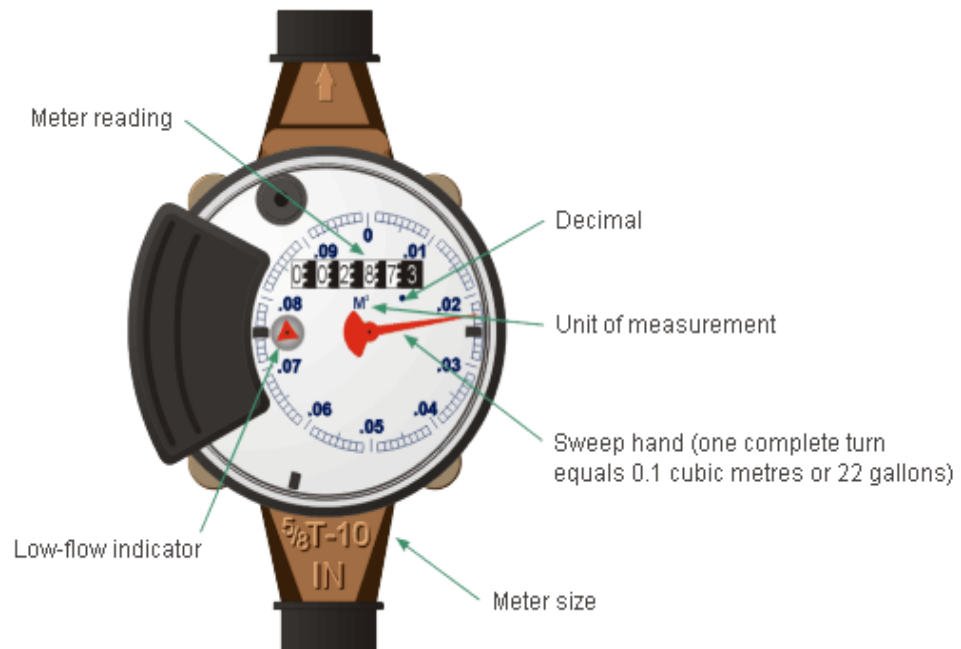
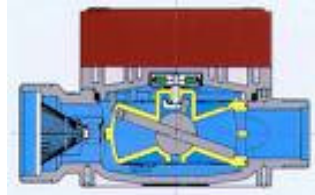
Advantages:

- Measures wide variations of flow rates in preset parameters
- Accurate at measuring low flows

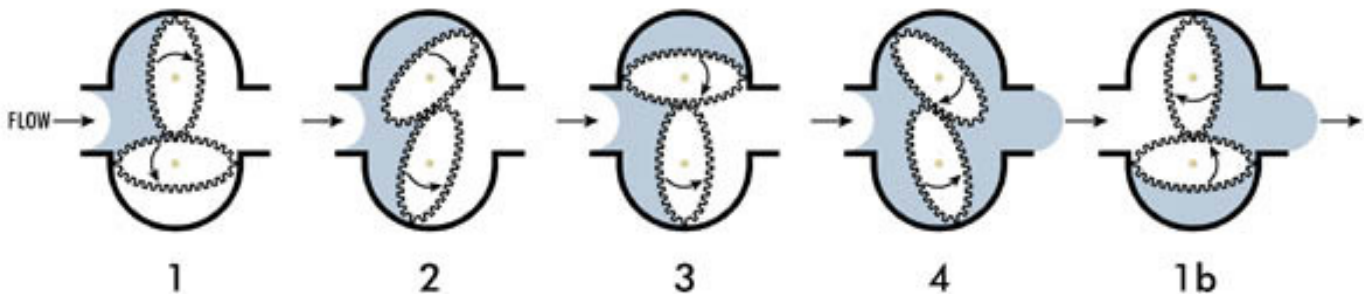
Disadvantages:

- Limited capacity (less accurate at greater sizes)
- Corrosion, scale or other foreign matter will cause meter to under-register or stop

NUTATING DISC



POSITIVE DISPLACEMENT



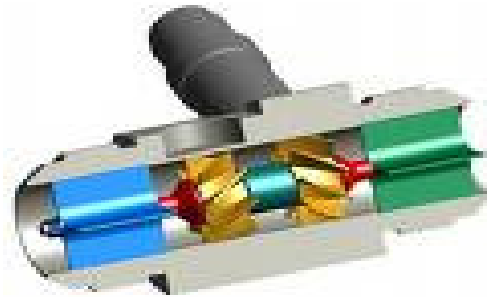
VELOCITY METERS

Normally used for pumps with continuous high-flow rates, main transmission lines, irrigation, industrial and other high-flow applications. Available in sizes anywhere up to 36 inches and above. Most measure the velocity of flow past a known cross-sectional area.

Venturi Meter



Turbine Meter



Propeller Meter



Advantages

- Handles high-flow rates without damage to working parts.
- Rugged and easy to maintain
- Low headloss relative to high-flow rates

Disadvantages

- Unreliable at low flows
- Mainly used for continuous high-flow applications, such as irrigation, industrial, large golf courses and other high-flow applications

COMPOUND METERS



Used for accurate measurement of widely varied flows

Combines both the velocity and displacement meters into one unit

The amount of headloss controls which unit is being utilized during a given time

Advantages

- Will measure a wide variance of flows from entities, such as hospitals, distilleries, factories, etc.

Disadvantages

- Friction or headloss is higher than velocity meters
- Accuracy is limited during changeover from one meter to the other
- Large, heavy, cumbersome and expensive

ELECTRONIC METERS

Measures a differential in electrical (magnetic) current passing through a cross-sectional area.



SONIC METERS

Uses sound waves to measure the flow of water

Advantages

- Very accurate
- No headloss

Disadvantages

- Accuracy diminishes by anything that changes the velocity of the water. (i.e. pumps, valves, elbows).
- At least ten (10) diameter lengths of unobstructed pipeline upstream are needed from the flow meter to be considered useful and accurate.



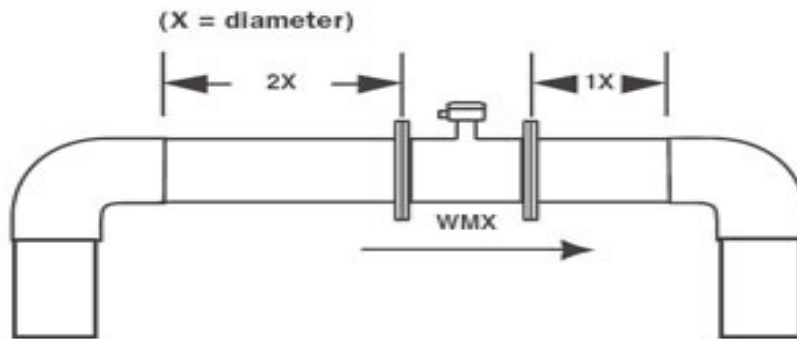
PROPORTIONAL METERS

- A certain amount of the total flow is diverted and measured through a second bypass meter
- The register that displays the usage is proportionally geared
- Relatively accurate except for low flows
- Principally used for fire lines



MAG METERS

A magnetic flow meter (mag flowmeter) is a volumetric flow meter that does not have any moving parts. The operation of a magnetic flowmeter is based upon Faraday's Law, which states that the voltage induced across any conductor as it moves at right angles through a magnetic field is proportional to the velocity of that conductor. For obtaining viable measurements, there should be a length of straight pipe 2 times the diameter of the meter in front of the mag meter and a length of pipe equivalent to 1 times the diameter of the meter after the placement of the meter. Elbows within this range will negatively affect the accuracy of the flowmeters.



Faraday's Formula:

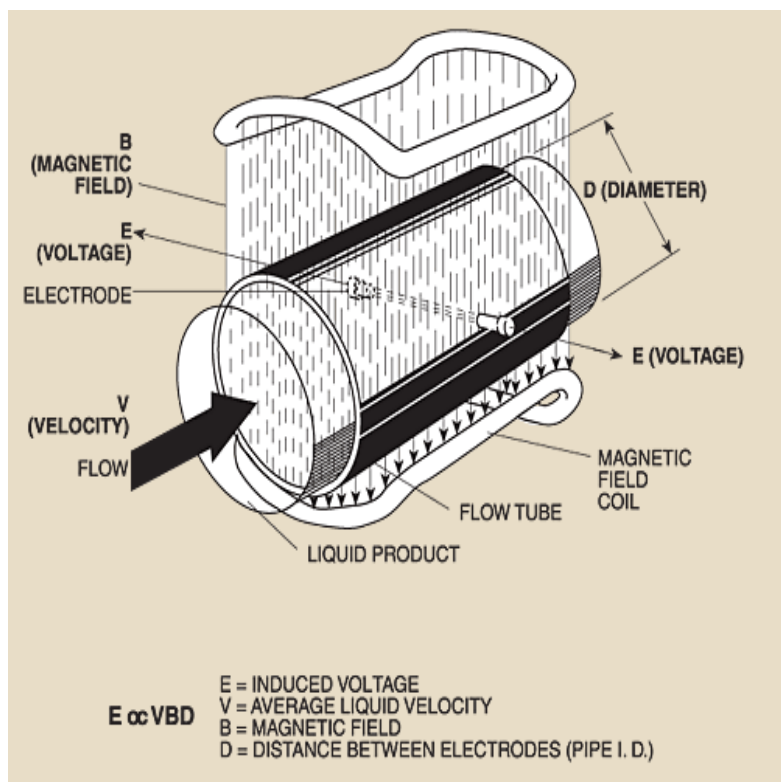
E is proportional to $V \times B \times D$ where:

E = voltage generated in a conductor

V = velocity of the conductor

B = magnetic field strength

D = length of the conductor



AMR (Automatic Meter Reading)

AMR's is an improved method of recording water usage. The old fashioned way sends meter readers out every month to open every meter vault and record the volume used for that month has now become improved. AMR's has changed how distribution systems record water usage. Benefits will be recognized by the utilities with increased efficiencies, outage detection, backflow detection, tamper notification and reduced labor cost as a result of automating reads, connections and disconnects.

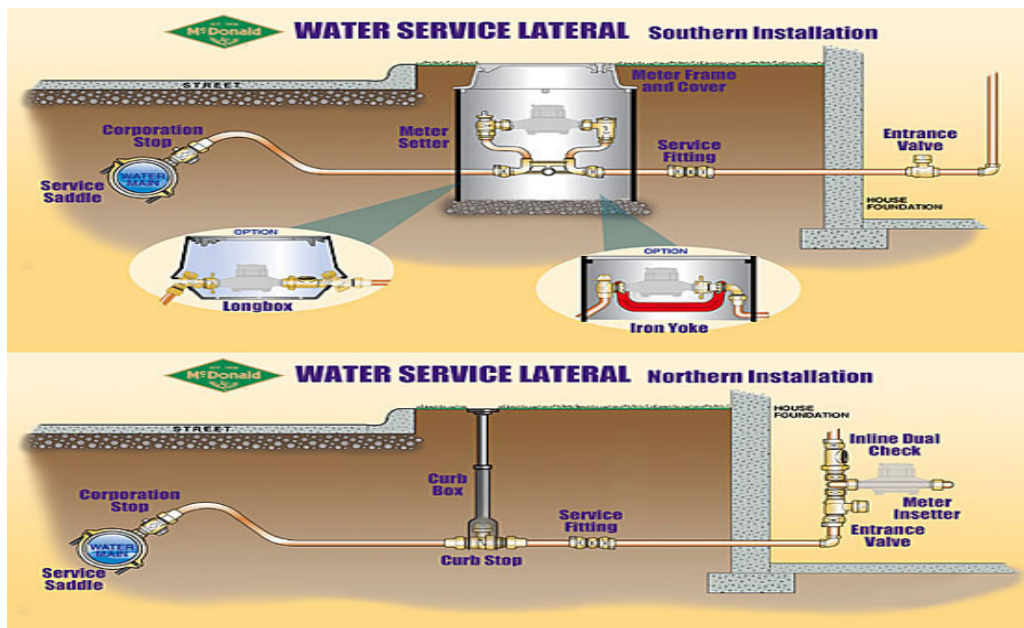
Accuracy Requirements for New Meters^{a b}

AWWA Standard Designation and Type of Meter	At Normal Test Flow %	At Minimum Test Flow %
C700 Displacement Type	98.5 – 101.5	95
C701 Turbine Type	98.0 – 102.0	Not Required
C702 Compound Type ^c	97.0 – 103.0	95
C703 Fire Service Type ^c	97.0 – 103.0	95
C704 Propeller Type	98.0 – 102.0	95

a. AWWA Standards

b. There is no standard for the accuracy of repaired meters, but the practice at many utilities is to require the same accuracy as for new meters at the normal test flow and at least 90 percent at the minimum test flow.

c. The accuracy during the "change over period," which is defined in the applicable standards, shall not be less than 85 percent.



****Water meters should be installed behind the curb and protected in a meter vault.****

Water loss, sometimes referred to as non-accounted for or non-revenue water, is something EVERY water system should be aware of and should also be endeavoring to reduce. The Capacity Development Team from the DOW will scrutinize every system's water loss figures. **An acceptable percentage of water loss is < 15%.** There are a number of different formulas to determine the amount of non-revenue water, but two of the simplest are included below.

1. Tucumcari treated 96 MG during December of 2009. Records indicate that Tucumcari billed 88,673,249 gallons for December 2009. What is their percent of water loss?

a. 10.8 % b. 7.6 % c. 9.2 % d. 12.3 %

$$\begin{array}{r} 96,000,000 \text{ gallons treated} \\ - 88,673,249 \text{ gallons billed} \\ \hline 7,326,751 \text{ gallons "lost"} \end{array}$$

$$7,326,751 \text{ gallons} \div 96,000,000 \text{ gallons} = 0.076$$

(to get percentage, multiply .076 X 100 = 7.6%)

2. Pigeon Drop billed 31, 928,735 gallons in November of 2009 and treated 37,264,048 gallons. What is the percent water loss if the meter error has been determined to be 2%?

a. 14% b. 12% c. 10% d. 8%

$$\begin{array}{r} 37,264,048 \text{ gallons treated} \\ - 31,928,735 \text{ gallons billed} \\ \hline 5,335,313 \text{ gallon difference} \end{array}$$

$$\begin{array}{r} 5,335,313 \text{ gallons non-accounted for water} \\ \div \\ 37,264,048 \text{ gallons treated} = 0.14 \text{ or } 14\% \end{array}$$

$$14\% \text{ non-revenue water} - 2\% \text{ meter error} = 12\% \text{ water loss}$$

Remarks:

Chapter 9 Review

1. Which meters have no moving parts?
2. What happens when you install an undersized meter in a supply line?
3. The two types of meters commonly used for residential services are the _____ and the _____.
4. When a meter has excessive wear, what will be the results in relation to registering flow?
5. Which meter has the lowest headloss?
6. If the city of Rays treated 117.8 MG of water during the month of May and their non-accounted water totaled 2,238,200 gallons, what is your water loss percentage?

Answers for Chapter 9 Review

1. Venturi, Mag Meter
2. It under registers
3. Nutating disc and piston
4. Under registers
5. Venturi or Mag Meter
6. $\frac{117,800,000}{2,238,200} = 19\%$

Chapter 10: BASIC HYDRAULICS

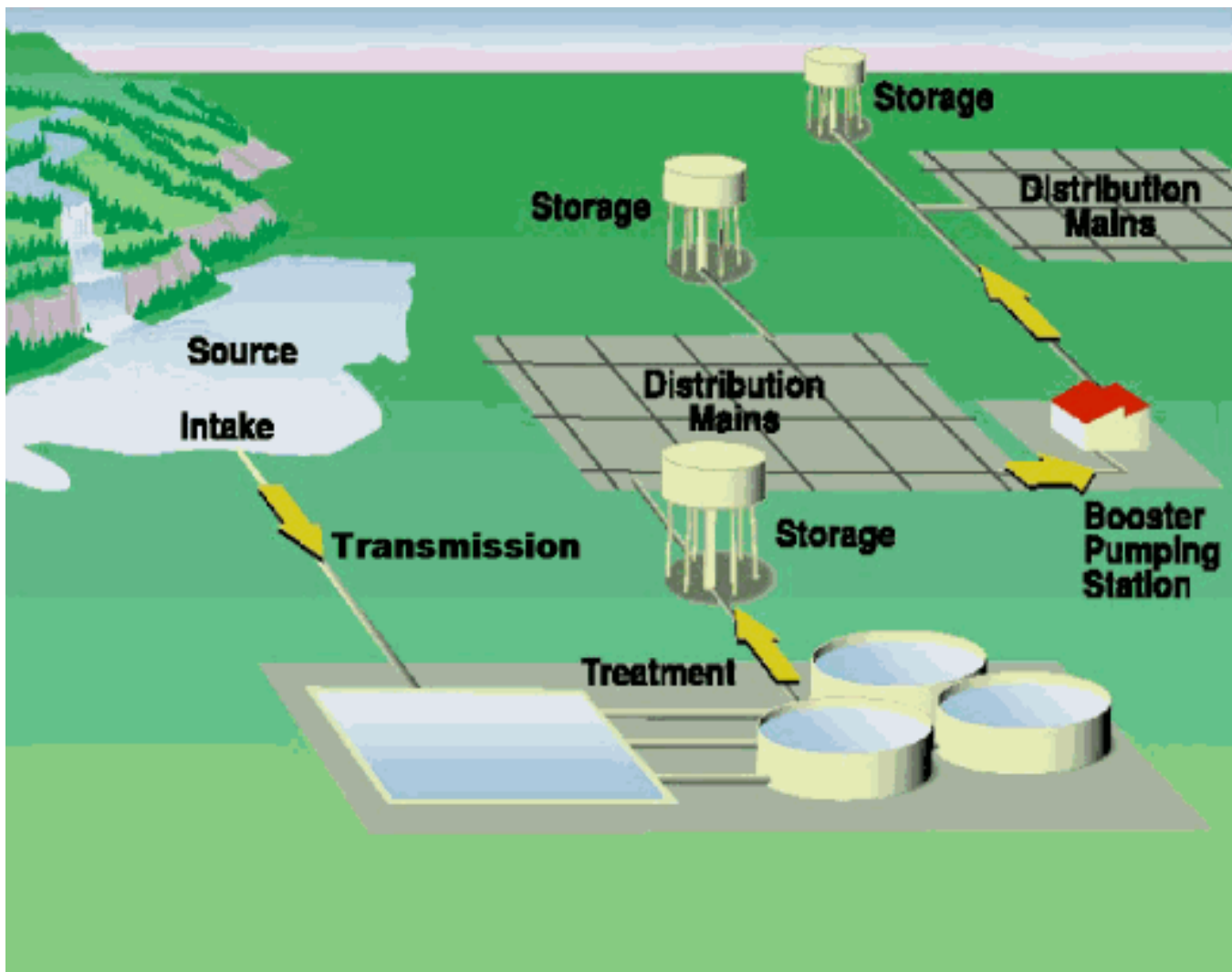
Chapter 10 Objectives

1. Identify major hydraulic concerns.
2. Demonstrate the ability to calculate flow (as functions of velocity and cross-sectional area) in gpm from different size fire hydrant openings and pipes.
3. Differentiate between several concepts related to basic hydraulics (e.g., psi vs. head, pressure vs. velocity, static vs. dynamic) and describe the major hydraulic concerns of a water distribution system.
4. Describe hydraulic grade line as it relates to pressure head and elevation head.
5. Explain and calculate "C" factor using Williams & Hazen's formula while describing equivalent flow, the rule of continuity, and pressure loss related to undersized piping.
6. Demonstrate the ability to determine appropriate line size and storage facility height (pressure and head).

The major hydraulic concerns of a water distribution system are

1. Will water flow?
2. In what direction?
3. With sufficient pressure?

In Kentucky, the minimum design pressure is 30 psi and the minimum working pressure is 20 psi. When pressure falls below 20 psi, public notification is mandatory by regulation. Accomplishing these hydraulic goals can be a difficult task, especially in terrain that includes substantial changes in elevation, which are present in our state.



Flow Rate Calculations

$$Q = A \times V$$

Q = FLOW expressed as ft³/sec (cubic feet/second—sometimes presented as cfs)

V = VELOCITY expressed as fps (feet per second)

A = CROSS-SECTIONAL AREA expressed as ft²

Flow is volume/time. Can be expressed as:

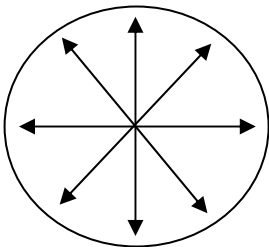
- gallons per minute (gpm)
- million gallons a day (MGD)

Velocity is speed or time/distance. Can be expressed as:

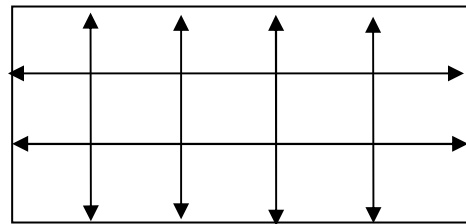
- feet per second (fps)
- yards per hour
- blocks per day

Area is a two-dimensional measurement. In this equation, we calculate the cross-sectional area. All we are concerned with in this equation relative to area is a two-dimensional measurement.

.785 X D (ft) X D (ft) for a circle



Length X Width for a rectangle



Anytime you have a measurement notated in inches, we must convert it to feet in order for us to calculate a correct answer. All that is needed to convert inches to feet is to divide by 12, which is how many inches are contained in a foot. Here are some examples:

8 inches is .666 feet (8 inches ÷ 12 inches)

2 inches is .166 feet (2 inches ÷ 12 inches)

42 inches is 3.5 feet (42 inches ÷ 12 inches)

Area Examples

1. What is the cross-sectional area of a 16 inch pipe?

The formula to obtain this is $.785 \times D \text{ (in feet)} \times D \text{ (in feet)}$

$$\text{Area} = .785 \times (16 \text{ inches} \div 12 \text{ inches}) \times (16 \text{ inches} \div 12 \text{ inches})$$

$$\text{Area} = .785 \times 1.33 \text{ ft} \times 1.33 \text{ ft}$$

$$\text{Area} = 1.4 \text{ ft}^2 \text{ (actually } 1.3885865 \text{ rounded up)}$$

So, the cross-sectional area for a 16 inch pipe is 1.4 ft²

2. Water is flowing through a 4 foot long by 2 feet wide by 3 feet deep trench at 1.9 feet per second. What is the flow?

$$Q = A \times V$$

$$Q \text{ (flow) expressed as ft}^3/\text{sec} = (4 \text{ ft} \times 2 \text{ ft}) \times 1.9 \text{ fps}$$

$$Q \text{ (flow) expressed as ft}^3/\text{sec} = 8 \text{ ft}^2 \times 1.9 \text{ fps}$$

$$A = 15.2 \text{ ft}^3 \text{ (cubic feet per second)}$$

Since we are figuring area (a two-dimensional measurement), we do not need to use the depth into the equation for the question above.

3. The flow through a 10 inch water line is 4.4 ft³/sec. What is the velocity of this water?

$$V = Q \div A$$

$$V \text{ (velocity) expressed as feet per second (fps)} = Q \text{ (flow) ft}^3/\text{sec} \div A \text{ (area) ft}^2$$

$$V = 4.4 \text{ ft}^3/\text{sec} \text{ (flow)} \div (.785 \times (10'' \div 12'') \times (10'' \div 12''))$$

$$V = 4.4 \text{ ft}^3/\text{sec} \div (.785 \times .83' \text{ (ft)} \times .83' \text{ (ft)})$$

$$V = 4.4 \text{ ft}^3/\text{sec} \div .54 \text{ ft}^2$$

$$V = 8.1 \text{ fps (feet per second)}$$

4. The flow through a 22 inch pipe is 3142 gpm. What is the velocity?

$$V = Q \div A$$

In order to satisfy the equation, we have to convert 3142 gpm into ft³/sec

The conversion sheet shows that 1 ft³/sec = 448.8 gpm

$$V = (3142 \text{ gpm} \div 448.8 \text{ gpm}) \div .785 \times (22" \div 12") \times (22" \div 12")$$

$$V = 7 \text{ ft}^3/\text{sec} \div (.785 \times 1.83' \times 1.83')$$

$$V = 7 \text{ ft}^3/\text{sec} \div 2.6 \text{ ft}^2$$

$$V = 2.69 \text{ or } 2.7 \text{ fps}$$

Practice Examples

1. The flow through the 36 inch main transmission line in the community of Getz is 3.6 MGD. What is the velocity of the water through this pipeline?
2. Water is traveling through a 4 inch pipe at 0.7 fps. What is the flow in this pipe expressed as gpm?
3. You are preparing to attach a line extension to a 12 inch pipe when the track hoe operator punctures the existing line. It took 3 minutes for water to travel through the trench that was 4 feet deep, 3 feet wide and 200 feet long. What is the velocity of this water?
4. Water is flowing through a completely filled 24 inch water main at 75 feet per minute. What is the flow in this pipeline?
5. In order for your new pump to work efficiently it needs a minimum of 700 gpm to be pumped through it. The line attached to the pump is 18 inches, and the water is flowing at 1.1 fps. How many gallons per minute are moving through the new pump?

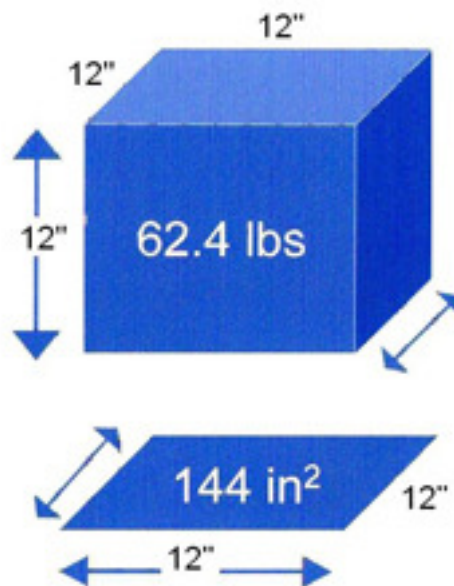
Solutions

- 1) $V = Q \div A$
 $V = (3.6 \text{ MGD} \times 1.55 \text{ ft}^3/\text{sec}) \div .785 \times (36'' \div 12'') \times (36'' \div 12'')$
 $V = 5.58 \text{ ft}^3 \div (.785 \times 3' \times 3')$
 $V = 5.58 \text{ ft}^3/\text{sec} \div 7.07 \text{ ft}^2$
 $V = 0.8$ (actually 0.789) fps
- 2) $Q = A \times V$
 $Q = .785 \times (4'' \div 12'') \times (4'' \div 12'') \times 0.7 \text{ fps}$
 $Q = (.785 \times .33' \times .33') \times 0.7 \text{ fps}$
 $Q = 0.085 \text{ ft}^2 \times 0.7 \text{ fps}$
 $Q = .0595 \text{ ft}^3/\text{sec}$ How many gallons per minute would that be? 26.7 gpm
- 3) If it takes 3 minutes (180 seconds) for the water to travel 200 feet, then we could say that the water was traveling at a rate of 1.1 fps (200 feet \div 180 seconds).
- 4) $Q = A \times V$
 $Q = (.785 \times 2' \times 2') \times 1.25 \text{ fps}$ which has been converted from 75 feet per minute
 $Q = 3.14 \text{ ft}^2 \times 1.25 \text{ fps}$
 $Q = 3.9 \text{ ft}^3/\text{sec}$
- 5) $Q = A \times V$
 $Q = (.785 \times (18'' \div 12'') \times (18'' \div 12'')) \times 1.1 \text{ fps}$
 $Q = .785 \times 1.5' \times 1.5' \times 1.1 \text{ fps}$
 $Q = 1.94 \text{ ft}^3/\text{sec}$
 $Q = \text{Converting to gpm} \quad 1.94 \text{ ft}^3/\text{sec} \times 448.8 \text{ gpm} = 870.6 \text{ gpm}$
 $Q = 870.6 \text{ gpm}$

Hydraulics

Hydraulics is the branch of engineering that focuses on the practical problems of collecting, storing, measuring, transporting, controlling and using water or other liquids. Liquids in motion produce forces and pressure whenever the velocity, flow direction or elevation changes. Knowing pipe pressure and flow along certain points in our distribution systems can help determine the proper pipe size and sufficient capacity. These parameters can also help determine what pipe material would work best in certain situations. In addition, having an understanding of hydraulics can help systems determine if pressure reducers or pumps are necessary to transport water in an efficient manner. Also, basic understandings of hydraulics can save money, improve water quality, ensure regulatory compliance and help determine the cost and viability of distribution system improvements.

Some of the operational concerns that confront the successful delivery of our water are elevational changes in the topography (elevational head), energy lost through the transmission of the water through the pipelines (frictional head), (velocity head) and pressure requirements that have to be maintained to ensure water quality and regulatory compliance. Understanding that water has a unit weight is important. Water has a unit weight of 8.34 lbs per gallon or 62.4 lbs/ft³.



1 cubic foot of water (1 ft X 1 ft X 1 ft) weighs 62.4 lbs.

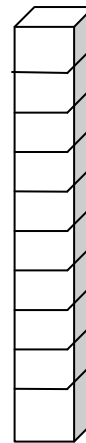
If it rests on 1 square foot or 144 square inches, then it exerts 0.433 **p**ounds per **s**quare **i**nch. (psi). Pressure is derived by the formula as Force/Area.

$$P = \frac{F}{A} = \frac{62.4 \text{ lbs.}}{144 \text{ sq. in.}} = 0.433 \text{ PSI}$$

Therefore, one square foot of water exerts 0.433 psi.

Now consider 10 feet of water. 10 cubic feet of water weighs 624 lbs if it rests on 1 square foot or 144 square inches.

$$P = \frac{F}{A} = \frac{62.4 \text{ lbs./ft}^3 \times 10 \text{ ft}^3}{144 \text{ sq. in.}} = 4.33 \text{ PSI}$$

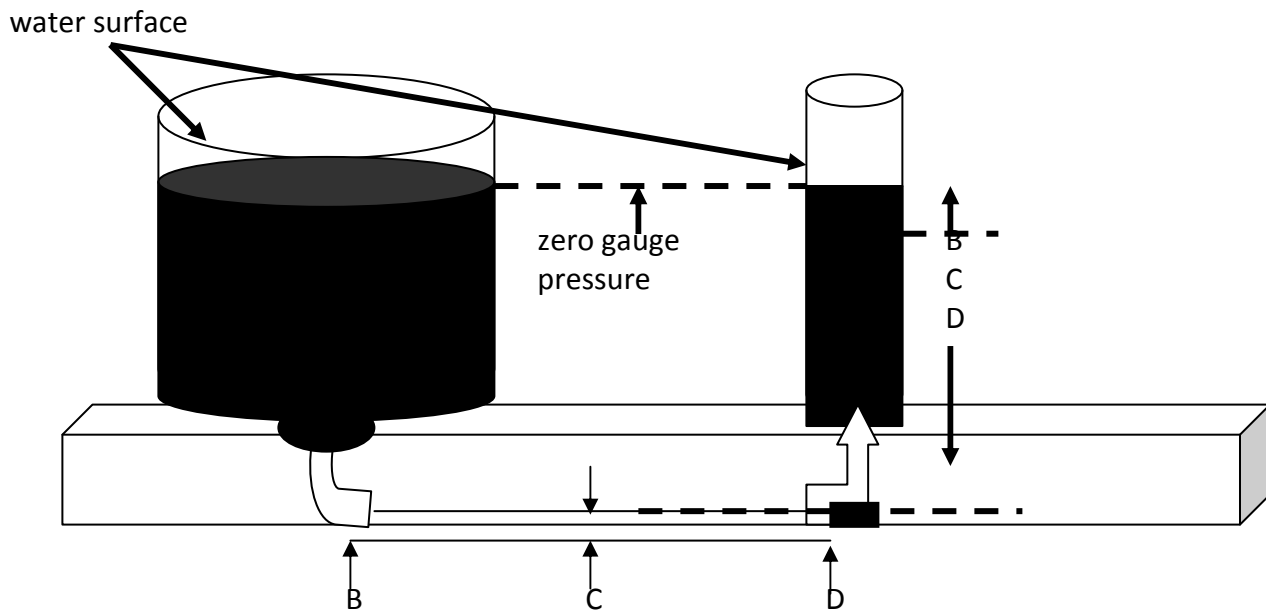


To determine pressure (psi) in relation to a column of water, use “H” (height) in feet of water. If a standpipe held 80 feet of water and the pressure gauge was at the base of the tank, then

$$\begin{aligned} \text{Pressure (psi)} &= \text{Height of water column} \times 0.433 \text{ psi} \\ \text{Pressure} &= 80 \text{ ft of water} \times 0.433 \text{ psi} \\ \text{Pressure} &= 34.64 \text{ psi} \end{aligned}$$

Water exerts force and pressure when it is confined in any type of structure. In our business, the types of structures we commonly use are pipes and tanks.

There is a difference between force and pressure. Pressure is defined as force per unit area, commonly expressed as pounds per square inch (lb/in² or psi).



The pressure at point **A** equals the pressure at point **E**, since these points are the same depth in the water. Likewise, the hydrostatic pressure at points **B**, **C**, and **D** are equivalent.

Hydrostatic Pressure

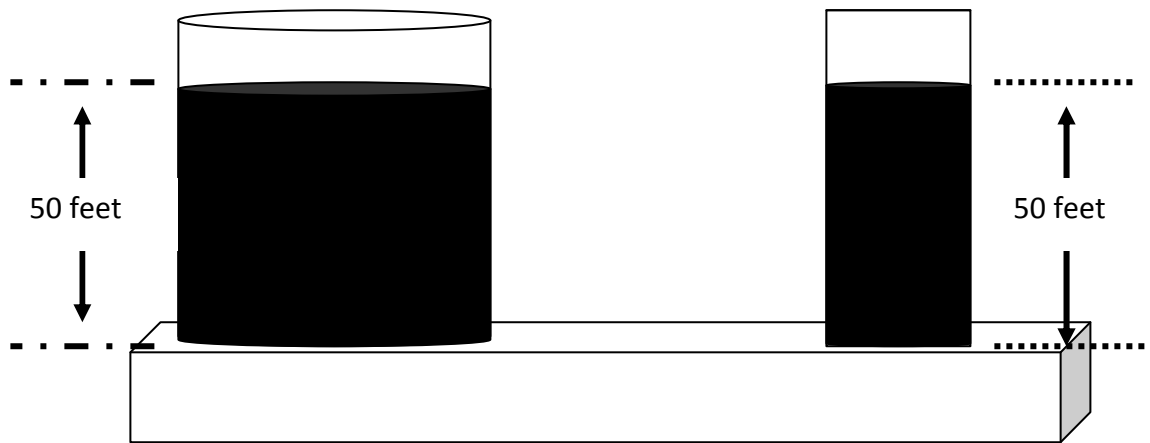
The pressure water exerts is called hydrostatic pressure.

These principles always apply to hydrostatic pressure.

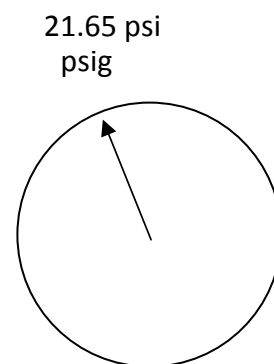
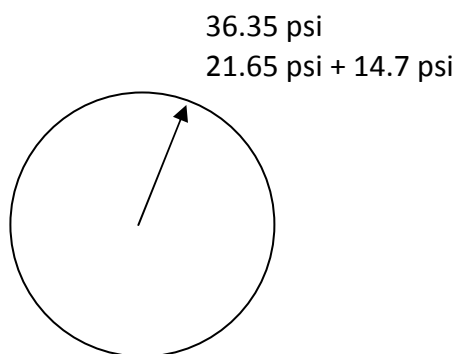
1. Pressure depends only on the depth of water above the point in question (not on the water surface area).
2. Pressure increases in direct proportion to the depth of water.
3. Pressure in a continuous volume of water is the same at all points that are at the same depth or elevation.
4. Pressure at any point in the water acts in all directions at the same magnitude.

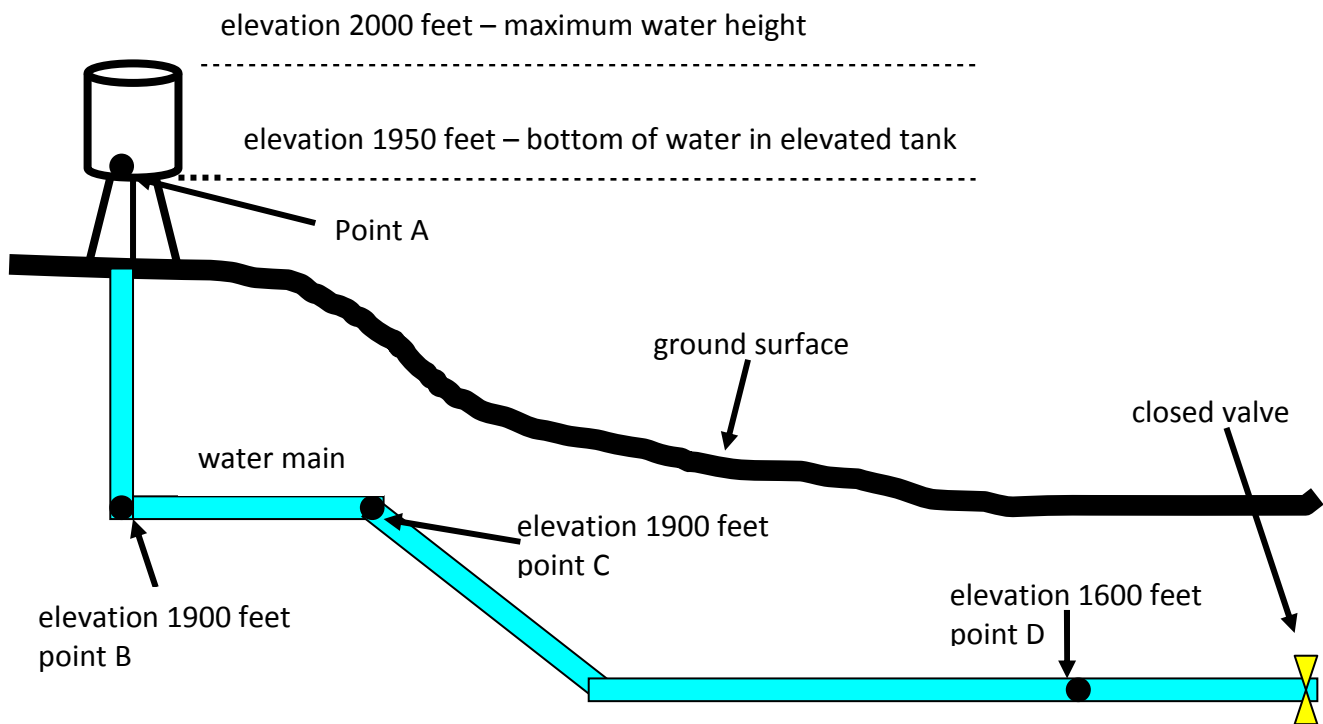
Pressure also exists at the surface of the water because of the weight of the air above.

This pressure is called atmospheric or barometric pressure. Atmospheric pressure at sea level is 14.7 psi. In practical applications for the distribution operator, atmospheric pressure is NOT used for hydraulic calculations and is considered to be a zero reference or starting point. The application that uses atmospheric pressure is called absolute pressure, as opposed to gauge pressure.



Here we have two storage tanks side by side with different volumes. Even though these tanks have differing volumes, the pressure is the same because the column of water in each tank is the same. Hydrostatic pressure depends solely on the height of water above given points and NOT on the volume of water. In this above illustration, the pressure at the base of each tank is 21.65 psi. (50 feet X .433 psi) If whatever pressure gauge we used included atmospheric pressure (psia), they would NOT read the same.



**Example 1**

A small water system has an elevated water storage tank and connecting ductile iron pipeline as shown in the illustration above. Calculate the hydrostatic pressure at points A, B, C, D.

- A. $2000 \text{ feet} - 1950 \text{ feet} = 50 \text{ feet}$ elevational difference. $50 \text{ ft} \times 0.433 \text{ psi} = 21.65 \text{ psi}$
- B. The total depth of the water at point B is $2000 \text{ feet} - 1900 \text{ feet} = 100 \text{ ft}$
 $100 \text{ feet} \times 0.433 \text{ psi} = 43 \text{ psi or lbs/in}^2$
- C. The pressure at point C is the same as point B because there is no elevational difference.
- D. The total depth of the water above point D is $2000 \text{ feet} - 1600 \text{ feet} = 400 \text{ feet}$
 $400 \text{ feet} \times 0.433 \text{ psi} = 173.2 \text{ psi or lbs/in}^2$

Example 2

Another issue that is becoming more commonplace is providing a water service for a customer with a house at the top of a hill. In this example, a 6-inch main provides the line to be tapped for the service line. The pressure at the water meter is 51 psi. The house sits on a hill and through topographic maps it is determined the meter is at an elevation of 325 feet and the house is at an elevation of 405 feet.

Calculate the pressure drop from the meter to the house.

$$325 \text{ feet} - 405 \text{ feet} = - 80 \text{ feet}$$

$$- 80 \text{ feet} \times 0.433 = - 34.64 \text{ psi}$$

$$51 \text{ psi} - 34.64 \text{ psi} = 15.36 \text{ psi}$$

Let's take this a step further. This house has a bathroom on the second floor. Estimate another 10 feet of elevation to the second floor. So, there is an additional 10 feet of elevation, which will result in an additional 4.33 psi pressure drop up to the second floor bathroom.

If we add the additional 4.33 psi to the original 34.64 psi, we now have a pressure drop of 38.97 or 39 psi.

Our starting measurement of 51 psi – 39 psi (pressure drop) means that the water in the second floor bathroom would be measured at approximately 12 psi.

In the above example, it would most likely be wise to inform the customer that the water pressure in their house will be low. ALSO, explain to the customer that THEY can INSTALL and MAINTAIN a small booster pump to increase pressure. This is done at THEIR expense, not the utility's expense.

Example 3

The inverse of this situation could very well arise. A customer has a house in a valley with an elevation 156 feet below the connecting point at the transmission main. The pressure at the top of the hill is 94 psi.

What would the pressure be at the service connection entrance to the house?

$$P \text{ (pressure)} = 0.433 \times \text{Height}$$

$$P = 0.433 \times 156 \text{ ft}$$

$$P = 67.55 \text{ psi}$$

Pressure reading at the top of the hill 94 psi + additional pressure gained by elevation gained equals 67.55 psi.

$$94 \text{ psi} + 67.55 \text{ psi} = 161.55 \text{ psi at the service entrance to the house.}$$

It is sometimes more convenient to express pressure in terms of the height of a column of water in feet. This is called pressure or elevational head and is the actual or equivalent height of water above the point in question.

For example, when evaluating a water distribution system to determine how high the pressure will push the water up a hill. By turning the equation around, we get:
 $H = P \div 0.433$ or by using your conversion sheet, which dictates that every 1 psi is equivalent to 2.31 ft of head. (The 2.31 comes from dividing 1 by 0.433, which gives you 2.309 or 2.31.)

So if we had to overcome a rise in elevation of 65 feet and our starting pressure at the main is 65 psi, then we could take $65 \text{ psi} \div 0.433 \text{ psi}$, which equals 150.1 feet.

OR

Multiply the pressure in psi (65) X 2.31 ft of head, which equals 150.1 ft of head.

So theoretically, the water in the main could rise to a height of 150.1 feet, which is equivalent to 65 psi.

If we had to overcome a 65 foot rise in elevation and our starting pressure was 65 psi, we could determine what the pressure would be at the crest or apex of the hill.

Starting pressure of 65 psi minus the elevational difference in feet (65 feet) X 0.433 psi means that we would lose 28.1 (65 psi X 0.433 psi) psi, overcoming the 65-foot hill. If we start at the bottom of the hill with 65 psi and subtract the 28.1 psi, we would lose that energy overcoming the elevational difference and the theoretical pressure at the top of the hill would be 36.9 or 37 psi (65 psi – 28.1 psi = 36.9 or 37 psi).

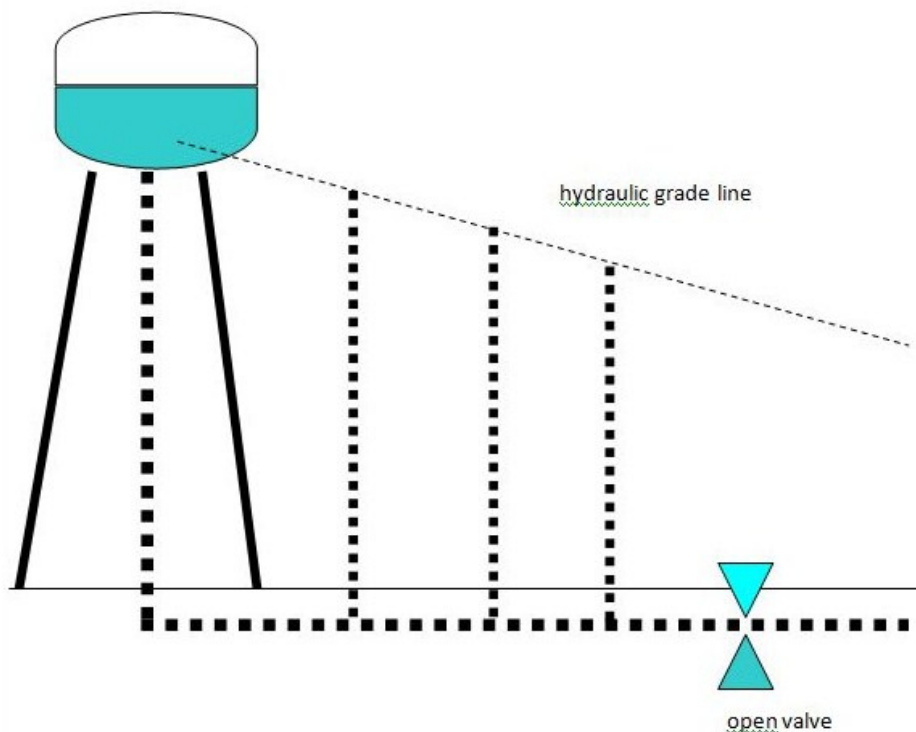
Hydraulic Grade Line

The hydraulic grade line is an imaginary line that displays the sum of the pressure head and the elevational head. The hydraulic head represents the height to which a water column would rise in open standpipes.

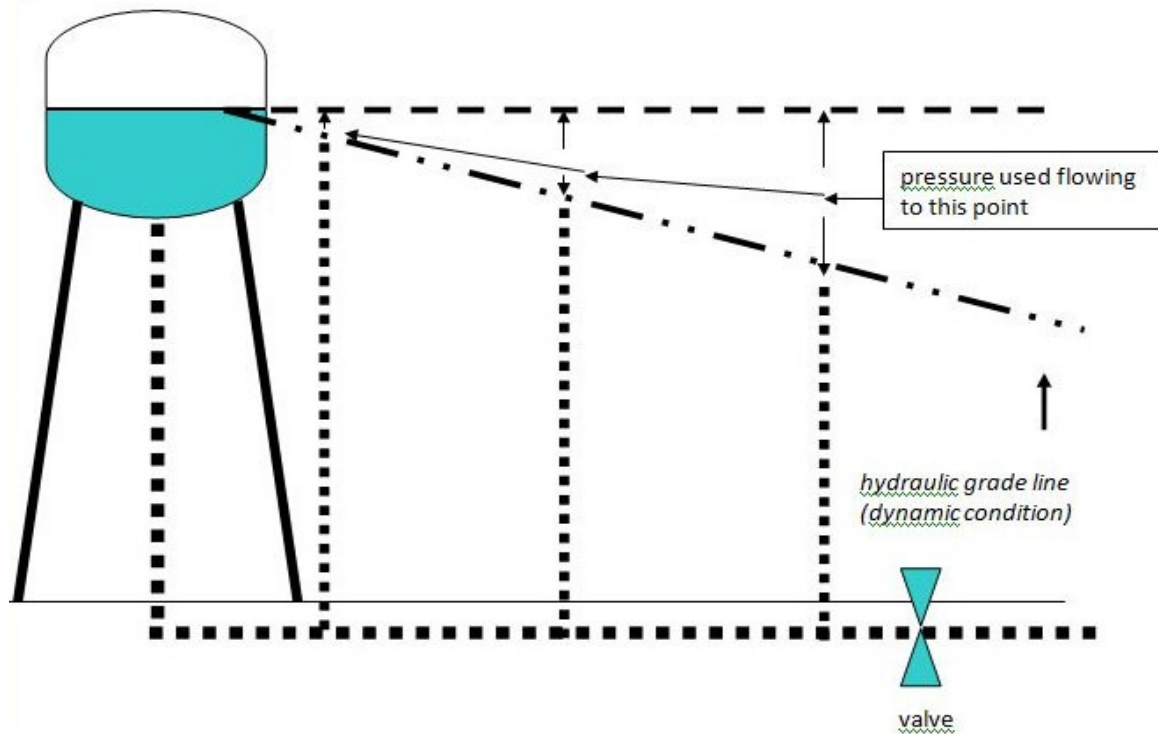
- A **static** condition means the water in the pipeline is NOT moving, but is stationary.
- A **dynamic** condition indicates that the water is moving and impacting all the entities that are affected by the movement of water.

Water will flow from the area of higher pressure to areas of lower pressure. This is the principle of how our distribution system works. In the static condition, water is not moving. In the dynamic or kinetic condition, water is moving through the pipelines. Different factors impact the water whether it is in a static or dynamic condition.

Hydraulic Gradeline of Water in Motion (Dynamic or Kinetic Condition)



Hydraulic Grade Line (Static Condition)



C Factor or Values

The “C” Factor or Value denotes the interior smoothness of a pipe. These factors are given a numerical value in order to determine the amount of energy lost in a pipeline due to friction. The energy losses are usually expressed as loss in feet of head, loss in psi, or loss in the velocity of the water in the pipe. The higher the “C Factor,” the higher the number associated with it.

Common Friction Factor Values Used for Design Purposes

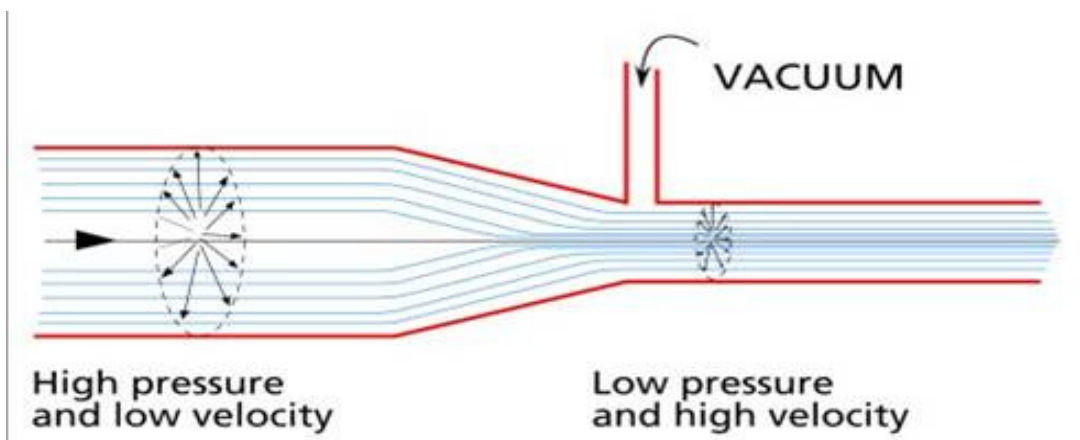
Asbestos Cement	140
Cast Iron (new)	130
Cast Iron (40 yr old)	62
Copper Tube	130
PVC Pipe	140
Steel Pipe	120
Ductile Iron Pipe (lined)	140

WHICH PIPE WOULD HAVE THE HIGHEST C FACTOR?

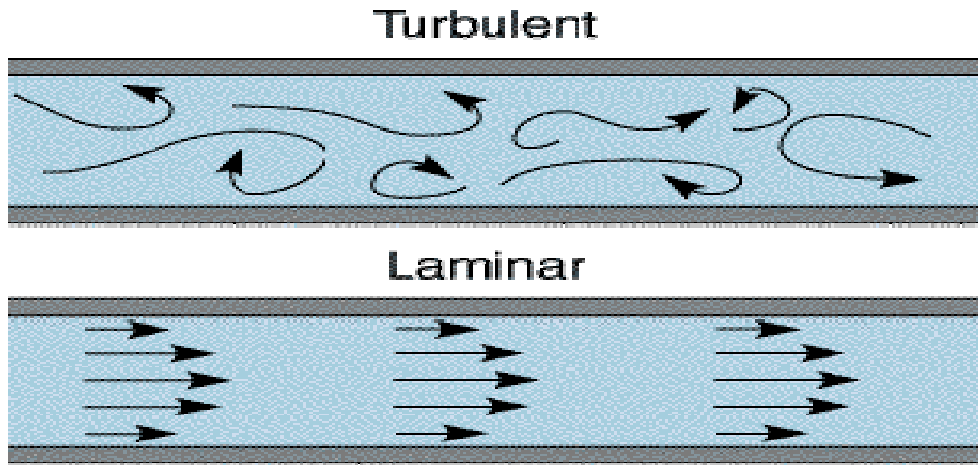


A common misconception is that pressure increases when the pipe size decreases. Decreasing pipe diameter will increase the velocity, but will reduce the pressure – meaning that as the pipe size decreases, the water merely travels faster. The friction created by a liquid flowing in a pipe depends upon its **velocity** and the **diameter of the pipe**.

- Friction increases as velocity increases.
- Velocity and pressure have an inverse relationship. The faster the velocity, the more energy is lost due to friction between the water and the pipe wall.
- When flow is restricted, velocity increases.



When the interior of a pipe has substantial tuberculation, flow in the pipe turns from laminar to turbulent flow. Turbulent flow requires more energy to deliver the same volume of flow.



When connections or additions to pipelines are needed, some calculations of all the factors that will come into play must be accounted for if we want to maintain the water quality as well as deliver the necessary volume of water at sufficient pressures to our customers. Frequently just throwing a pipe in the ground to deliver water without factoring in all the variables will not accomplish what we need and will most likely make conditions worse for anyone hooked to this new line.

Over the years, engineers have developed a number of ways to determine pressure loss relative to flow demand and pipe size. If we install undersized lines, we run the risk of:

- a. not delivering enough pressure to the customer
- b. not delivering enough volume to the customer
- c. creating backflow incidents
- d. degrading water quality

Example

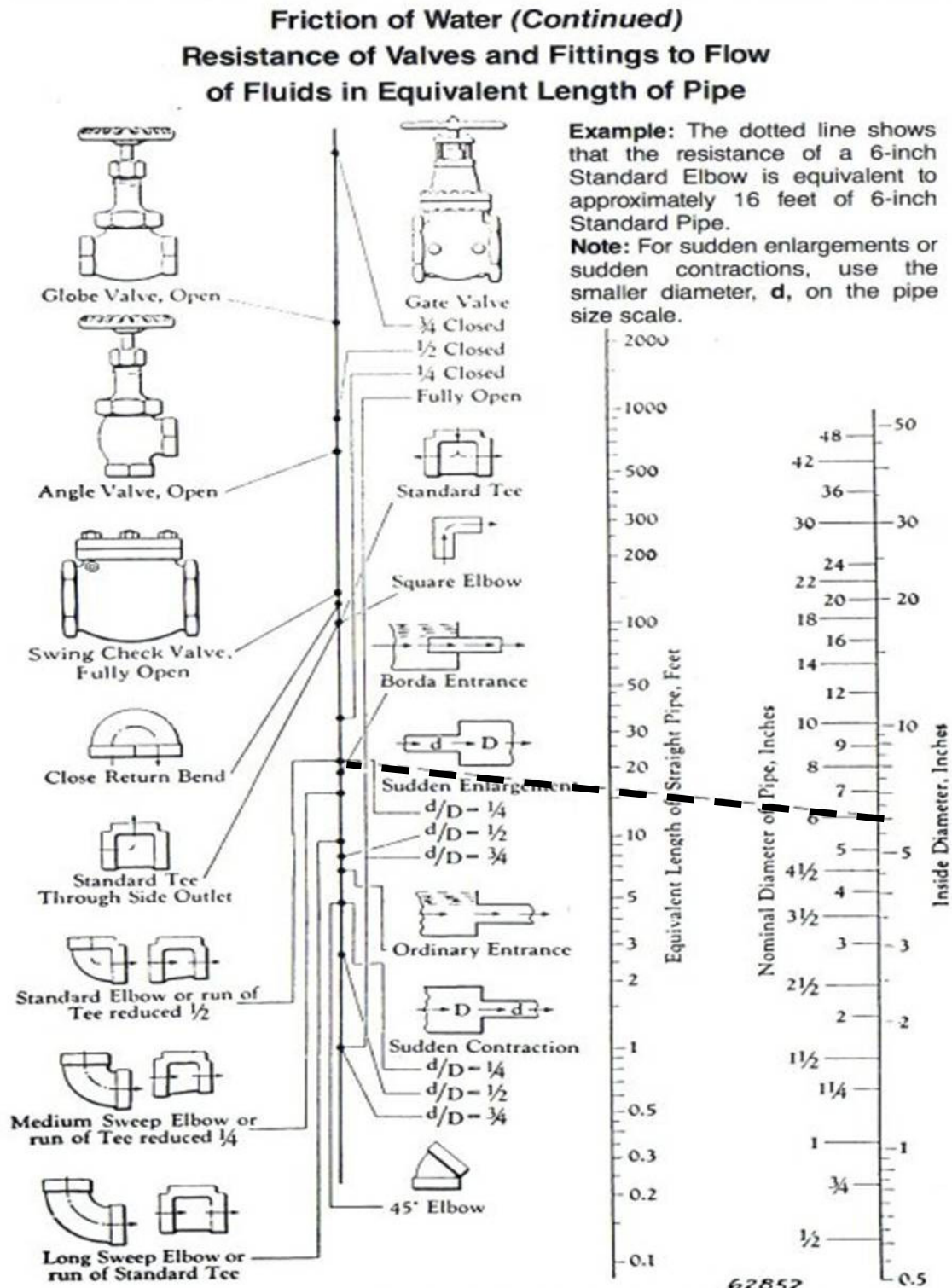
You are going to install a new 6" line to supply water to a new subdivision. The distance from the line you are tapping into to supply this new line all the way to the last house in this subdivision is 2000 feet. There is also an elevational incline of 75 feet to overcome when delivering the water. Pressure at the spot that you are tying the new line to has been measured at 100 psi, and the flow rate you want to achieve in the new pipe is 450 gpm. You are going to valve off each intersection so there will be 6 gate valves installed on this line.

Looking at the chart above, we can determine that for each 100 feet of pipe, we will be adding the equivalent of 2.74 ft of head at 450 gpm. We are installing 2000 feet of pipe, so that would be 20, 100 foot sections. We also need to push the water up 75 feet of incline. As well, the six gate valves will be equivalent to 16 feet of head per valve. Now we can factor in all of these parameters and determine if the six inch line is large enough to accomplish what we need.

20 pipe sections X 2.74 ft of head plus 75 feet of elevational head plus 6 valves @ 16 feet apiece.

20 pipe sections X 2.74 ft of head + 75 ft of head + 96 feet of head = 225.8 ft/head.

Starting pressure 100 psi minus (225.8 ft of head X 0.433 psi) = 2.23 psi. You're in trouble!



From Crane Co. Technical Paper No. 409. Data based on the above chart are satisfactory for most applications; for more detailed data and information refer to pages 3-110 to page 3-120 which are based on Crane Co. Technical Paper No. 410.

3-121

Transients

Five mechanisms/pathways can lead to water quality failures namely intrusion, regrowth, breakthrough, internal corrosion / leaching and permeation.

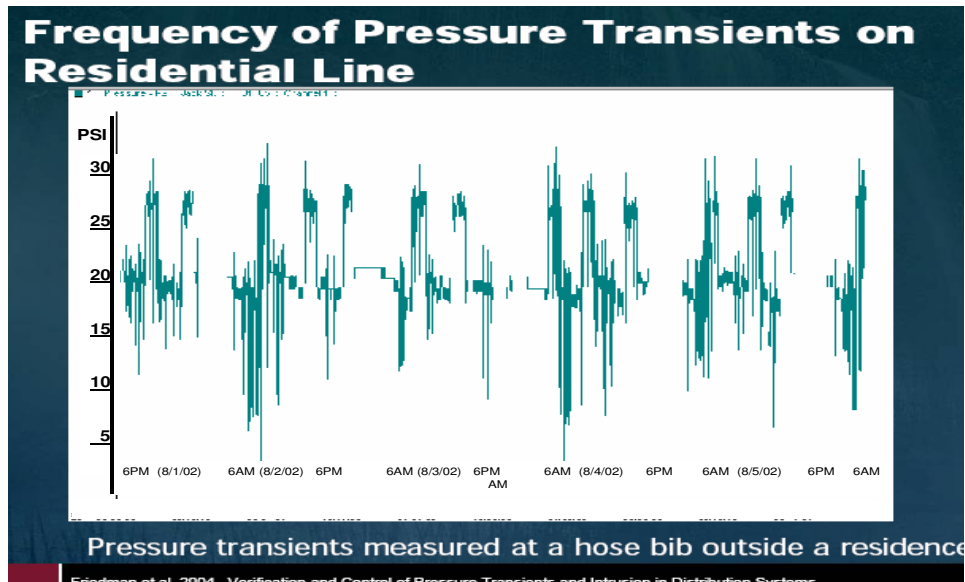
Intrusion, a primary mechanism of water quality failures in distribution networks, has accounted for approximately 15% of the total documented cases of waterborne illnesses in the United States in the last 30 years. Intrusion through water mains may occur during maintenance and repair events, through broken pipes and gaskets in the presence of contaminated soil and/or cross-connections. The potential of contamination through backflow or through leaky pipes increases whenever the water pressure in a pipe is very low or negative. This can occur when the pipe is de-pressurized for repair or when it is used to extinguish fire or during episodes of transient pressures.

Intrusion of contaminants into water distribution networks requires the simultaneous presence of three elements; contamination source, pathway and driving force. Intrusion of contaminants (hitherto referred to simply as “intrusion”) into the water distribution network can occur through pipes and storage tanks (animals, dust-carrying bacteria, infiltration). Intrusion through deteriorated water mains can occur during maintenance and repair events, through broken pipes and gaskets, and cross-connections. Kirmeyer et al. (2001) ranked pathogen (contaminant) entry routes into the distribution network based on responses from an expert panel, the members of which were instructed to identify and rank the importance of routes of entry. Intrusion was rated mostly “high”. In addition to pathogens, intrusion can also introduce into the pipe chemicals, such as pesticides, herbicides, hydrocarbons (gasoline spills) as well as physical contaminants, such as plant debris and soil particles. Intrusion into water mains requires the simultaneous presence of three elements, a pathway, driving force (negative pressure differential between the pipe and its environment) and a contamination source. Contamination sources can be either chemical (pesticides, herbicides, petroleum products, fertilizers, solvents, detergents, pharmaceuticals, etc.) or microbiological (microbes, viruses, bacteria). Karim et al. (2003) reported concentrations of total coliform, fecal coliform, clostridium, bacillus, and viruses in soil and water samples taken around the water mains

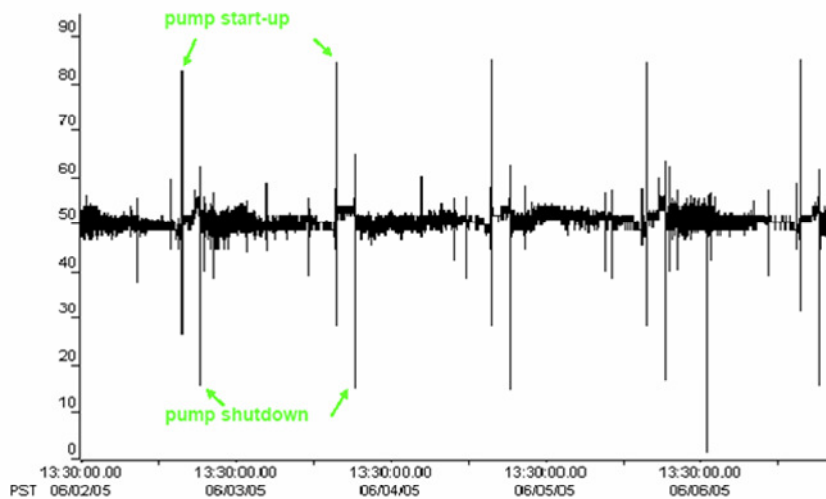
The deterioration of pipe structural integrity can have a multi-faceted impact on water quality, especially in the domain of contaminant intrusion. Frequent pipe breaks increase the possibility of intrusion through the compromised sections in several ways. During repairs, intrusion can occur if flushing and local disinfection procedures are not appropriately followed. Furthermore, pipes are de-pressurized in the vicinity of a break during repair. This

low pressure increases the potential of contaminant intrusion through unprotected cross connections. If the pipe has holes then de-pressurization will increase the likelihood of contaminant intrusion, which can be especially detrimental if the surrounding soil is contaminated by leaky sewers nearby, chemical spills, herbicides, pesticides, etc. Total and fecal coliforms were found in more than 60% and 40% of the samples, respectively. Enteroviruses, Norwalk and Hepatitis A viruses were also found around the pipe giving a strong indication of human and animal sources of contamination. Karim et al. (2003) also reported the range of concentrations for various organisms found in soil samples collected in the vicinity of the water mains.

A water distribution network can never be completely water tight due to the existence of pipe cracks, holes, faulty gaskets and/or faulty appurtenances, which can serve as intrusion pathways. The driving force required for intrusion is usually a pressure differential. Therefore, it is improbable that intrusion will occur as long as the water pressure inside the network is greater than the pressure outside although movement of microbial or viral contaminants against the pressure gradient is possible. Pressure differential can occur during maintenance activities, such as during break repairs, flushing, etc., when parts of the distribution network are de-pressurized. Sources of contaminants include sewage water ex-filtrated from adjacent broken sewers, contaminated groundwater/soil and backflow through unprotected cross-connections. In addition to pressure differentials arising due to de-pressurization of pipes, extreme transient pressures can also cause pressure differentials. Extreme transient pressures in a water supply system can occur as a result of power failure in a pumping station, fast closure of valves, fire flows, pipe rupture, etc. These transients can cause negative pressures in pipes, which sometimes may be exacerbated by peculiar topographical conditions. These negative pressures may provide a driving force for contaminants to intrude through compromised pipe walls and joint gaskets. Extreme transient pressures are more likely to occur in long transmission mains than in an urban distribution network in which users' faucets effectively serve as widely distributed pressure relief valves. An exception may be during fire flows or in the vicinity of a wet industrial facility. The volume of the inflow of the contaminated solute is typically quite small (less than 1% of the flow in the pipe) since the duration of transient pressures is quite short (Kirmeyer et al. 2001).



Distribution System Pressure @ 1 per sec



Equivalent Flow Rate Calculation

The Williams and Hazen formula can also be used for C-Values other than 100 by first converting the **actual flow rate** to an **equivalent flow rate** to obtain a C-Factor of 100.

The equation used to find the equivalent flow rate is:

$$\text{Equivalent Flow Rate} = \frac{(\text{Actual Flow Rate}) (100)}{\text{C-Value}}$$

Example

A 10 inch pipeline is 3000 feet long and has a smoothness coefficient (C-Factor) of 130. Water is moving through the pipe at a flow rate of 3900 gpm. Using the Williams and Hazen formula, you can determine the friction head loss.

Since the C-Factor is NOT 100, first use the equation shown above to find the equivalent flow rate.

$$\text{Equivalent Flow Rate} = \frac{(\text{Actual Flow Rate}) (100)}{\text{C-Value}}$$

$$\text{Equivalent Flow Rate} = \frac{(3900 \text{ gpm}) (100)}{130}$$

$$\text{Equivalent Flow Rate} = 3000 \text{ gpm}$$

Now, use the Equivalent Flow Rate to find the friction loss from the Williams and Hazen formula. Find a flow rate of 3000 gpm and then move to the 10 inch column. At that point, the table shows a friction loss of 7.62 feet of head per 100 feet of pipe. The pipeline in the example is 3000 feet long, so now calculate how many 100 foot sections of pipe are contained in 3000 feet. Now multiply this number by the friction loss per 100 foot section of pipe.

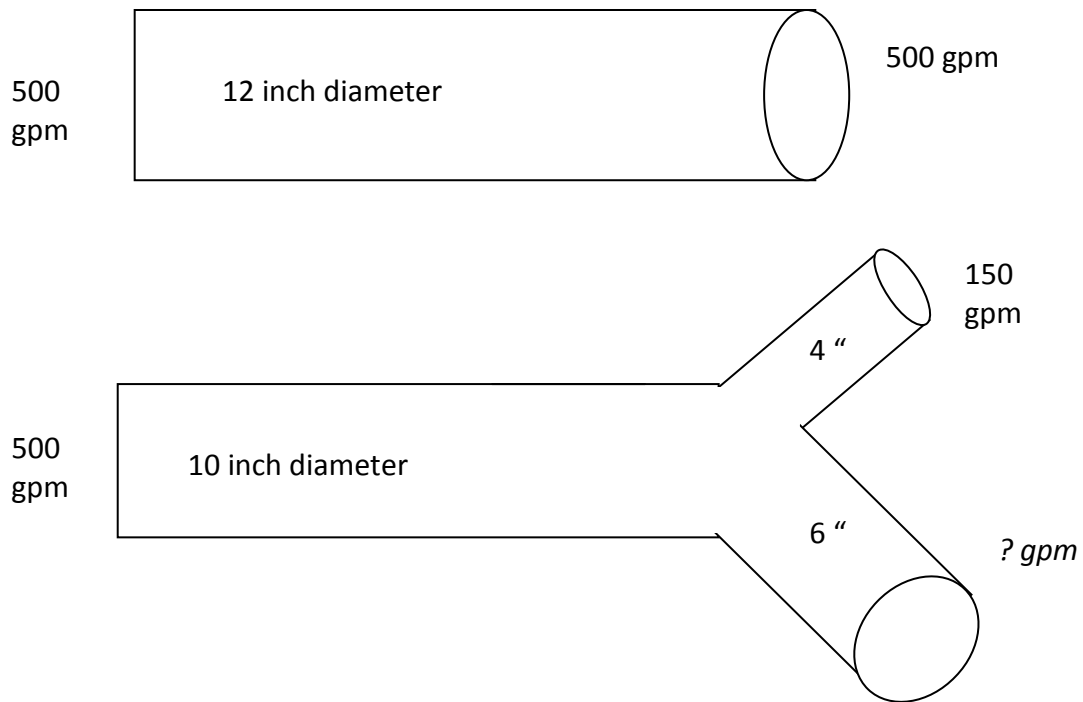
$$(30) (7.62) = 228.6 \text{ ft friction loss.}$$

If we wanted to find out how much pressure would be lost through this pipe (assuming it was level and straight with no fittings), we would multiply this figure by 0.433 psi (1 ft of head = .433 psi).

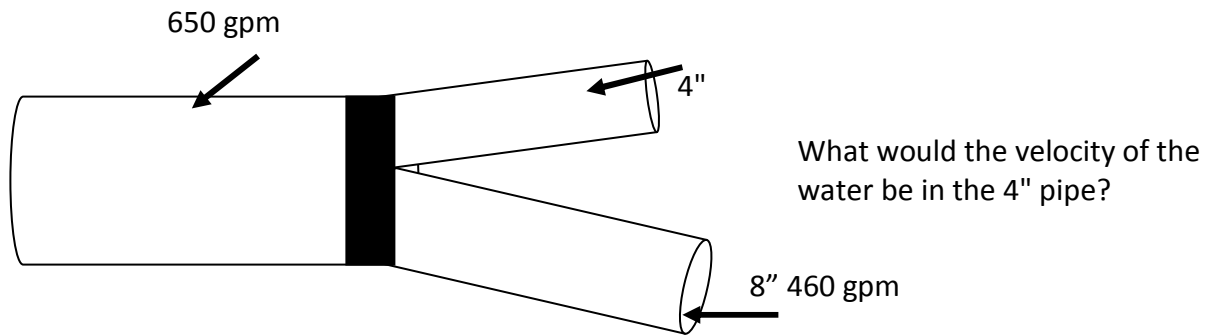
$$228.6 \text{ ft friction loss} \times .433 \text{ psi} = 98.98 \text{ psi}$$

By using the Williams and Hazen formula, we can determine that 98.98 psi would be lost in this 3000 foot pipeline.

Rule of Continuity



What goes in has to come out! What would the flow in the 6 inch pipe be? If we have 500 gpm coming into the 10" pipe and 150 gpm flowing through the 4" pipe, that leaves us with 350 gpm flowing through the 6" pipe.



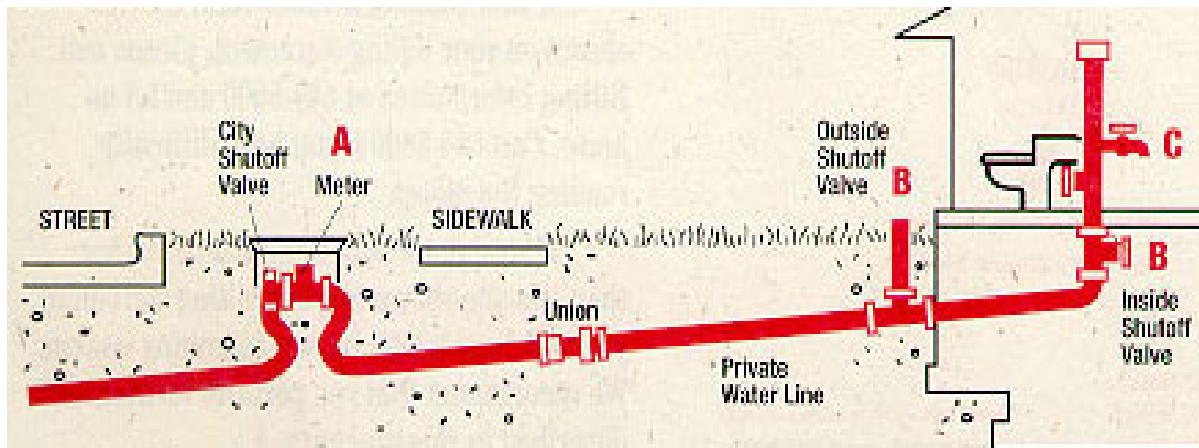
$$V = Q \div A$$

$$V = 190 \text{ gpm } (650 \text{ gpm} - 460 \text{ gpm}) \div (.785 \times .33' \times .33')$$

$$V = .42 \text{ ft}^3/\text{sec } (190 \text{ gpm} \div 448.8 \text{ gpm}) \div .085 \text{ ft}^2$$

$$V = .42 \text{ ft}^3/\text{sec} \div .085 \text{ ft}^2$$

$$V = 4.9 \text{ fps}$$



Review questions provided at the end of Chapter 11.

Chapter 11: CENTRIFUGAL PUMPS

Chapter 11 Objectives

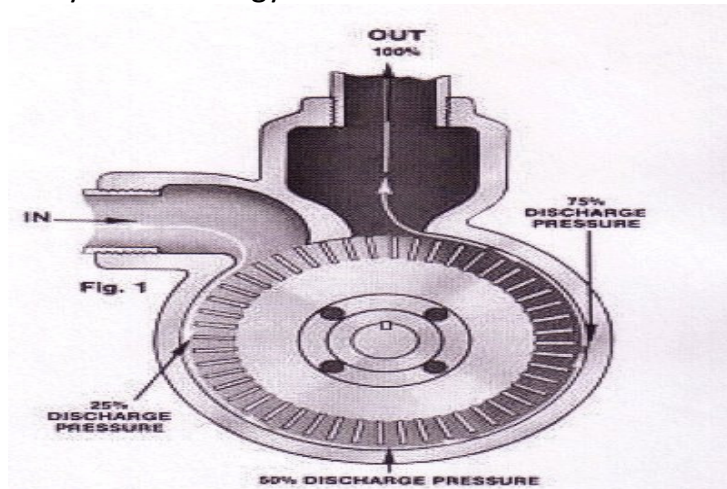
1. Define NSHA as opposed to NSHR.
2. Attain the ability to read a pump curve.
3. Discern how small differences in impeller type and size affects flow rate and pump characteristics.
4. Determine what differences occur when centrifugal pumps are used in parallel and in series.
5. Discover why centrifugal pumps are the pumps of choice in distribution systems.

PUMPS

APPLICATION	FUNCTION	PUMP TYPE
Low Service	To lift water from the source to the treatment process or from storage to filter backwashing system.	Centrifugal
High Service	To discharge water under pressure to the distribution system.	Centrifugal
Booster	To increase pressure in the distribution system.	Centrifugal
Well	To lift water from shallow or deep wells and discharge it to the treatment plant, storage facility, or distribution system.	Centrifugal Or Jet
Chemical Feed	To add chemical solutions at the desired dosages for treatment or disinfection processes	Positive Displacement
Sampling	To pump water from sampling points to the lab or to automatic analyzers.	Positive Displacement Or Centrifugal
Sludge	To pump sludge from the sedimentation facilities for further treatment or disposal.	Positive Displacement Or Centrifugal

What is a centrifugal pump?

It is a machine that imparts energy to a fluid, causing it to flow, rise to a higher level, or both. It actually uses acceleration (not centrifugal force) to transform mechanical (rotational) energy into hydraulic energy.

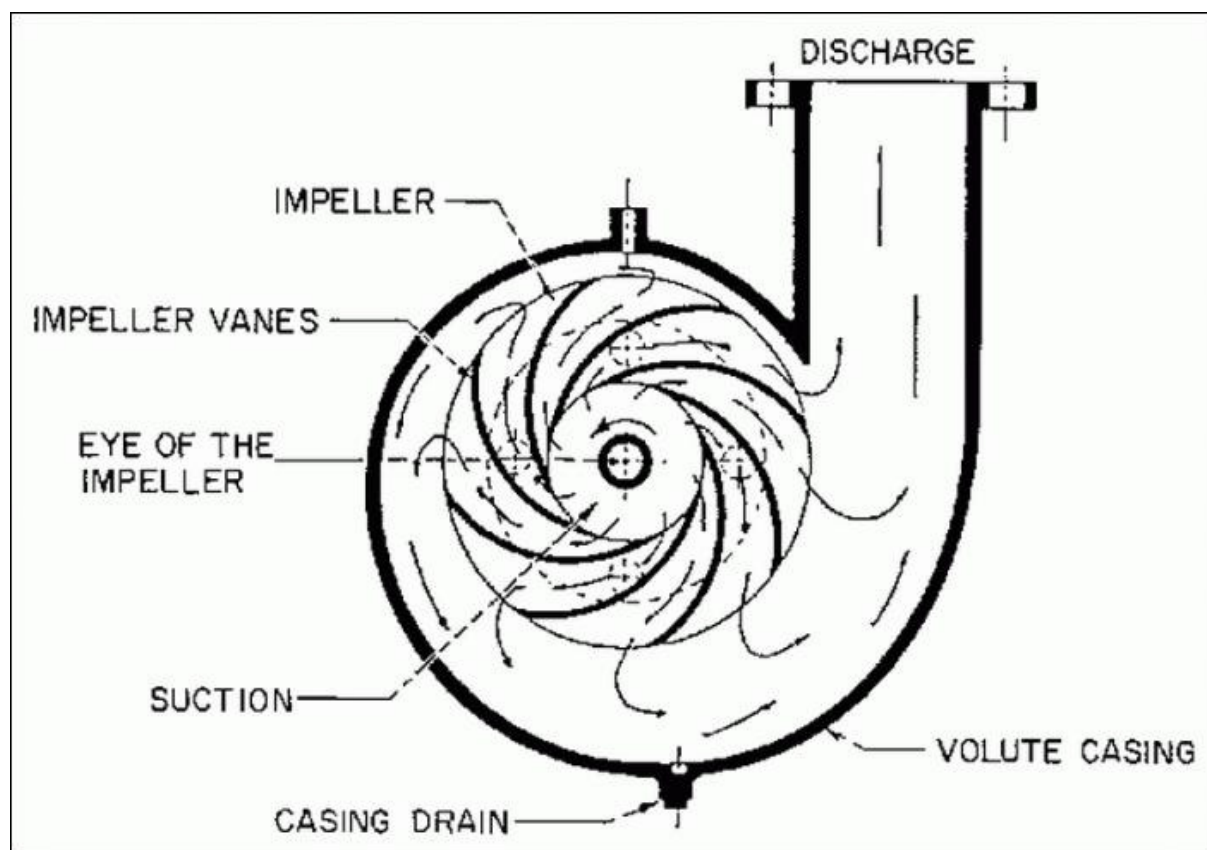
**How does it work?**

A centrifugal pump consists of an impeller and a volute (most end suction pumps) or a diffuser (submersible turbine pumps).

Rotation of the impeller forces water from its entry point, at the impeller eye, through the impeller's vanes and into the volute or diffuser. As water moves from the center of the circular impeller to its periphery, its velocity increases. When it reaches the volute, velocity is transformed into pressure.

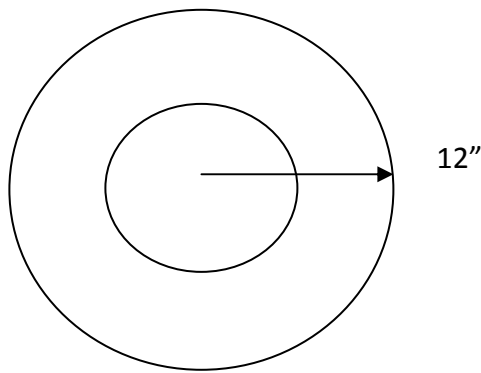
The flow and pressure created by a centrifugal pump depends upon the design of its impeller and its peripheral velocity. An impeller's peripheral velocity is dependent upon its diameter, its rotational speed or both.

A pump cannot pull or "suck" water into its impeller because water has no tensile strength. It lifts water by creating a low pressure area in its impeller eye that allows atmospheric or some other outside pressure to push the water into the eye that causes the suction.

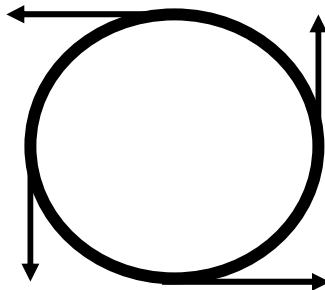


Peripheral Velocity

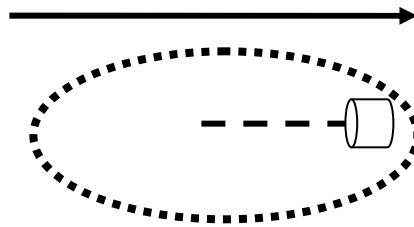
On the disc below, there are two points, one at 6" from its center and one at 12". The circle described at 6" has a circumference of 37.7" and the one at 12" has a circumference of 75.4". At a speed of one rotation per second, a point 12" from the center will travel twice the distance of a point that is 6" from the center. Therefore, its velocity must be ***twice*** as great.



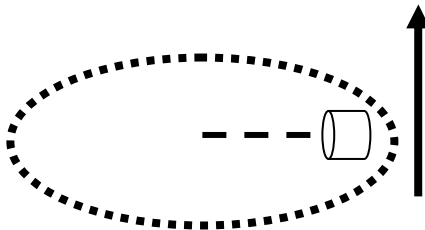
When an object is traveling in a circle, it can actually be moving in a straight line at any given time. The arrows at 12, 3, 6, and 9 o'clock show the direction of travel at that instant in time.



If centrifugal force were a true outward force, the swinging can would move in the direction of the arrow when the string was released.

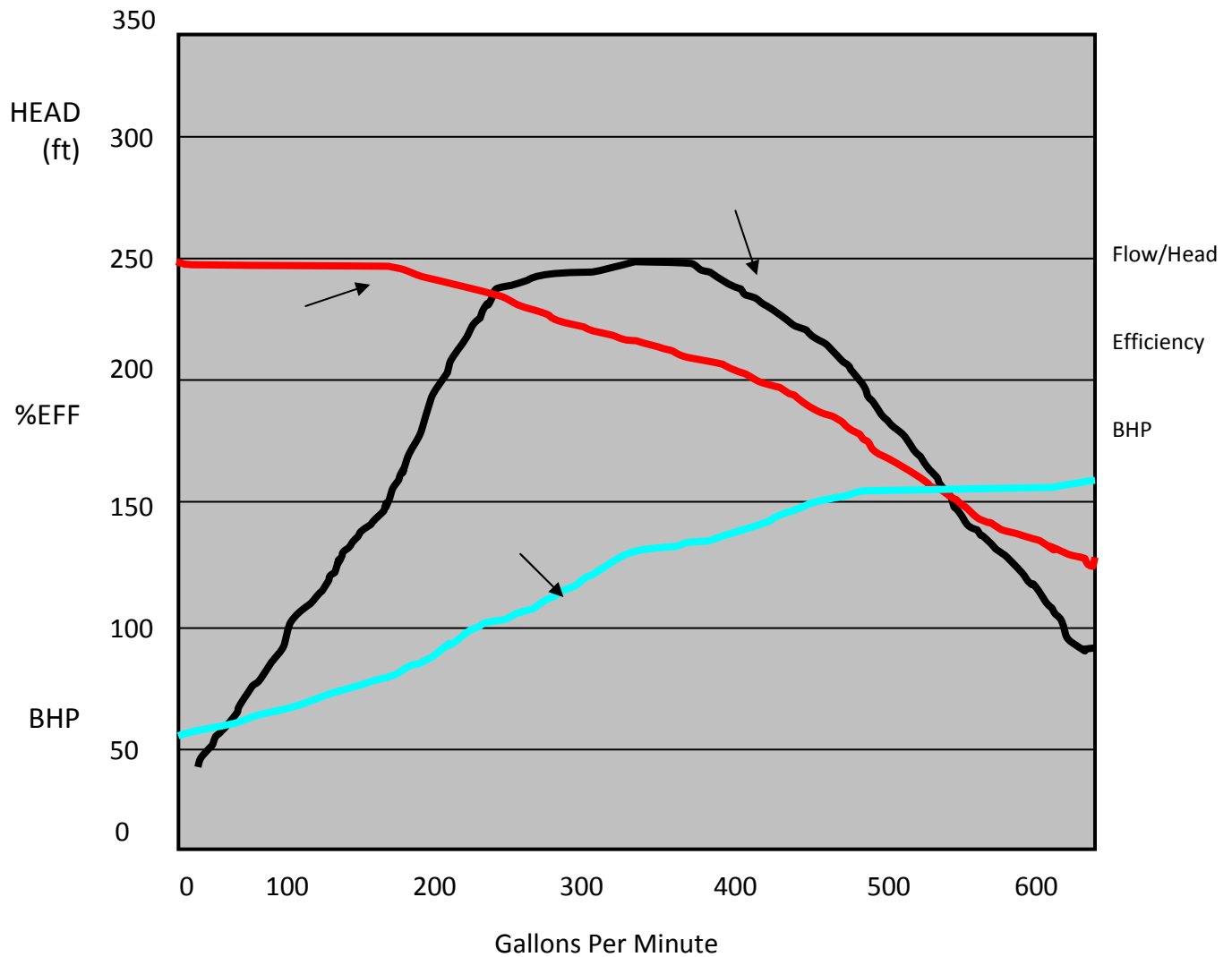


Instead, it actually moves in the same direction it was traveling at the exact moment it was released.

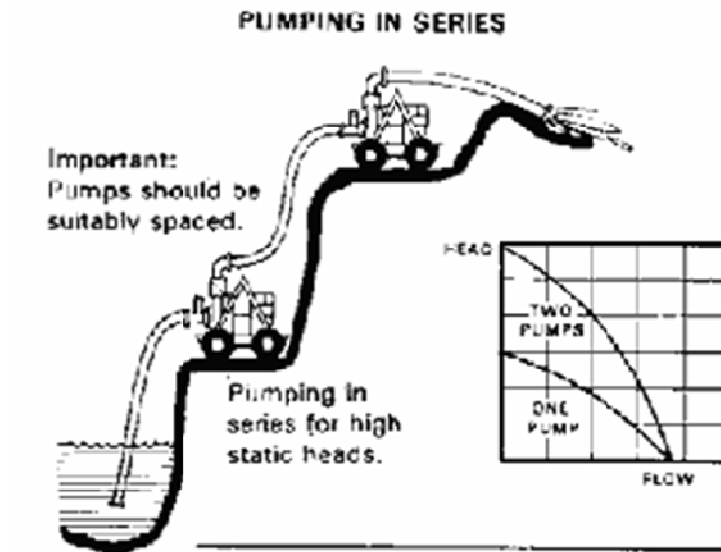


Pump Curves

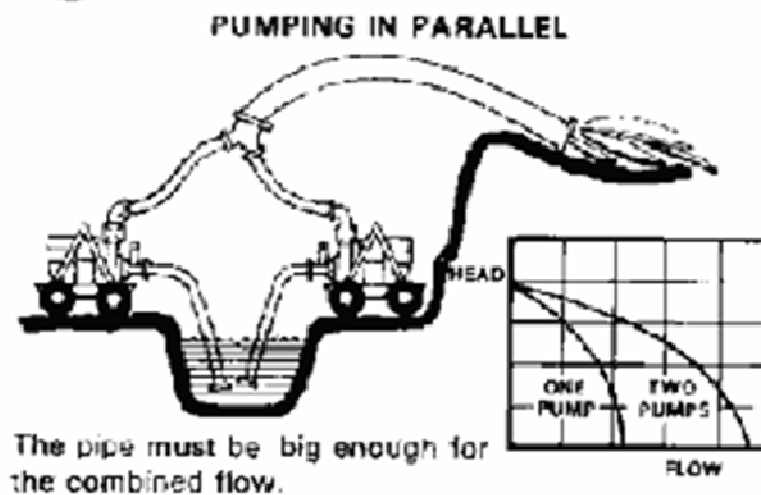
1. Flow and pressure at several points
2. Hydraulic Efficiency
3. Horsepower
4. Net Positive Suction Head Required (NPSHR)

Typical Pump Curve

A unique feature of centrifugal pumps is their ability to operate in series (one discharging into another) or parallel (common suction and discharge).

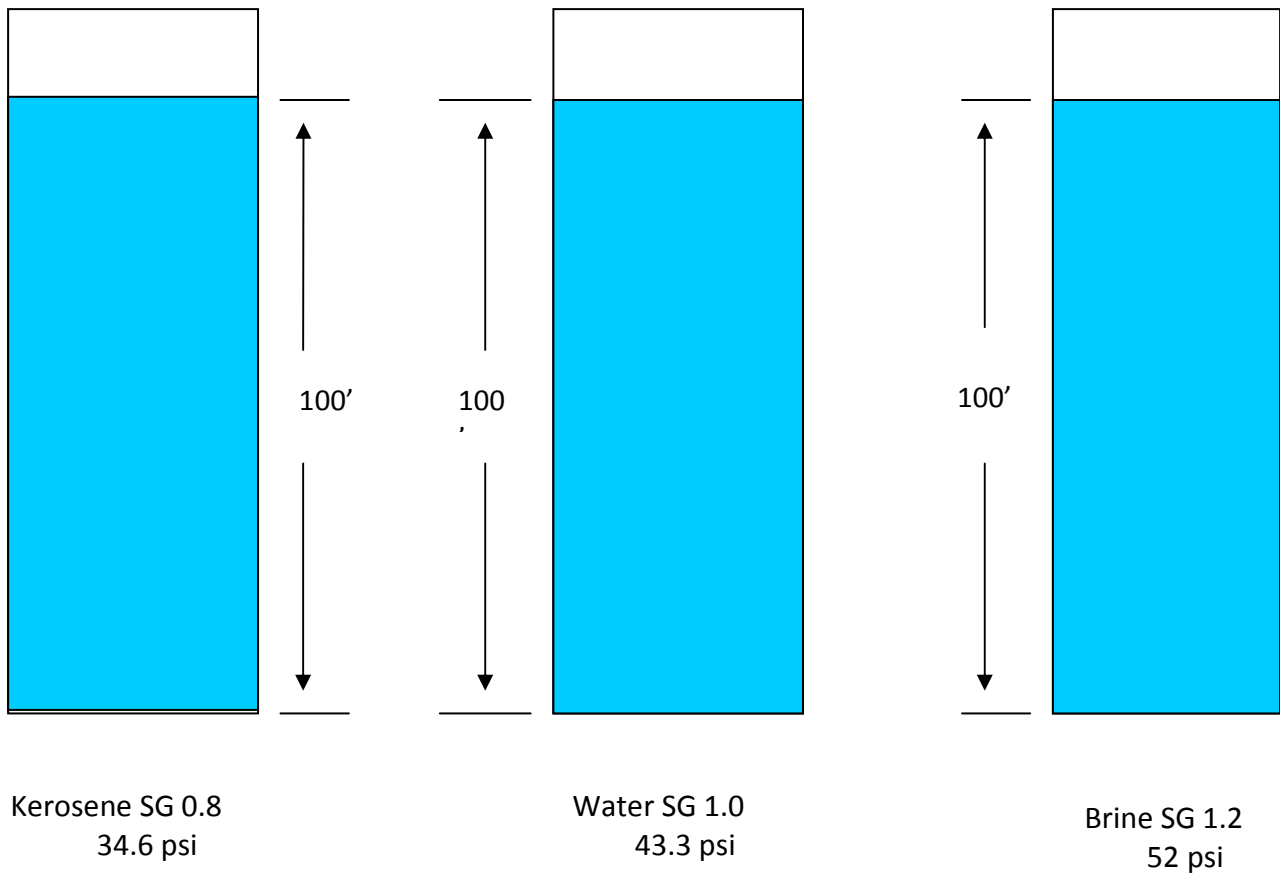


When operated in series, pressure is additive, while flow remains the same. In parallel, flows are additive, while pressure remains the same.



Why is pump pressure frequently stated in feet rather than PSI?

It is all about **specific gravity**. The specific gravity of a liquid is the ratio of its weight of a given volume to that of an equal volume of water. A centrifugal pump will always develop the same head in feet regardless of a liquid's specific gravity (weight). Pressure (psi) varies directly with specific gravity. The brake horsepower (BHP) required to perform a task varies directly with specific gravity.



What would register on the pressure gauge if the liquid had a specific gravity of 1.8 and the column of liquid was also 100' high?

$$100 \text{ ft} \times .433 \text{ psi} \times 1.8 \text{ SG} = 77.9 \text{ or } 78 \text{ psi}$$

1 gallon of this liquid weighs 15 lbs.

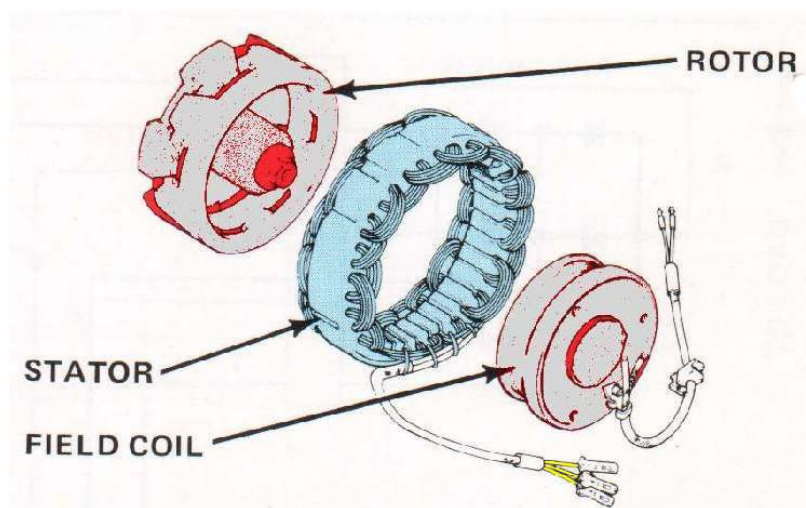
Affinity Laws

- Capacity (gpm) varies directly with a change in speed or impeller diameter.
- Head (feet or psi) varies as the square of a change in speed or impeller diameter.
- Brake horsepower (BHP) varies as the cube of a change in speed or impeller diameter.

Variable Frequency Operation

The rotational speed of an AC induction motor depends upon the frequency (Hz-hertz) of the current and the number of poles (coils) in its stator. At 60 Hz, a two-pole motor will complete 3600 rotations per minute (RPM).

For each frequency reduction of 1 Hz, a motor's speed will be reduced by 1/60 of its 60 Hz speed (60 RPM in this case).



Single-speed drives start motors abruptly, subjecting the motor to high torque and current surges up to 10 times the full-load current. In contrast, variable-frequency drives offer a "soft start" capability, gradually ramping up a motor to operating speed. This lessens mechanical and electrical stress on the motor system and can reduce maintenance and repair costs and extend motor life.

A variable-frequency drive is an electronic controller that adjusts the speed of an electric motor by modulating the power being delivered. Variable-frequency drives are an excellent choice for adjustable-speed drive users because they allow operators to fine-tune processes while reducing costs for energy and equipment.

For applications where flow requirements vary, mechanical devices such as flow-restricting valves or moveable air vanes are often used to control flow, which is similar to driving a car at full throttle while using the brake to control speed. This process uses excessive energy and may create punishing conditions for the mechanical equipment involved.

Variable-frequency drives enable pumps to accommodate fluctuating demand, running pumps at lower speeds and drawing less energy while still meeting pumping needs.

For a 25 horsepower motor running 23 hours per day (2 hours at 100% speed; 8 hours at 75%; 8 hours at 67%; and 5 hours at 50%) a variable-frequency drives can reduce energy use by 45%. At \$0.10 per kilowatt hour, this saves \$5,374 annually.

Because this benefit varies depending on system variables such as pump size, load profile, amount of static head, and friction, it is important to calculate benefits for each application before specifying a variable-frequency drive.

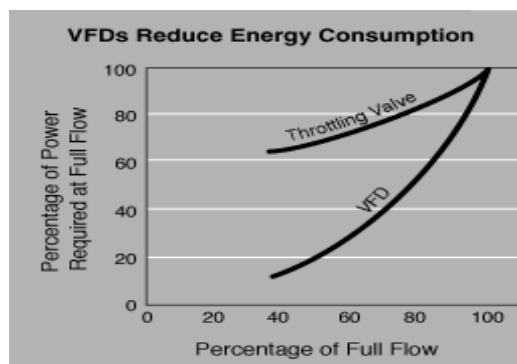


Figure 1. Energy consumption of VFDs and throttling valves.

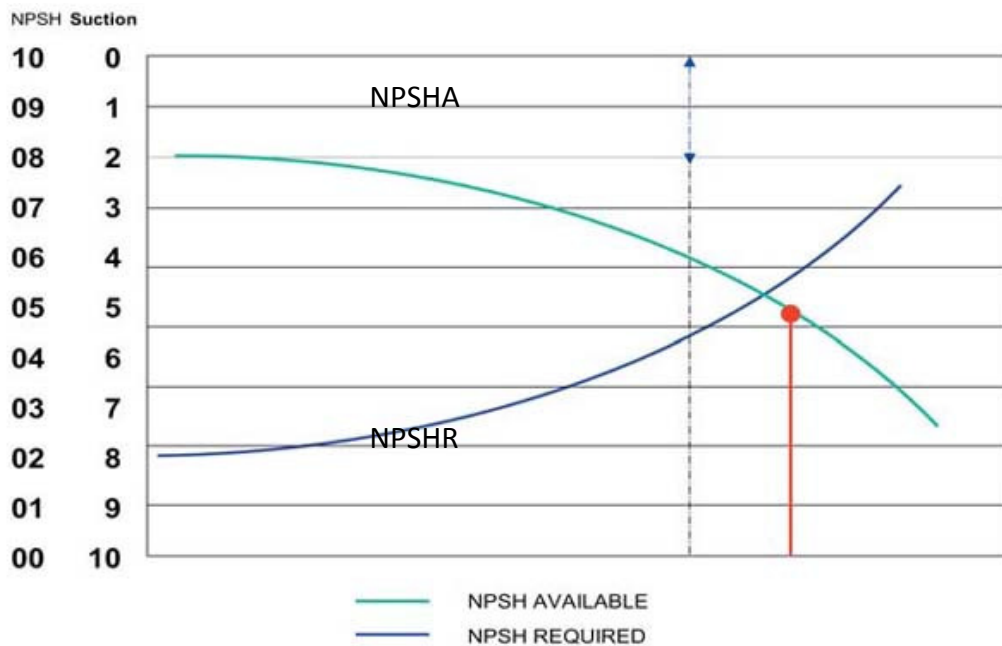
NPSHR & NPSHA

Maximum suction lift depends upon the pump design (NPSHR), the pressure applied to the surface of the water and the vapor pressure of the water under existing conditions (NPSHA). Net Positive Suction Head (NPSH) can be broken down into two components:

- NPSHR – The Net Positive Suction Head Required.
- NPSHA – The Net Positive Suction Head Available.

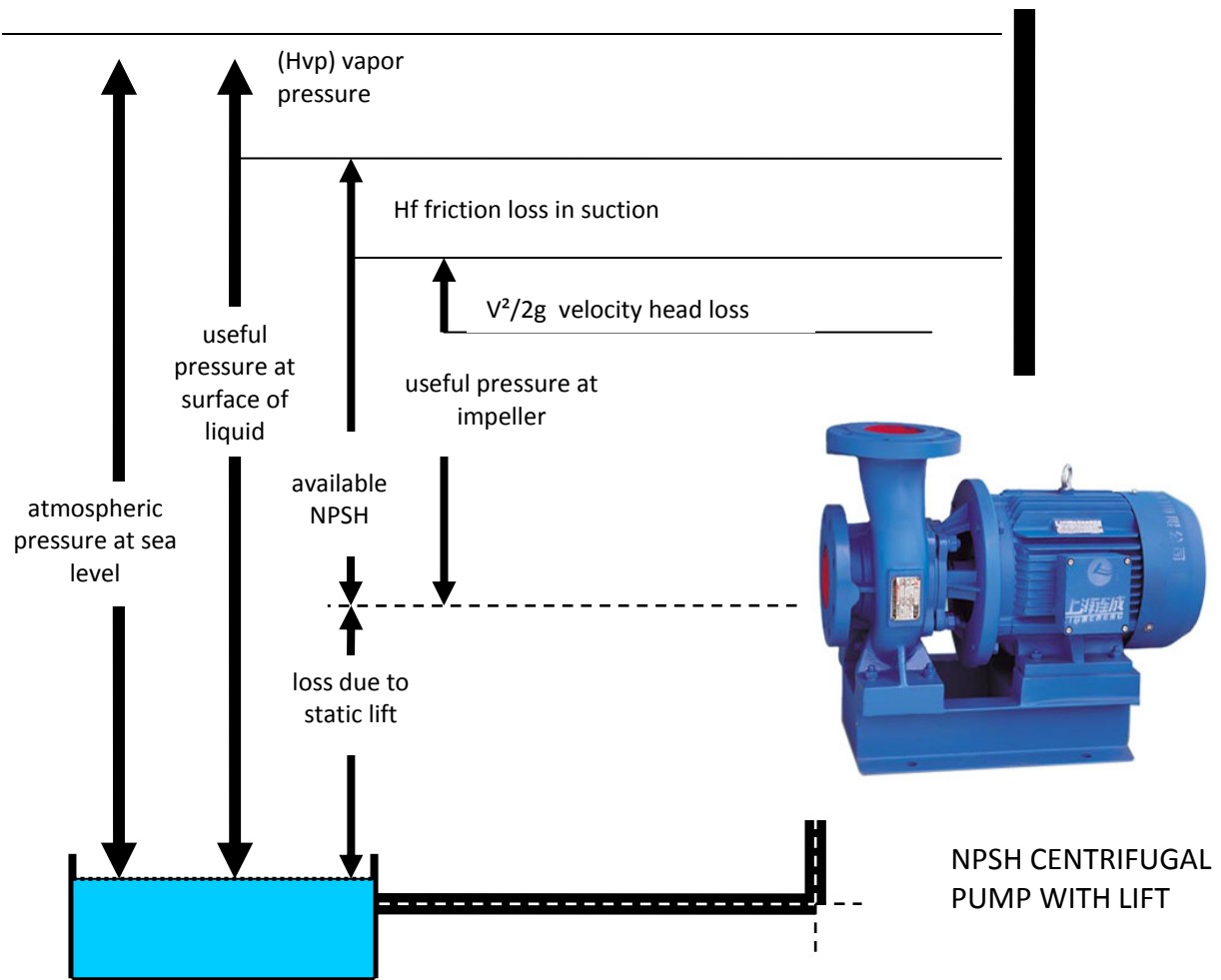
NPSHR is a function of a specific pump design. Simply stated, it is the pressure, measured at the centerline of the pump's suction, required to satisfactorily function at a given flow.

NPSHA is dependent upon the elevation or pressure of the suction supply, friction in the suction line, elevation of the installation (the pump itself in relation to other factors) and the vapor pressure of the liquid being pumped.

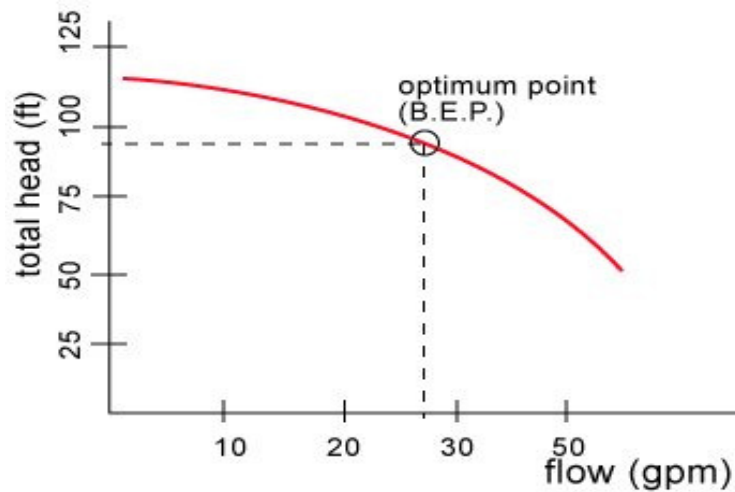


When a pump is in a suction lift condition, the only available energy is the atmospheric pressure of 14.7 psia or approximately 34 feet of water (14.7 psi X 2.31 ft of head = 33.957 feet). The amount of energy remaining after the energy consumption conditions have been satisfied is called **available net positive suction head (NPSHA)**.

$$\text{NPSHA} = H_a + - H_s - H_{vp}$$



Best Efficiency Point

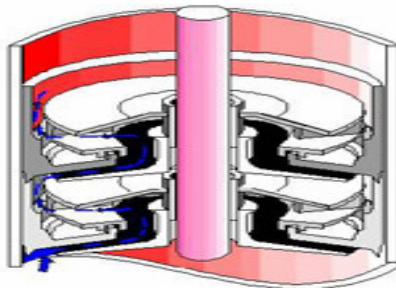


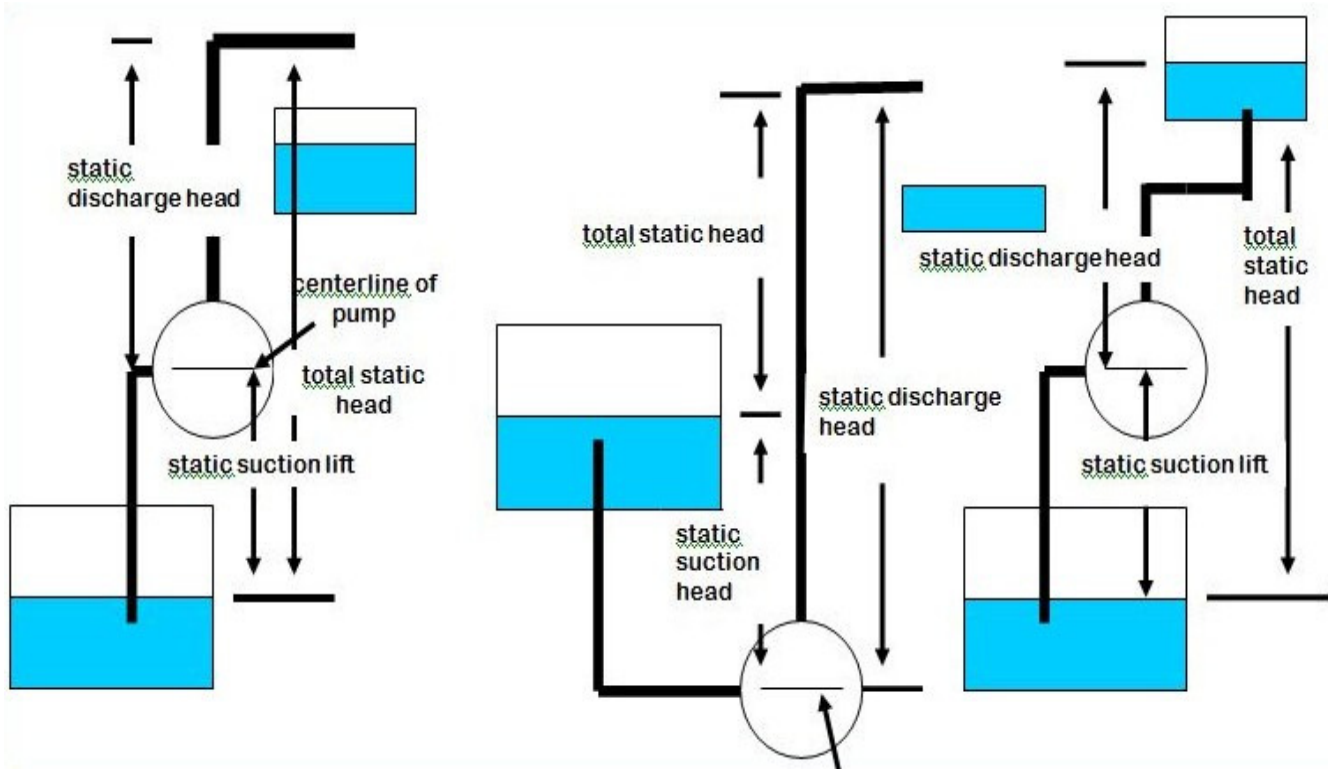
One of your 40 HP pumps breaks down and you have to order another one. You can ask your supplier for a 40 HP pump, but what you get may not work for the application you need it for. A quick glance at almost any major pump manufacturer's catalog would show twenty 40 HP pumps.

A pump's BEP (best efficiency point) is defined as the point on the head/capacity curve that has the highest hydraulic efficiency. The only information needed to use BEP pump sizing is the pump rotational speed, impeller diameter and suction size and discharge size.

All points above and below BEP have a lower efficiency, and the impeller is subject to vibration, heat and cavitation and causes premature bearing and mechanical seal failures due to shaft deflection. and heat can help produce cavitation.

A high-efficiency pump uses less energy (\$\$\$\$) to operate than a low-efficiency pump. If possible, it is recommended not to buy any pump that has efficiency of 55% or less.

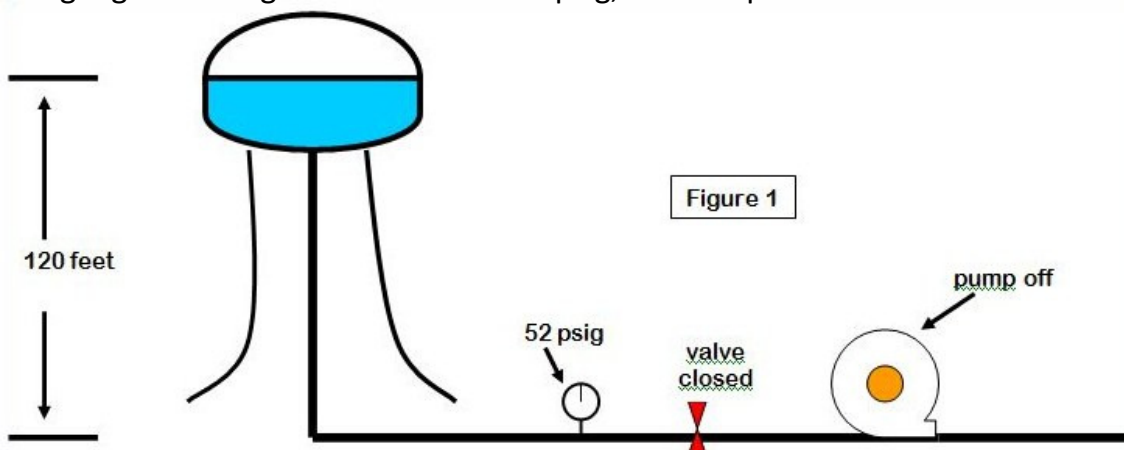




Pump Hydraulics

Head

Head is a measurement of the energy possessed by the water at any point in a water system. This is an indication of the pressure or force exerted by the water. Head can be expressed in feet or meters to represent the height of water above some reference point (Figure 1), or it can be expressed as pressure in pounds per square inch gauge (psig). The pressure gauge in the figure below reads 52 psig, which equals 120 ft.



Pressure Head

Pressure head is the amount of energy in water due to pressure. The reading on the pump on the previous page can be converted to feet by multiplying it by 2.31 ft. In Figure 1 (previous page), a pump would have to develop 52 psig or $52 \times 2.31 = 120$ feet of head to get water into the tank.

Velocity Head

Velocity head is the amount of energy in water due to its velocity or motion. The greater the velocity, the greater the energy and therefore, the greater the velocity head.

If V = velocity, the velocity head may be determined by the following formula:

$$\text{Velocity head (ft)} = \frac{V^2}{64.4 \text{ fps}}$$

For example, assume the water velocity in a pipeline on the discharge side of a pump is 6 fps. Therefore, the velocity head is $6 \text{ fps} \times 6 \text{ fps} \div 64.4 = 0.56$ ft of head. A centrifugal pump's volute or diffuser vanes change the velocity head produced by the spinning impeller in the water to pressure head.

Elevation Head

This is a measurement of the energy that water possesses because of elevation.

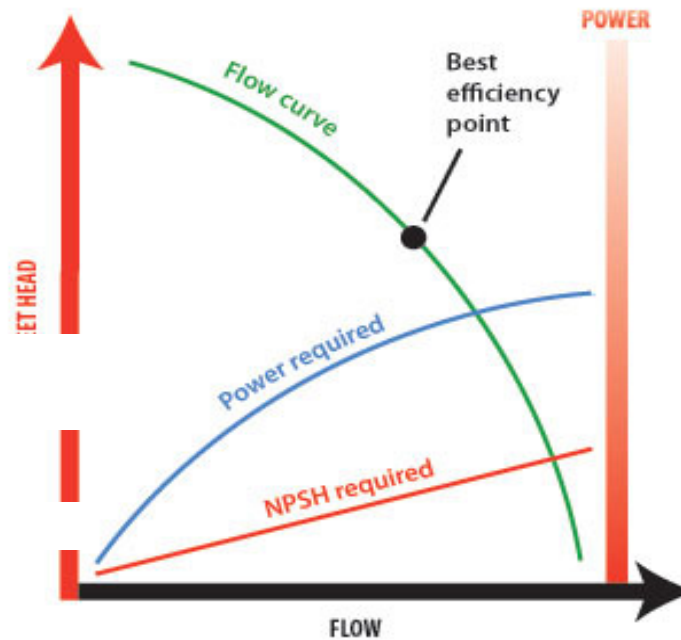
Total Dynamic Head

Total dynamic head is the amount of energy that a pump must transmit to move water. TDH varies for each pumping system and for flow conditions within a specific pumping system. **The term dynamic, again, indicates that the water is in motion rather than static (at rest).** TDH may be defined as the difference in height between the hydraulic grade line (HGL) on the discharge side of the pump and the HGL on the suction side of the pump. This is a measure of the total energy that the pump must impart to the water to move it from one place to another. Although there are others, one way to find the difference between the head on the discharge side and the head on the suction side of the pump is shown below.

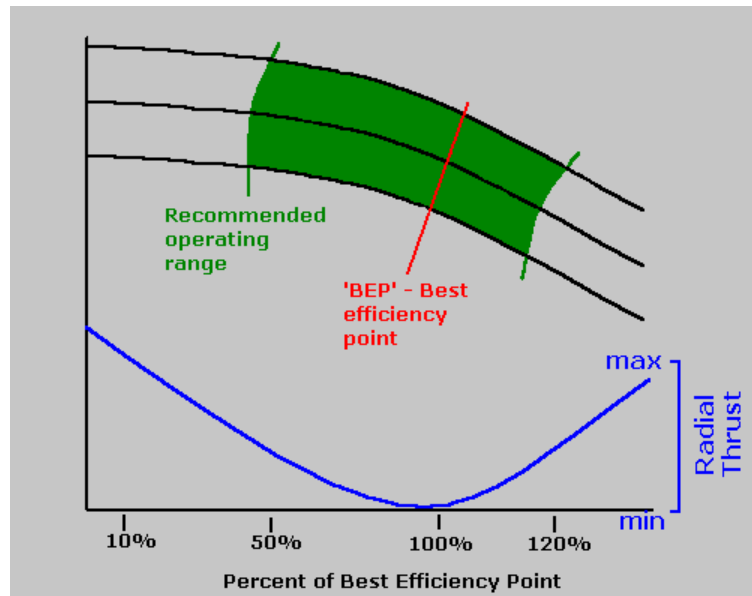
$$\text{Total Discharge Head} - \text{Total Suction Head} = \text{Total Dynamic Head}$$

Or, in a situation involving a suction lift, the TDH against which the pump must operate is the sum of the heads on the suction and discharge sides of the pump.

$$\text{Total Discharge Head} + \text{Total Suction Lift} = \text{Total Dynamic Head}$$

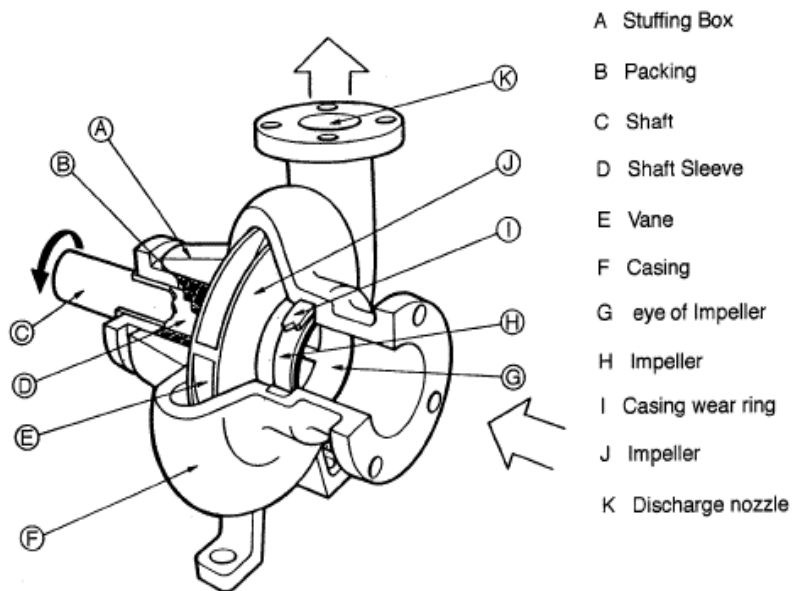
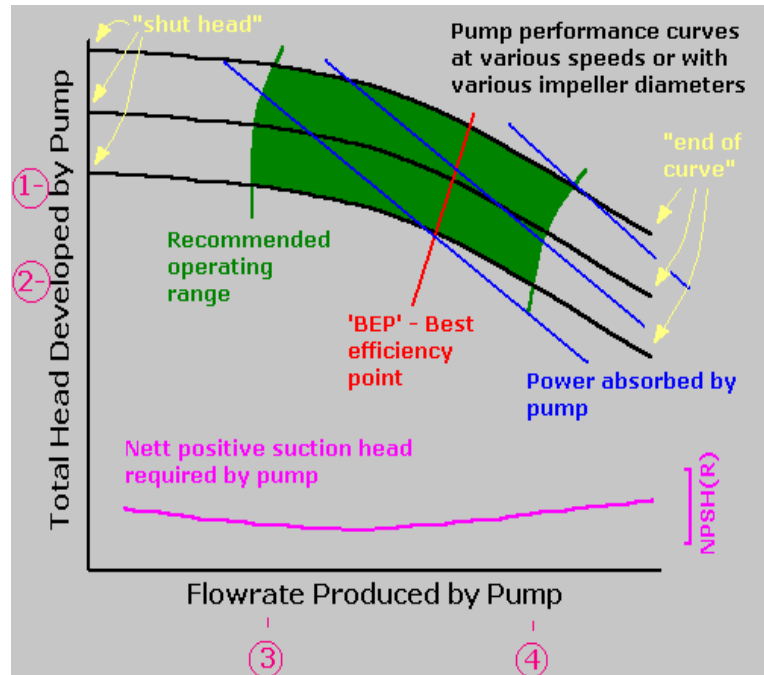


Pump curves are useful tools that will display the optimum performance that the particular pump can achieve. If the BEP doesn't meet your needs, then choose another pump that fulfills your needs.

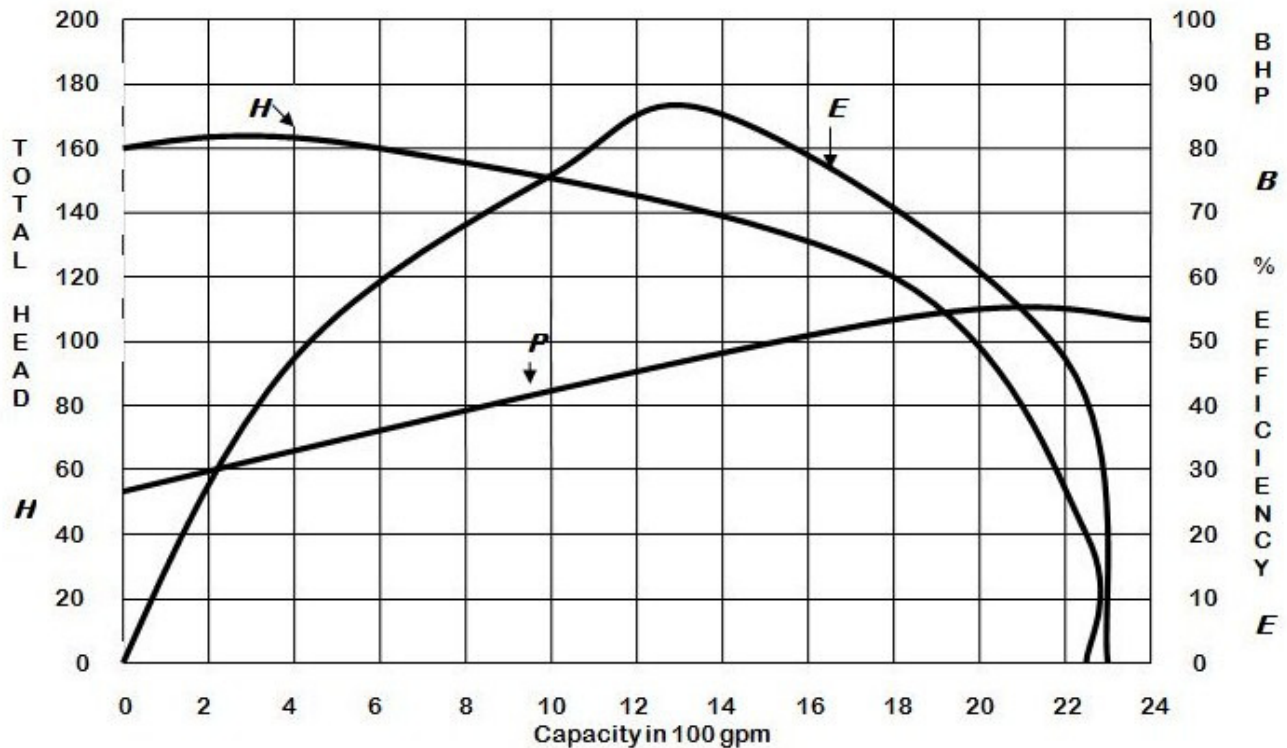


The circled numbers indicate the following for the pump curve to the right:

1. maximum recommended head
2. minimum recommended head
3. minimum recommended flow
4. maximum recommended flow



Pump Efficiency Curve



If this pump is producing 1400 gpm, find the corresponding total head, power, and efficiency.

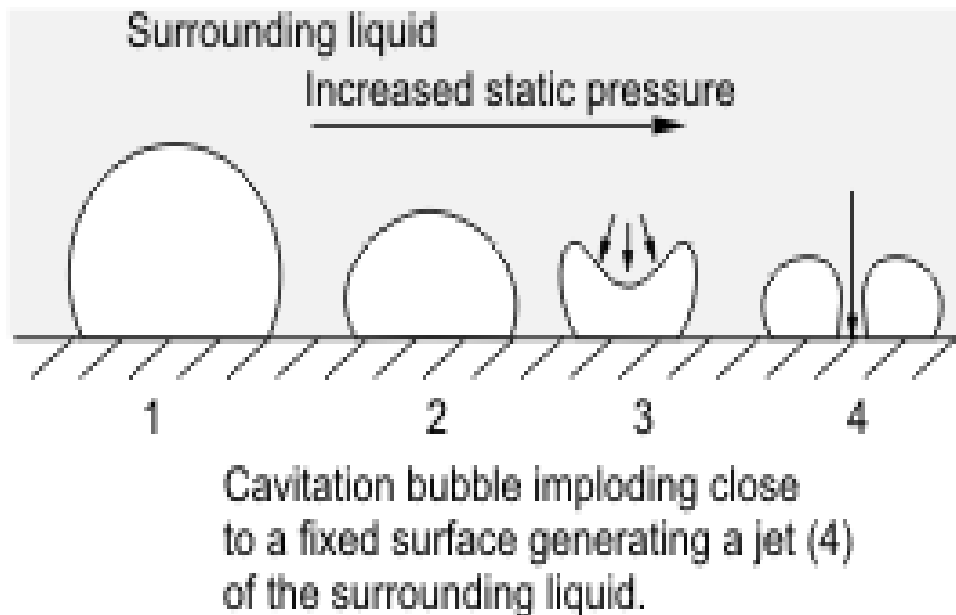
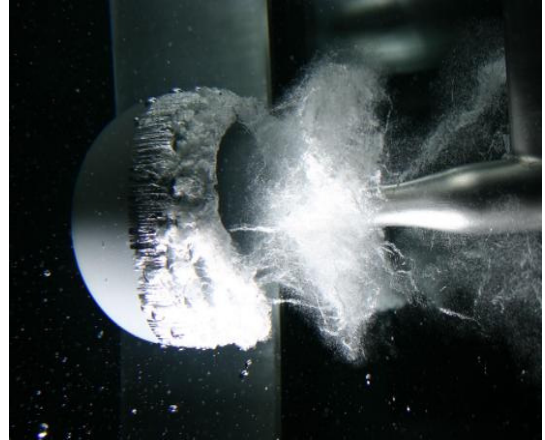
For a total head of 160 feet, use this pump curve to determine capacity, power, and efficiency of this pump.

$$\text{ELECTRICAL COST} = \text{MHP} \times .746 \times \text{hours of operation} \times \text{costs (\$/kWhr)}$$

Power costs \$0.05/kWhr, total head is 122 feet and motor efficiency is 80 percent. How much does it cost to operate this pump represented in the above pump curve for 18 hours?

Cavitation

This is the rapid formation and collapse of bubbles in a liquid, caused by the movement of something in the liquid, such as an impeller. These bubbles form at the low pressure or suction side of the pump, causing numerous things to happen at once. These bubbles or cavities will collapse when they pass into the higher regions of pressure, causing noise, vibration and damage to many of the pump components (such as a loss in pump capacity and the pump can no longer build the same amount of head).



*Pump efficiency drops.

Both suction cavitation and discharge cavitation can occur. Their causes and remedies can be similar, but not always. **Suction cavitation** occurs when the **Net Positive Suction Head Available** to the pump is less than what is **Required**.

$$\text{NPSHA} < \text{NPSHR}$$

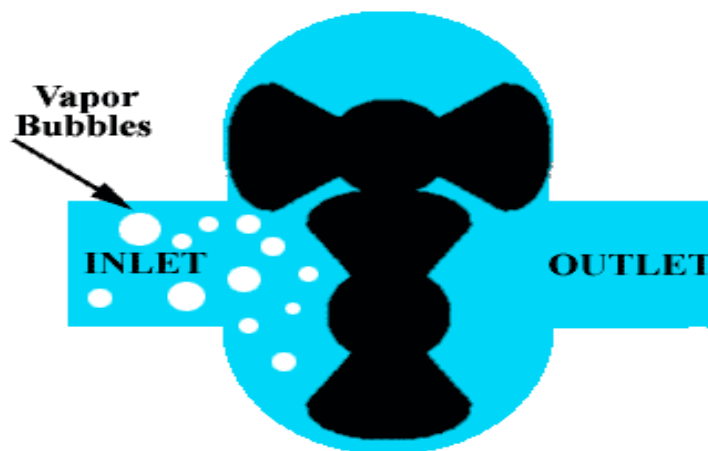
$$\text{NPSHA} = H_a \pm H_z - H_f + H_v - H_{vp}$$

Where:

- **H_a** is the atmospheric pressure.
- **H_z** is the vertical distance from the surface of the water to the centerline of the pump's suction (positive or negative).
- **H_f** is the friction formed in the suction piping.
- **H_v** is the velocity head at the pump's suction.
- **H_{vp}** is the vapor pressure of the water at its ambient temperature.

Depending on pump design, it is recommended that a NPSHA/NPSHA margin of 1.1 to 2.5 be maintained to limit cavitation occurrences.

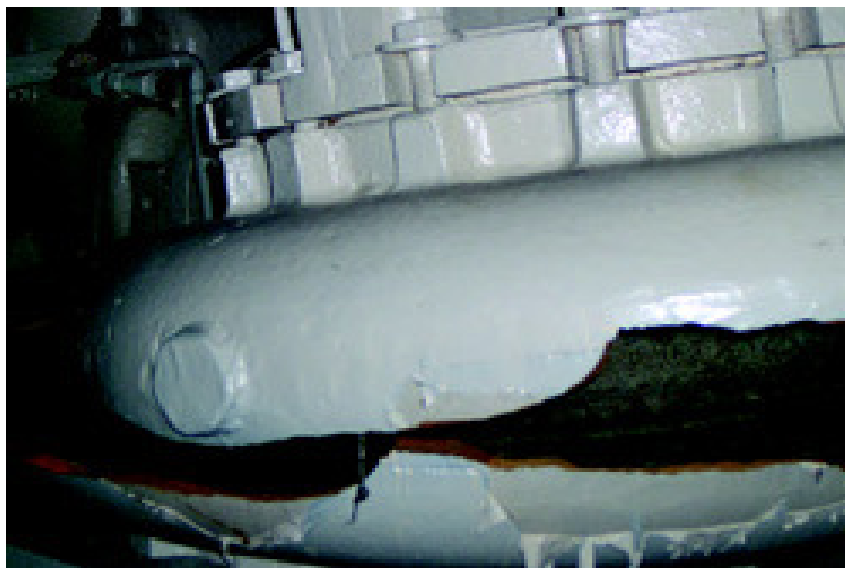
Discharge cavitation occurs when the pump discharge head is too high where the pump runs at or near shutoff.



Another detrimental effect of cavitation besides the decreased performance of the pump is the damage that can be incurred on impellers and other pump parts.



Misalignment, partially closed foot valves and too high of suction lift are frequent reasons for pump vibration.



Chapter 11 Review

1. What type of pump is the most frequently used pump in the distribution system and why?
2. A single volute centrifugal pump can only push water _____ feet because of _____ pressure.
3. The hydraulic grade line is an _____ line that denotes how high water would rise in a freely vented standpipe.
4. The C-Factor refers to energy lost moving water because of _____ factors.
5. In a pressurized pipeline, when velocity increases _____ decreases.
6. The terms kinetic and dynamic refer to water that is _____.
7. What horsepower is needed to overcome 180 ft of head at a flow of 1100 gpm with a pump efficiency of 85% and a motor efficiency of 71%?
8. Water is flowing through a 10" line at a velocity of 2.5 fps. What is the flow?
 - a. 4.8 ft³/sec
 - b. 2.9 ft³/sec
 - c. 610 gpm
 - d. 1.36 ft³/min
9. Cavitation occurs when water becomes _____ in the volute of a centrifugal pump.
10. Velocity and pressure have an _____ relationship.

Answers for Chapter 10 Review

1. Centrifugal
2. 33.9 ft, Atmospheric
3. Imaginary
4. Friction
5. Pressure
6. Moving
7. $HP = \frac{gpm \times ft \text{ of head}}{3960 \times Ep \times Em}$

$$HP = \frac{1100 \text{ gmp} \times 180 \text{ ft}}{3960 \times .85 \times .75}$$

$$HP = \frac{198,000}{2389.86}$$

$$HP = 82.9$$

8. $Q = A \times V$
 $Q = (.785 \times .833 \times .833) \times 2.5 \text{ fps}$
 $Q = .544 \text{ ft}^2 \times 2.5 \text{ fps}$
 $Q = 1.36 \text{ ft}^3/\text{sec} \text{ or } 610 \text{ gpm}$

9. Vaporized
10. Inverse

Chapter 12: CORROSION

Chapter 12 Objectives

1. Define the purpose of corrosion control.
2. Delineate associated problems with corrosion.
3. Define galvanic corrosion.
4. Define cathodic protection.
5. Enumerate the approaches to corrosion control.
6. Know, calculate, and understand the significance of Langelier's Saturation Index.
7. Identify the use of coupons in corrosion control.
8. Identify the purpose and significance of chemical inhibitors.
9. Delineate maintenance techniques for corrosion control.

Corrosion/Scaling

Both corrosion and scale can have detrimental effects on distribution systems.

- a. The presence of scale or tuberculation caused by corrosion on pipe walls increases friction losses and velocities while reducing water pressure.
- b. Larger pumps or extended pumping times will be necessary to deliver the same volume of water as was able with clean pipes, which, in turn, will require higher energy costs.
- c. Uncontrolled corrosion causes water leaks. Water loss and leak repair both can substantially increase operating costs.
- d. High levels of scale can significantly reduce water heater and boiler efficiency which, in turn, increase costs.
- e. Last, but certainly NOT least, is water quality deterioration. Microbial growths; the leaching of harmful metals, such as lead, copper and cadmium into the water; color, taste, odor and laundry and fixture staining are but a few of the myriad of water quality issues associated with corrosion and scale from chemically unstable water.



This is the mathematical formula for calculating reduced flows from scale or tuberculated piping:

$$Q \text{ (flow)} = A \text{ (area in ft}^2\text{)} \times V \text{ (velocity in feet per second (fps))}$$

If we start with a six (6) inch pipe with a velocity of 5 feet per second and account for the loss of flow because of reduced carrying capacity caused by scale or tuberculation, we can demonstrate this reduced carrying capability.

The flow in a 6 inch pipe:

$$Q = .196 \text{ ft}^2 \times 5 \text{ fps}$$

$$Q = \mathbf{.98 \text{ ft}^3/\text{sec}}$$
 or 439.8 gpm

The flow in a 5 inch pipe:

$$Q = .14 \text{ ft}^2 \times 5 \text{ fps}$$

$$Q = \mathbf{.7 \text{ ft}^3/\text{sec}}$$
 or 314.2 gpm

The flow in a 4 inch pipe:

$$Q = .08 \text{ ft}^2 \times 5 \text{ fps}$$

$$Q = \mathbf{.4 \text{ ft}^3/\text{sec}}$$
 or 179.5 gpm

The flow in a 3 inch pipe:

$$Q = .05 \text{ ft}^2 \times 5 \text{ fps}$$

$$Q = \mathbf{.25 \text{ ft}^3/\text{sec}}$$
 or 112.2 gpm

Purpose

- Prevent the effects of corrosion or excessive scaling in the distribution system.

Associated Problems

Public Health

Leaching of metals (lead, copper, cadmium, zinc, arsenic, etc.)

Shielding of pathogens from disinfection

Backflow from pressure loss due to leaks

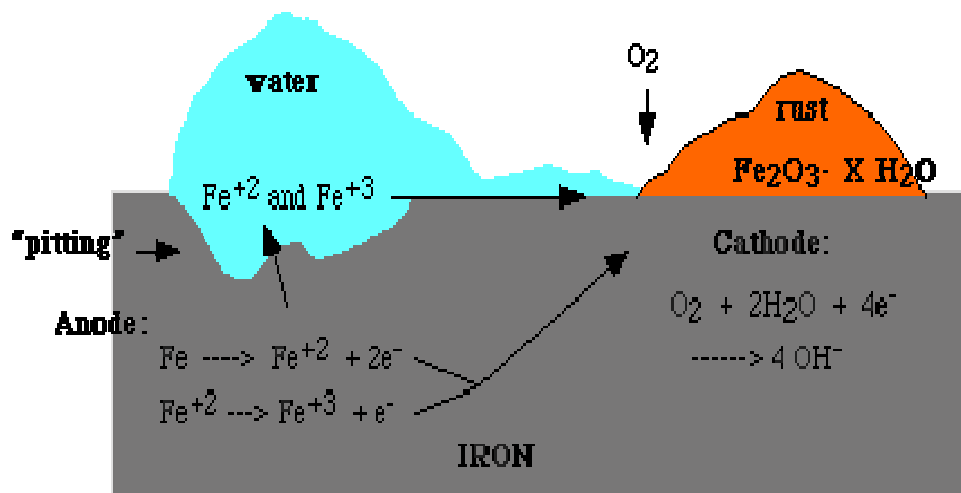
Destruction of fittings, pipelines, etc.

Economic

- Corrosion costs \$37,000,000,000 annually (estimated cost of corrosion in U.S.)
- Reduced pressure and carrying capacity due to increased friction on pipe walls and reduced inner diameters
- Higher energy costs because of the necessity of increased pumpage
- Increased energy costs for boilers and hot water heaters
- Increased maintenance costs associated with leaks
- Regulatory compliance

Water Quality and Aesthetics

- Increases in complaints for color, taste and odor, staining of laundry fixtures
- Increased turbidity
- Regulatory compliance



Factors Affecting Corrosion

pH

- Lower pH values increase the rate of corrosion
- Higher pH values promote the precipitation of calcium carbonate (scale), which can offer protection against corrosion

Alkalinity

- Carbonate and bicarbonate forms will form a protective scale coating that can reduce the rate of corrosion
- Helps control pH changes (buffering agent)

Hardness

- Calcium hardness may precipitate as calcium carbonate (CaCO_3) to reduce corrosion rate.
- Soft water is extremely corrosive

Dissolved Oxygen

- High levels of DO increases the rate of corrosion

Total Dissolved Solids (TDS)

- High TDS levels increase the conductivity of the water and in turn can accelerate the rate of corrosion

Temperature

- Higher temperatures speed chemical reactions and thereby corrosion rates

Chlorine Residual

- Increases metallic corrosion

Biological Activity

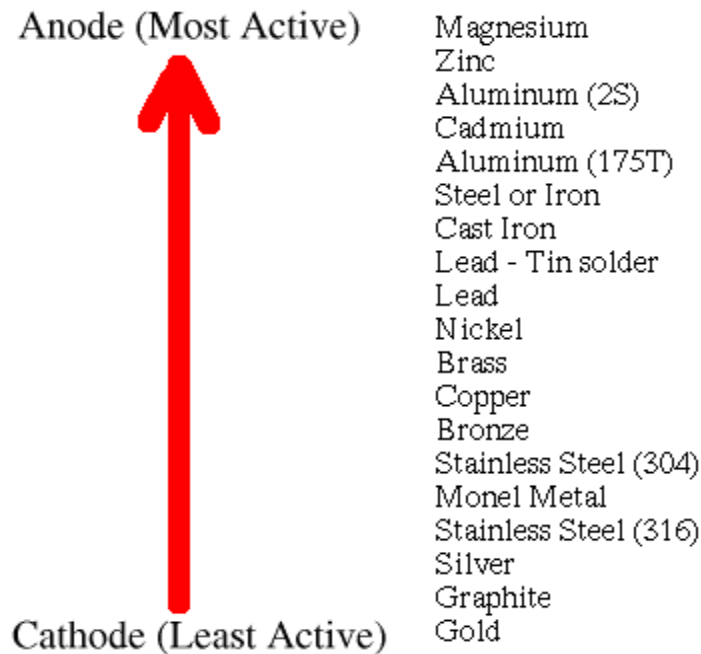
- This can increase or decrease the rate of corrosion dependent upon the type of organism and/or the degree of its activity.

Hydrogen Sulfide, Chloride, Sulfate, Electrical Current

- All of these increase the rate of corrosion.

Dissimilar Metals, Cathodic/Anodic Rate

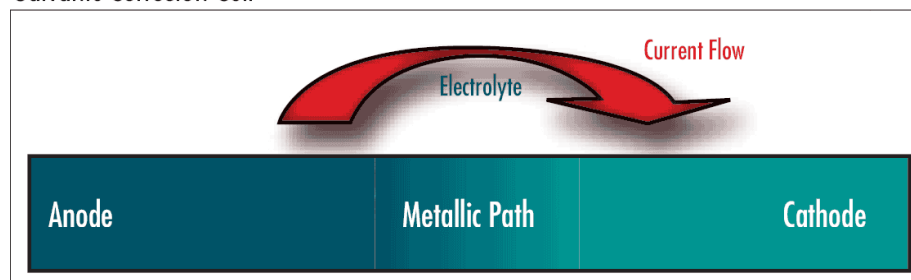
- Have an effect on the rate of corrosion



Galvanic Corrosion

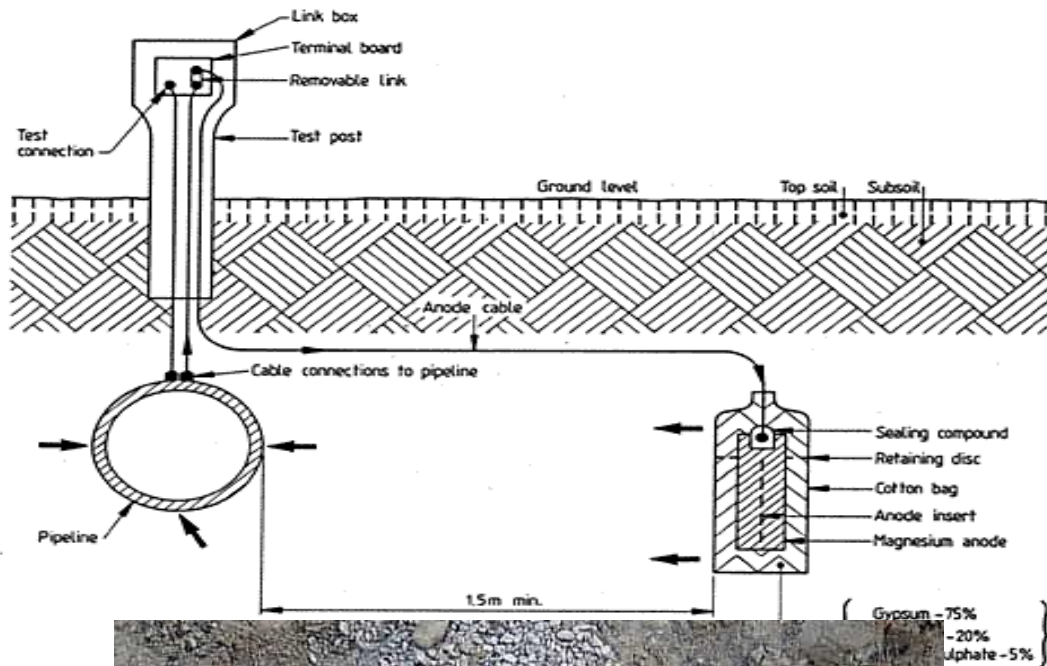
Involves the direct current electricity that is electrochemically generated between dissimilar metals. The further apart metals are on the galvanic series (chart on previous page), the faster the rate of corrosion. For instance, copper and lead would not react together as much as zinc and silver.

Galvanic Corrosion Cell



Cathodic Protection

Just as in storage tanks, one way to combat corrosion is to employ cathodic protection. Corrosion likes to pick on the easiest substance possible, so if we employ a sacrificial anode for corrosion to act upon, the corrosion rate relative to our underground piping can be greatly reduced.



Sacrificial anodes before being placed into water.



Sacrificial anodes after water contact.

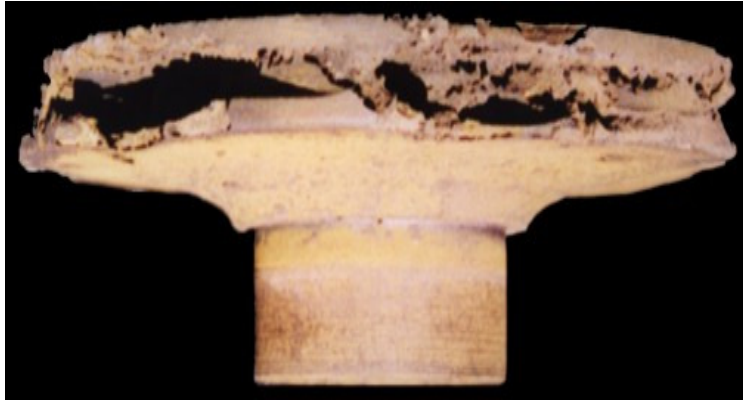


Three Basic Approaches to Corrosion Control:

1. Modify the water chemistry so that it is less corrosive to pipes and tanks.
2. Place a protective barrier or lining at the interface of the water and metal.
3. Use materials and design system so it is not corroded by a given water.

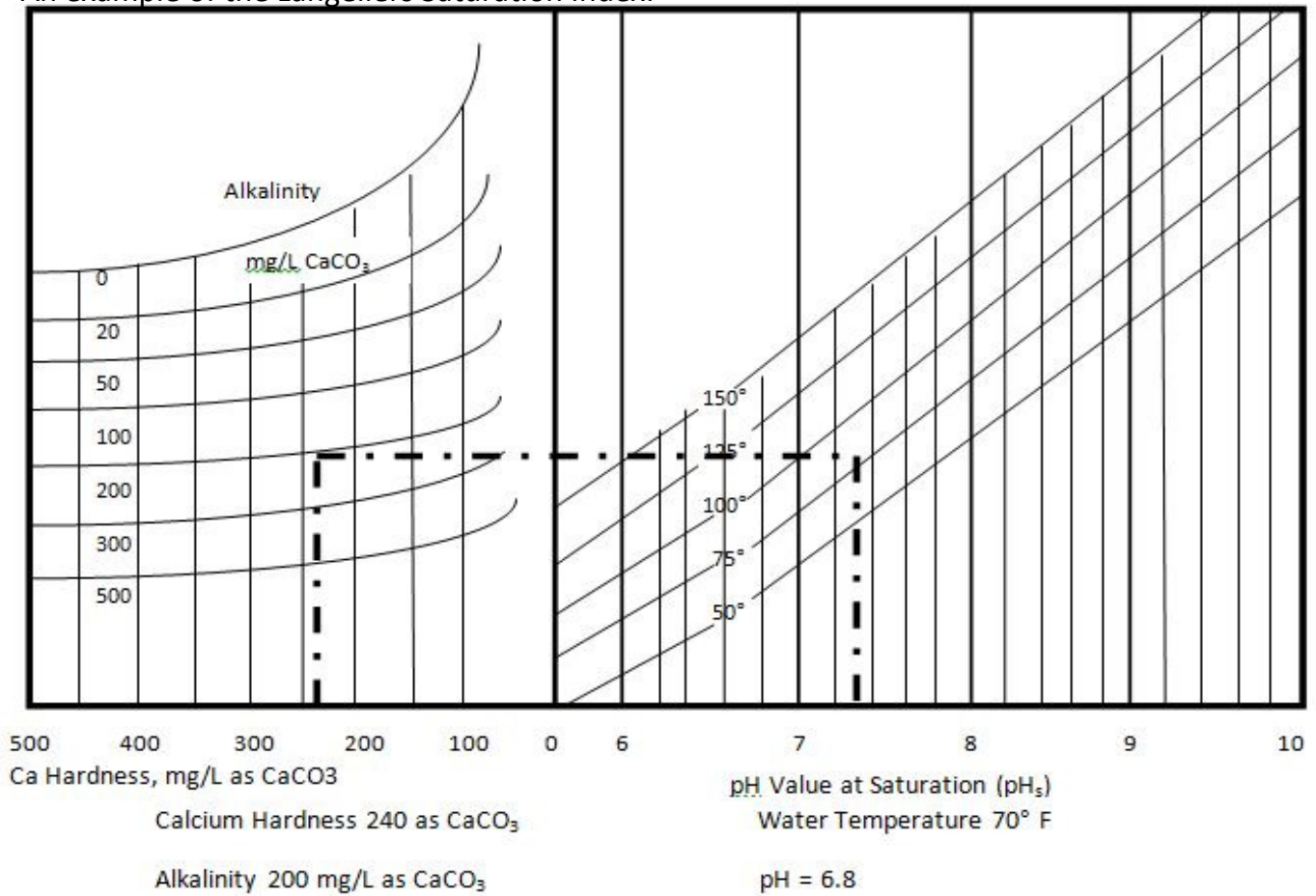
Common Ways of Achieving Corrosion Control:

1. Properly select system materials and system design
2. Modify water chemistry
3. Use corrosion inhibitors – ortho and polyphosphates.
4. Provide cathodic protection – sacrificial anodes, etc.
5. Use corrosion-resistant linings, coatings, paints, etc.



Langeliers Saturation Index

An example of the Langeliers Saturation Index:



Example

- Find the point marking calcium hardness of 240 mg/L as shown (bottom axis of graph between 200-300)
- Draw a line straight up to the 240 mg/L alkalinity curve.
- Draw a line from the 200 mg/L alkalinity curve across the graph to the point indicating a temperature of 70°F. (between the 50 and 75 diagonal lines)
- Draw a line straight down from the temperature point to the bottom of the graph, intersecting the line marked "pH Value"
- Identify the pH value (point at which your line intersects the pH line)
- In the above example, the pH value is 7.3

Calcium hardness	240 mg/L
Water temperature	70° F
Alkalinity	200 mg/L
Actual pH (pHA)	6.8
Saturation pH (pHS)	7.3
pHA	6.8
pHS	<u>7.3</u> - 0.5

This water has the likelihood of being corrosive.

Formula

$$\text{pHA} - \text{pHS}$$

If answer is positive (+) = scale forming

If answer is negative (-) = corrosive

- A negative number when you employ the Langeliers Saturation Index means your water has the likelihood to be aggressive (corrosive).
- A positive number derived when you employ the LSI indicates that your water has the likelihood to be scale-forming.

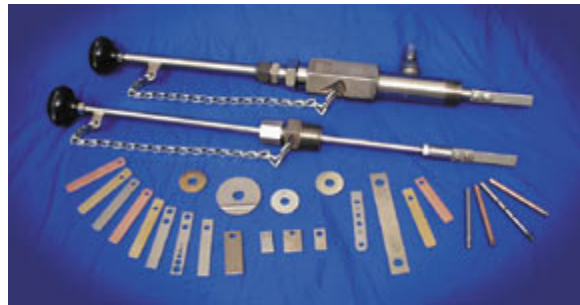
Example

After taking into account the alkalinity levels, temperature, and calcium hardness, as well as the actual pH (pHA) of 7.3, the saturation pH (pHS) is 7.0. Is your water corrosive or scale-forming?

$$\begin{array}{r} \text{pHA } 7.3 \\ \text{pHS } \underline{7.0} \\ + 0.3 \end{array}$$

In this case, your water has the likelihood of being scale-forming because a positive sum was reached.

Coupons placed in water can give the operator a good idea of water chemistry in the system. These coupons can be pieces of pipe removed for service connections or can be bought and placed into flow to determine whether the water is corrosive or scale forming.

***Chemical Inhibitor Feed***

Ortho or polyphosphates can be added to the water to impart a thin scale deposition on the inside of pipes to protect against corrosion.



Water Quality and Consumer Complaint Record Evaluation

Customers can provide some pertinent information if you record their complaints. Preparing a standard form for customer complaints and then compiling that information can provide a basis for pipe inspection, which could reduce the probability of a leak.

Maintenance

Line change out program - Start with a dollar amount or a percentage of proposed revenue and replace your worst pipes in an orderly, logical manner. The pipe condition will not improve over time.



Chapter 12 Review

1. True/False PVC pipes corrode.
2. _____ corrosion is the placement of dissimilar metals in contact with each other.
3. Some health factors that corrosion can cause are _____ and _____.
4. An important source of data would include _____.
5. Economic issues associated with corrosion or scaling could include _____, _____, _____ and _____.
6. If actual pH measures 7.2 pHA and the saturation pH is 7.7 pHS, is the water corrosive or scale forming?

Answers for Chapter 12 Review

1. True
2. Galvanic
3. Permeation, Leaching
4. Pipe coupons, customer complaints
5. Reduced carrying capacity, degradation of pumps and appurtenances, leaks, water loss
6. 7.2
-7.7
-0.5 - Corrosive

Chapter 13: CROSS - CONNECTION CONTROL

Chapter 13 Objectives

1. Define cross connection.
2. Define the necessity of cross connection controls.
3. Define backflow and delineate its two types.
4. Delineate the necessity of intense focusing on cross connection issues.
5. Delineate backflow prevention devices and determine the proper uses of each.
6. Notate and differentiate the two types of cross connection programs.
7. Differentiate the degree of hazards in cross connections.
8. Delineate the necessities of a successful cross connection control program.

What is a cross - connection?

A cross - connection is a physical link between a potable water supply and one of unknown or questionable quality.

Why are controls necessary for cross – connections?

A cross connection can link our expensive, well-maintained, safe to drink from, distribution system to a source that contains something that could make people sick or even kill them. This potentially dangerous source can enter our potable water supply by means of **backflow**.

What is backflow?

Backflow is the flow in the opposite direction to the normal flow or the flowing back of a substance toward its source. In distribution systems, it is the flowing of water from a service connection into our main lines or in the opposite direction from which it was intended. Backflow can occur in two ways: Backpressure or Back-siphonage

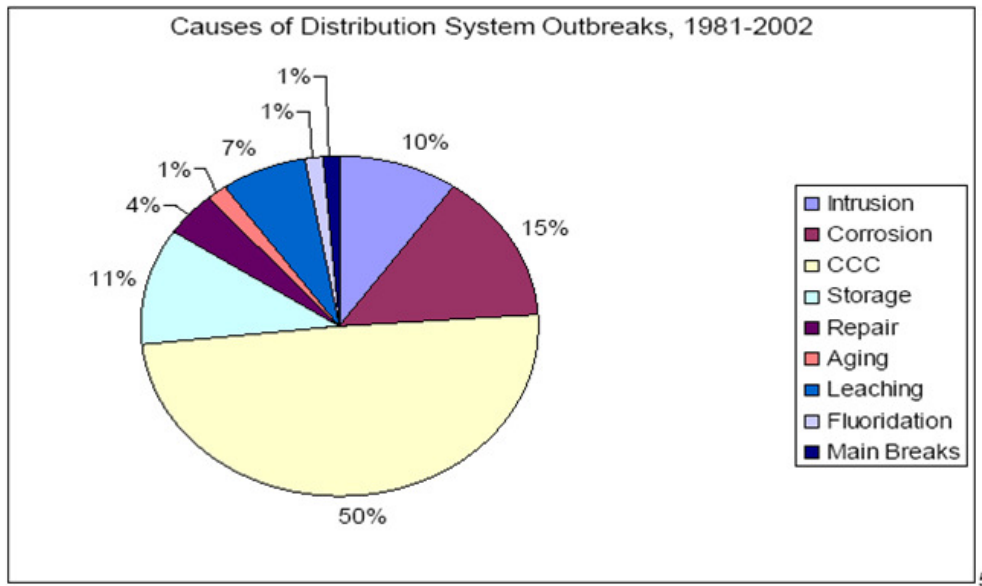
Backpressure occurs when the pressure in the service connection receiving the potable water becomes greater than the pressure in the distribution system. When this occurs, water and other substances in the customer's piping will flow from higher pressure to lower pressure and force this potentially contaminated water back into the distribution system. Potential sources of backpressure include boilers, pumps and elevated tanks on the customer's side of the service connection.

Back-siphonage occurs when the pressure in the distribution system falls below atmospheric pressure (14.7 psia) that creates a partial vacuum or siphon. Siphoning gasoline from a car or even drinking from a straw lowers the air pressure creating a partial vacuum. Liquid flows from an area of higher pressure (the gas tank or soda can) to one of lower pressure (the end of the siphon tube or straw). Some conditions that can cause low pressure in a distribution system at certain times are listed below:

- a. Flushing
- b. Fire fighting
- c. Main breaks
- d. Undersized piping (which increases velocity and thereby lowers pressure)

BACKFLOW HAS BEEN IMPLICATED IN MORE WATERBORNE ILLNESS OUTBREAKS THAN ANY OTHER FACTOR. Because most pathogens are microscopic and cannot be detected by the naked eye, experts suspect that the majority of backflow incidents are never detected or recorded.

Why Focus on Cross Connections in Initial Model?



AWWARF

The American Water Works Research Foundation did a study on backflow incidents. There were between 2800 and 4100 backflow incidents among those who responded to their questionnaire. Of those, there were anywhere from 1100 to 1750 backflow incidents that were substantially documented. Initially, only about half of those utilities questioned responded to their questionnaire. If the numbers given were extrapolated or carried out to include the entire number of those questioned, then the number of backflow incidents could number from 800,000 to 1,000,000 since 1970.



Incident # 1

The services to seventy-five apartments, housing approximately three hundred people, were contaminated with chlordane and heptachlor in a city in Pennsylvania. The insecticides entered the water supply system while an exterminator was applying them as a preventative measure against termites. While the pesticide contractor was mixing the chemicals in a tank truck with water from a garden hose coming from one of the apartments, a workman was cutting into a 6-inch main line to install a gate valve. The end of the garden hose was submerged in the tank containing the pesticides, while at the same time, the water to the area was shut off and the lines drained prior to the installation of the gate valve. When the workman cut the 6-inch line, water started to drain out of the cut, thereby setting up a back-siphonage condition. As a result, the chemicals were siphoned out of the truck, through the garden hose, and into the system, contaminating the seventy-five apartments. Repeated efforts to clean and flush the lines were not satisfactory, and it was finally decided to replace the water main and all the plumbing that was affected.

Incident # 2

When a utility has no backflow prevention program for its town water system, the system can plummet out of control. An emergency intrusion of a foreign substance causes confusion and fear, which is then usually followed by a tremendous expense. Members of the Woodsboro, Maryland Fire Department went door-to-door warning citizens of the danger of using town water, which may have been contaminated with lethal pesticides.

“Yellow gushy stuff” had poured from some of the faucets in town and the State of Maryland had placed a ban on drinking Woodsboro city water. Residents were warned not to use the water for any reason except to flush toilets. The incident drew widespread attention and made the local newspapers, in addition to being the lead story on the ABC news affiliate in Washington, D.C., and virtually all the newspapers in the area. The investigation disclosed that the water pressure in the town water mains was temporarily reduced due to water pump failure in the town’s high-service pumps. Coincidentally, a gate valve had been left open. A lethal cross – connection had been created that permitted the herbicide to flow into the potable water supply system. Upon restoration of water pressure, the herbicide flowed into the many faucets and outlets in the town’s distribution system.

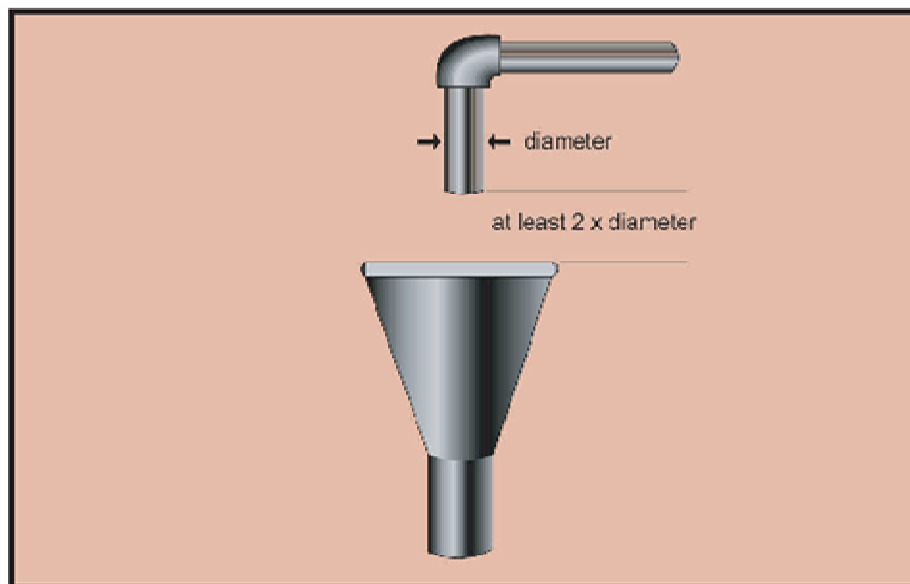
THERE ARE HUNDREDS OF THESE INCIDENTS PUBLISHED, SOME OF WHICH INVOLVE FATALIES DUE TO THE INGESTION OF WATER THAT HAS BEEN CONTAMINATED.

Who is responsible when a backflow incident occurs? Your entity, as a federally regulated water purveyor or supplier, is bound by the SDWA to supply potable water to your customers. If it is within your power to prevent contaminated water from entering your distribution system and someone gets sick or dies because of contaminated water, then do you think you might get sued?

Methods and Devices for the Prevention of Backpressure and Back-siphonage

Air Gap

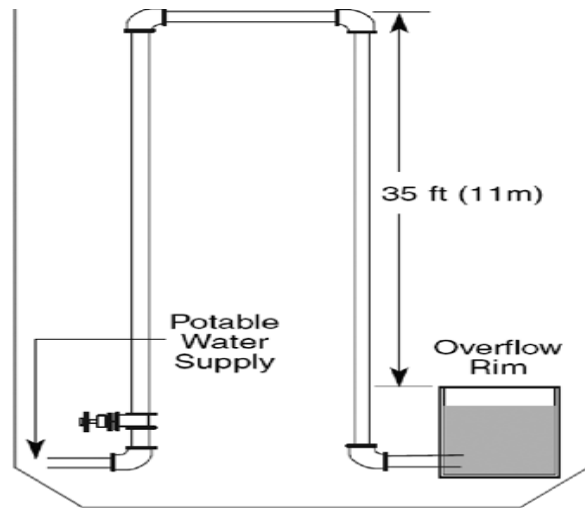
An air gap is a physical separation between the water supply and the intended end point. Air gaps are non-mechanical devices that are very effective against both backpressure and back-siphonage. The air gap must be twice the supply pipe diameter, but never less than one inch.



8.4.1 Figure 1: Air gap

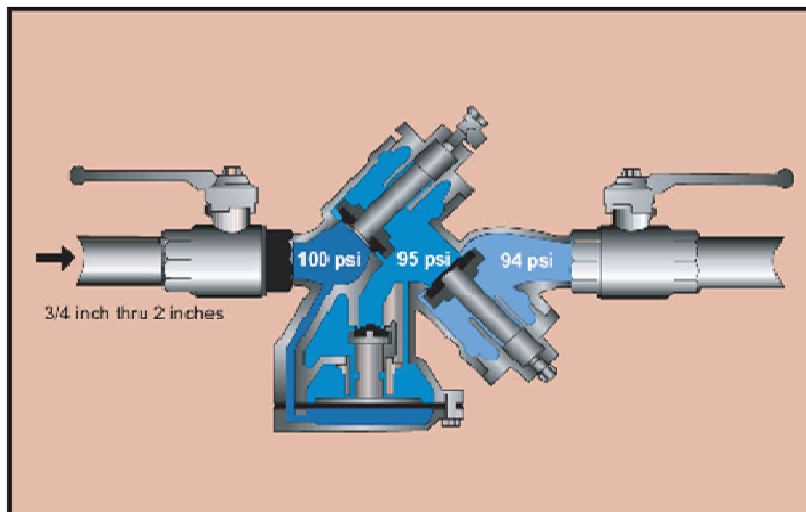
Barometric Loop

The barometric loop consists of a continuous section of supply piping that abruptly rises to a height of approximately 35 feet and then returns down to the originating level. In operation, for protection against back-siphonage, it employs the principle that a water column, at sea level, will not rise above 33.9 feet.



Reduced Pressure Zone

Reduced Pressure Zone (RPZ) assemblies are effective against backpressure and back-siphonage. These devices are essentially modified double-check valves with an atmospheric vent capability placed between the two check valves. This device is designed such that this “zone” between the two checks is always kept at least two pounds less than the supply pressure.

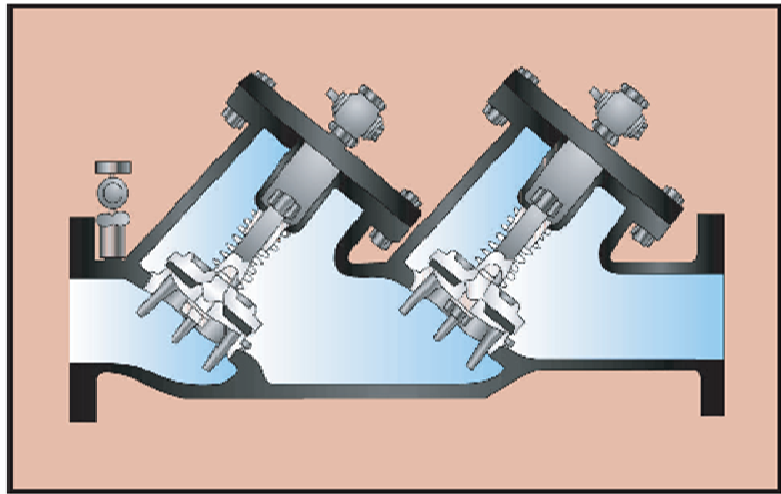


8.4.5 Figure 1: Reduced pressure principle assembly (RP)

In the event of backflow, the RPZ will send the water through the atmospheric vent instead of attempting to send the water back to the main.

Double-Check Valve

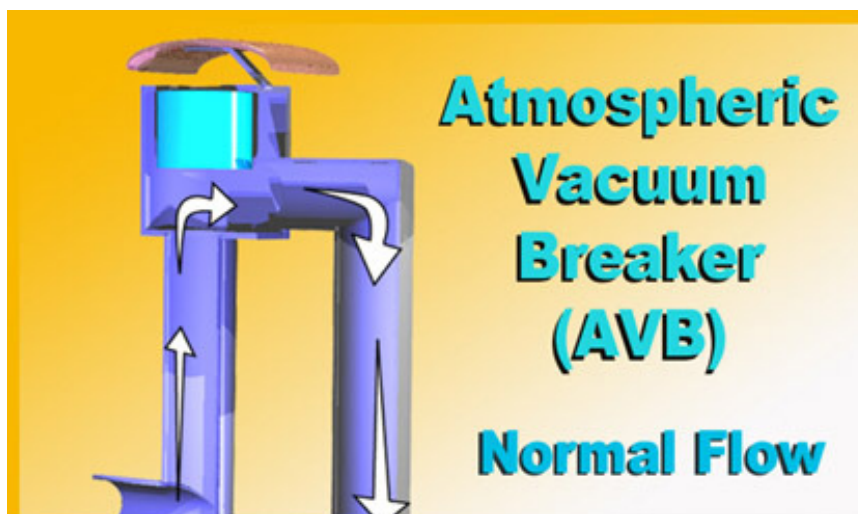
A double-check valve is basically two single-check valves coupled within one body. Each check is spring-loaded closed and requires approximately one pound of pressure to open. This device protects against back-siphonage and backpressure.



8.4.4 Figure 1: Double check valve assembly (DC)

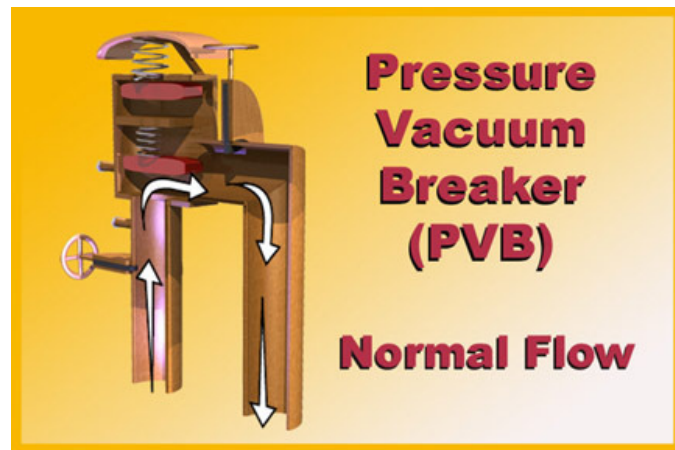
Atmospheric Vacuum Breakers

These devices are among the simplest and least expensive mechanical types of backflow preventers and, when installed properly, can provide excellent protection against back-siphonage. Water flow lifts the float, which then causes the disc to seal. Water pressure keeps the float in the upward sealed position. Termination of the water supply will cause the disc to drop down, thus preventing back-siphonage.



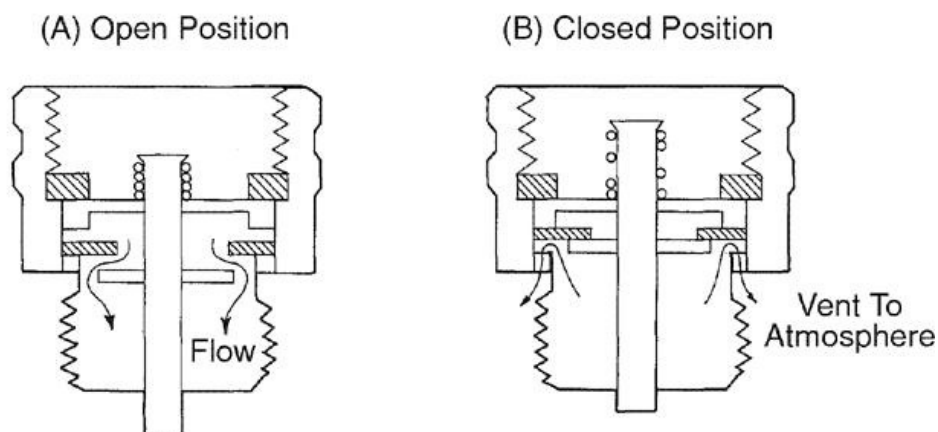
Pressure Vacuum Breaker

This device is an outgrowth of the atmospheric vacuum breaker and evolved in response to a need to have an atmospheric vacuum breaker that could be utilized under constant pressure and that could be tested in line. Just as with an atmospheric vacuum breaker, this device protects against back-siphonage, but not backpressure.



Hose Bibb Vacuum Breakers

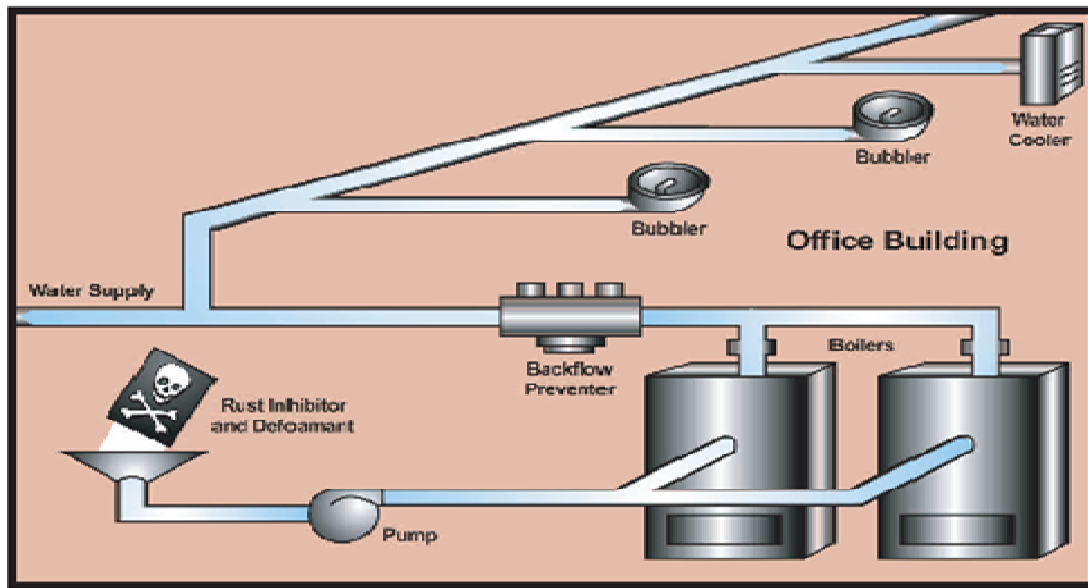
These are a specialized application of the atmospheric vacuum breaker. Generally attached to hose-supplied outlets, such as garden hoses, slop sink hoses, spray outlets, etc. When the water supply is turned off, the device vents to the atmosphere, thus preventing back-siphonage.



****Note: Single-check valves are NOT backflow prevention devices.****

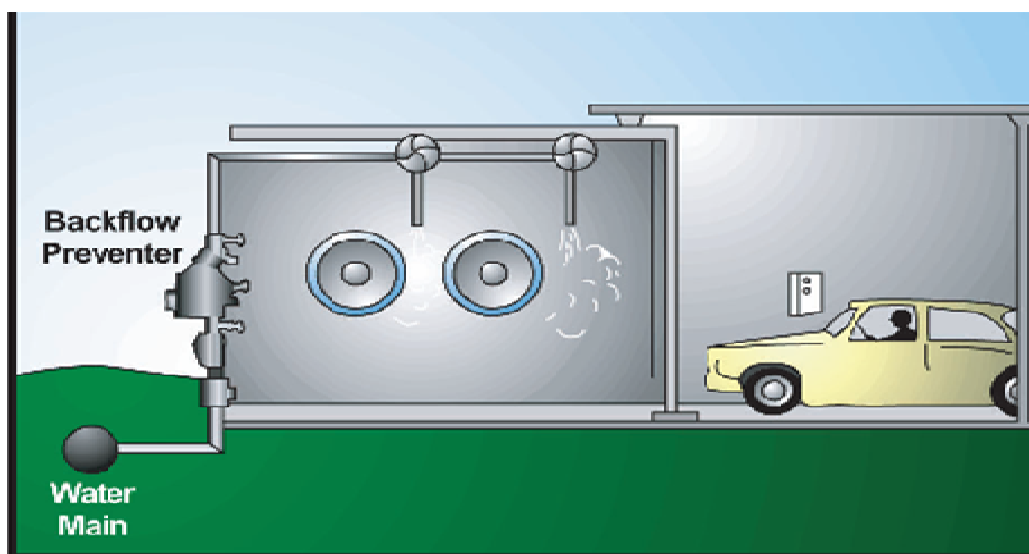
Two Types of Cross-connection Control Programs

1. **Isolation** protects the consumer at their tap.



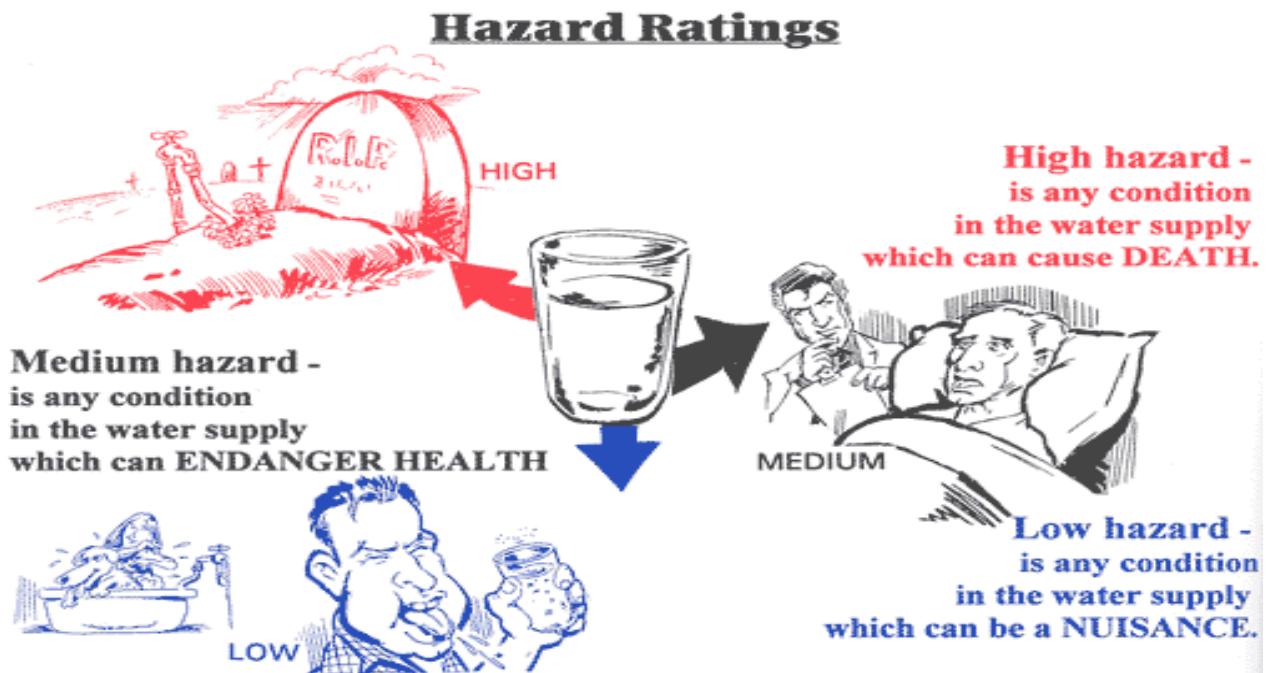
8.3 Figure 2: Isolation protects individual sites

2. **Containment** protects the potable water supply at the service connection.



8.3 Figure 1: Containment protects the public water system

Degree of Hazard

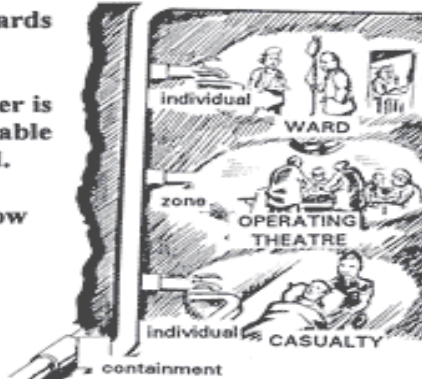


Selection of Backflow Prevention Devices

First it is necessary to identify the hazards within the property.

Upstream from these hazards, the water is considered as non-potable until a suitable backflow prevention device is installed.

It may be necessary to install a backflow prevention device for a fixture, a zone within a building, or at the boundary of the property.



Only the air gap, barometric loop and the reduced pressure zone backflow prevention devices are approved for high hazard applications.

Necessities for Implementing a Cross-connection Control Program

Authority to Implement a Program

- Conduct hazard surveys
- Install approved backflow prevention devices
- Annually test backflow devices
- Make necessary repairs

Certified Personnel

- Backflow assembly testers
- Trained CCC specialists

Defensible Records

- Assesses hazards through surveys and notate findings
- Retain annual test results

Education

- Internal
- Civic movers and shakers
- Public

Keep in mind:

- Backflow is real and needs to be addressed.
- Incidents are transient in nature.
- Hazards are both microbial and chemical.
- All water systems are susceptible.
- As a result of inaction by water purveyors, acute or fatal consequences could occur.

Chapter 13 Review

1. The single most important piece of a backflow protection program is a (n) _____.
2. The largest factor when determining the cause of waterborne disease and illness outbreaks is _____.
3. The difference between a RPZ and a double check is _____.
4. AVBs should be installed _____ the grade line.
5. PVBs protect against _____ only.

Answers for Chapter 13 Review

1. Ordinance
2. Backflow
3. The relief valve that sends the contaminated water to the atmosphere
4. Above
5. backsiphonage

Appendix

Certified Drinking Water Distribution Operator Regulations

401 KAR 11:001. Definitions for 401 KAR Chapter 11.

RELATES TO: KRS 223.160-220, 224.01-010(9), 224.73-110, EO 2009-538

STATUTORY AUTHORITY: KRS 223.200, 224.10-100, 224.10-110, 224.73-110, EO 2009-538

NECESSITY, FUNCTION, AND CONFORMITY: KRS 224.10-110 authorizes the cabinet to promulgate administrative regulations concerning the certification of wastewater operators. KRS 223.200 requires the cabinet to promulgate administrative regulations concerning the certification of water treatment and distribution system operators. EO 2009-538, effective June 12, 2009, establishes the new Energy and Environment Cabinet. This administrative regulation establishes definitions applicable to the certification of wastewater and water operators.

Section 1. Definitions. (1) "Applicant" means a person who has submitted an application to take an examination for certification.

(2) "Board" means:

(a) The Kentucky Board of Certification of Wastewater System Operators; or

(b) The Kentucky Board of Certification of Water Treatment and Distribution System Operators.

(3) "Cabinet" is defined by KRS 224.01-010(9).

(4) "Certificate" means a certificate of competency issued by the cabinet stating that the operator has met the requirements for the specified operator classification as established by 401 KAR Chapter 11.

(5) "Certified operator" means an individual that holds an active certified operator's certificate issued in accordance with 401 KAR 11:050.

(6) "Core content" means the information identified as essential by the board for purposes of certification examination and continuing education training.

(7) "Direct responsible charge" means personal, first-hand responsibility to conduct or actively oversee and direct procedures and practices necessary to ensure that the drinking water treatment plant or distribution system is operated in accordance with accepted practices and with KRS Chapters 223 and 224 and 401 KAR Chapters 8 and 11.

(8) "Operator" means a person involved in the operation of a wastewater treatment plant, wastewater collection system, drinking water treatment plant, or drinking water distribution system.

(9) "Primary responsibility" means personal, first-hand responsibility to conduct or actively oversee and direct procedures and practices necessary to ensure that the wastewater treatment plant or wastewater collection system is operated in accordance with accepted practices and with KRS Chapter 224 and 401 KAR Chapters 5 and 11. (35 Ky.R. 473; Am. 1210; eff. 3-6-2009; 36 Ky.R. 449; 1047; eff. 2-5-2010.)

401 KAR 11:010. Boards of certification.

RELATES TO: KRS 223.160-220, 224.73-110, EO 2009-538

STATUTORY AUTHORITY: KRS 223.160-200, 224.10-100, 224.10-110, 224.73-110, EO 2009-538

NECESSITY, FUNCTION, AND CONFORMITY: KRS 224.10-110 authorizes the cabinet to promulgate administrative regulations concerning the board of certification of wastewater system operators and the certification of wastewater

operators. KRS 223.160-220 authorizes the cabinet to promulgate administrative regulations concerning the board of certification for water treatment and distribution operators and the certification of water treatment and distribution system operators. EO 2009-538, effective June 12, 2009, establishes the new Energy and Environment Cabinet. This administrative regulation establishes the duties of the Kentucky Board of Certification of Wastewater System Operators and the Kentucky Board of Certification of Water Treatment and Distribution System Operators.

Section 1. Duties of the Board. The board shall:

- (1) Evaluate the qualifications of applicants and recommend qualified applicants to the cabinet for certification examination;
- (2) Review and provide comments to the cabinet on proposed administrative regulations regarding operator certification;
- (3) Review and make recommendations to the cabinet on core content for certification examinations and continuing education training for certification renewal;
- (4) Review and make recommendations to the cabinet on training proposed to provide continuing education to certified operators. During the evaluation of training courses and seminars, the board shall consider:
 - (a) The consistency of training material with the core content;
 - (b) The ability of the training to provide information that supports effective water conveyance, treatment, and quality; and
 - (c) The ability of the instructor to properly present the training;
- (5) Assist the cabinet in drafting examinations for the certification of operators;
- (6) Review and provide comments to the cabinet on proposed fees for the training and certification of operators;
- (7) Review applications for reciprocity and recommend to the cabinet the acceptance or denial of the application based on the criteria established in 401 KAR 11:050, Section 1(8); and
- (8) Review evidence and advise the cabinet regarding disciplinary actions for certified operators who fail to comply with KRS Chapters 223 and 224 or 401 KAR Chapter 5, 8, or 11. (35 Ky.R. 474; Am. 1211; 1746; eff. 3-6-2009; 36 Ky.R. 450; 1049; eff. 2-5-2010.)

401 KAR 11:020. Standards of professional conduct for certified operators.

RELATES TO: KRS 223.160-220, 224.73-110, EO 2009-538

STATUTORY AUTHORITY: KRS 224.10-100, 224.10-110, 224.73-110, EO 2009-538

NECESSITY, FUNCTION, AND CONFORMITY: KRS 224.10-110 authorizes the cabinet to promulgate administrative regulations concerning the certification of water and wastewater operators. EO 2009-538, effective June 12, 2009, establishes the new Energy and Environment Cabinet. This administrative regulation establishes standards for the performance of certified water and wastewater operator duties.

Section 1. Standards of Professional Conduct. (1) In order to safeguard the life, health, and welfare of the public and the environment and to establish and maintain a high standard of integrity in the certified operator profession, the following standards of professional conduct apply to persons certified in accordance with 401 KAR Chapter 11:

- (a) A certified operator shall, during the performance of operational duties, protect the safety, health, and welfare of the public and the environment;
- (b) A certified operator shall use reasonable care and judgment in the performance of operational duties;
- (c) If a certified operator's judgment is overruled by an employer under circumstances in which the safety, health, and welfare of the public or the environment are endangered, the certified operator shall inform the employer of the possible consequences;
- (d) A certified operator shall be objective, truthful, and complete in applications, reports, statements, and testimony provided to the cabinet; and
- (e) A certified operator shall ensure the integrity of the samples that the operator collects, prepares, or analyzes so

that results shall be a true representation of water quality.

(2) Proof of certification. While on duty, a certified operator shall carry the cabinet-issued wallet card showing the operator's current certification status.

(3) Maintenance of records. If information related to the operator's employment or mailing address changes from that provided in the application for certification, the certified operator shall provide written notification to the cabinet within thirty (30) days. (35 Ky.R. 475; Am. 1212; 1747; eff. 3-6-2009; 36 Ky.R. 452; 1051; eff. 2-5-2010.)

401 KAR 11:040. Water treatment and distribution system operators; classification and qualifications.

RELATES TO: KRS 223.160-220, 224.10-100, 224.10-110, EO 2009-538

STATUTORY AUTHORITY: KRS 223.160-220, 224.10-100, 224.10-110, EO 2009-538

NECESSITY, FUNCTION, AND CONFORMITY: KRS 224.10-100 and 224.10-110 authorize the cabinet to promulgate administrative regulations concerning the certification of water operators. EO 2009-538, effective June 12, 2009, establishes the new Energy and Environment Cabinet. This administrative regulation establishes classification of water treatment and distribution operator certifications and establishes the qualifications for certification.

Section 1. Classification of Water Operator Certifications. (1) Water treatment certifications.

(a) Limited certification. As provided in KRS 223.160(2), an operator issued a limited certificate may have primary responsibility for a water treatment facility for a school and for a semipublic water supply.

(b) Class IA-D treatment certification.

1. A Class IA-D treatment operator may be in direct responsible charge for a Class IA-D or Class IB-D water treatment plant as established in 401 KAR 8:030, Section 2.

2. A Class IA-D treatment operator shall not be in direct responsible charge for a water treatment plant with a larger design capacity.

(c) Class IB-D treatment certification.

1. A Class IB-D treatment operator may be in direct responsible charge for a Class IB-D water treatment plant, as established in 401 KAR 8:030, Section 2.

2. A Class IB-D treatment operator shall not be in direct responsible charge for a water treatment plant with a larger design capacity.

(d) Class IIA treatment certification.

1. A Class IIA treatment operator may be in direct responsible charge for a Class IIA water treatment plant or a Subclass A or B water treatment plant of an equal to or smaller design capacity, as established in 401 KAR 8:030, Section 2.

2. A Class IIA treatment operator shall not be in direct responsible charge for a water treatment plant with a larger design capacity.

(e) Class IIB-D treatment certification.

1. A Class IIB-D treatment operator may be in direct responsible charge for a Class IIB-D water treatment plant or a Subclass B water treatment plant of a smaller design capacity, as established in 401 KAR 8:030, Section 2.

2. A Class IIB-D treatment operator shall not be in direct responsible charge for a water treatment plant with a larger design capacity.

(f) Class IIIA treatment certification.

1. A Class IIIA treatment operator may be in direct responsible charge for a Class IIIA water treatment plant or a Subclass A or B water treatment plant of an equal to or smaller design capacity, as established in 401 KAR 8:030, Section 2.

2. A Class IIIA treatment operator shall not be in direct responsible charge for a water treatment plant with a larger design capacity.

(g) Class IIIB treatment certification.

1. A Class IIIB treatment operator may be in direct responsible charge for a Class IIIB water treatment plant or a Subclass B water treatment plant of a smaller design capacity, as established in 401 KAR 8:030, Section 2.

2. A Class IIIB treatment operator shall not be in direct responsible charge for a water treatment plant with a larger design capacity.

(h) Class IVA treatment certification.

1. A Class IVA treatment operator may be in direct responsible charge of any Subclass A or B water treatment plant, as established in 401 KAR 8:030, Section 2.

(i) Class IVB treatment certification.

A Class IVB treatment operator may be in direct responsible charge of any Subclass B water treatment plant, as established in 401 KAR 8:030, Section 2.

(2) Water distribution certifications.

(a) Class ID distribution certification.

1. A Class ID distribution operator may be in direct responsible charge for a Class ID water distribution system, as established in 401 KAR 8:030, Section 2.

2. A Class ID distribution operator shall not be in direct responsible charge for a water distribution system serving a larger population.

(b) Class IID distribution certification.

1. A Class IID distribution operator may be in direct responsible charge for a Class IID water distribution system, as established in 401 KAR 8:030, Section 2.

2. A Class IID distribution operator shall not be in direct responsible charge for a water distribution system serving a larger population.

(c) Class IIID distribution certification.

1. A Class IIID distribution operator may be in direct responsible charge for a Class IIID water distribution system, as established in 401 KAR 8:030, Section 2.

2. A Class IIID distribution operator shall not be in direct responsible charge for a water distribution system serving a larger population.

(d) Class IVD distribution certification.

1. A Class IVD distribution operator may be in direct responsible charge of any water distribution system.

(3) Bottled water certification. A bottled water operator may be in direct responsible charge for a bottled water system that bottles water for sale.

(4) Operator in Training designations.

(a) Except as provided in paragraph (c) of this subsection, a certified operator with an Operator in Training designation shall not serve in direct responsible charge of a water treatment plant or distribution system.

(b) A certified operator with an Operator in Training designation shall work under the direct supervision of a certified operator who works at the same facility and has obtained a certification level that is equal to or greater than the certification level required to serve in direct responsible charge of the facility.

(c) If a certified operator also has been issued a water treatment, distribution, or bottled water certification without an Operator in Training designation, the operator may serve in direct responsible charge a water treatment plant, distribution system or bottled water system as provided by this Section for the certifications that do not have an Operator in Training designation.

Section 2. Water Operator Qualifications: Experience, Education, and Equivalencies. An individual desiring to become a certified operator shall meet the following minimum qualifications prior to the cabinet approving the individual to take a certification examination as established in 401 KAR 11:050.

(1) The education and experience requirement for each class of water treatment certifications shall be as follows:

(a) Limited certification:

1. Education. A minimum level of education shall not be required.

2. Experience. A minimum level of experience shall not be required.

(b) Class IA-D treatment certification:

1. Education. A high school diploma or general education development (GED) certificate shall be required; and

2. Experience. One (1) year of acceptable operation of a Subclass A public water system, as established in 401 KAR

8:030, Section 2, with any design capacity

(c) Class IB-D treatment certification:

1. Education. A high school diploma or general education development (GED) certificate shall be required; and
2. Experience. One (1) year of acceptable operation of a Subclass A or B public water system, as established in 401

KAR 8:030, Section 2, with any design capacity shall be required.

(d) Class IIA treatment certification:

1. Education. A high school diploma or general education development (GED) certificate shall be required; and
2. Experience. Two (2) years of acceptable operation of a water treatment plant, with six (6) months in a Class IIA, IIIA, or IVA water treatment plant, as established in 401 KAR 8:030, Section 2, shall be required.

(e) Class IIB-D treatment certification:

1. Education. A high school diploma or general education development (GED) certificate shall be required; and
2. Experience. Two (2) years of acceptable operation of a public water system with six (6) months in a Class IA-D, II B-D, or higher water treatment plan, as established in 401 KAR 8:030, Section 2, shall be required.

(f) Class IIIA treatment certification:

1. Education. A high school diploma or general education development (GED) certificate shall be required; and
2. Experience. Three (3) years of acceptable operation of a public water treatment plant with one (1) year in a Class IIA, IIIA, or IVA water treatment plant, as established in 401 KAR 8:030, Section 2, shall be required.

(g) Class IIIB treatment certification:

1. Education. A high school diploma or general education development (GED) certificate shall be required; and
2. Experience. Three (3) years of acceptable operation of a public water treatment plant with one (1) year in a Class IIA, IIB-D, IIIA, IIIB, IVA, or IVB water treatment plant, as established in 401 KAR 8:030, Section 2, shall be required.

(h) Class IVA treatment certification:

1. Education. A baccalaureate degree in engineering, science, or equivalent shall be required; and
2. Experience. One (1) year of acceptable operation of a Class IIIA or IVA public water treatment plant, as established in 401 KAR 8:030, Section 2, shall be required.

(i) Class IVB treatment certification:

1. Education. A baccalaureate degree in engineering, science, or equivalent shall be required; and
2. Experience. One (1) year of acceptable operation of a Class IIIA, IIIB, IVA, or IV public water treatment plant shall be required.

(2) The educational and experience qualifications for water distribution certifications shall be as follows:

(a) Class ID distribution certification:

1. Education. A high school diploma or general education development (GED) certificate shall be required; and
2. Experience. One (1) year of acceptable operation of a water distribution system shall be required.

(b) Class IID distribution certification:

1. Education. A high school diploma or general education development (GED) certificate shall be required; and
2. Experience. Two (2) years of acceptable operation of a water distribution system shall be required.

Six (6) months of the required experience shall be in a water distribution system serving a population greater than or equal to 1,500.

(c) Class IIID distribution certification:

1. Education. A high school diploma or general education development (GED) certificate shall be required; and
2. Experience. Three (3) years of acceptable operation of a water distribution system shall be required.

One (1) year of the required experience shall be in a water distribution system serving a population greater than or equal to 1,500.

(d) Class IVD distribution certification:

1. Education. A baccalaureate degree in engineering, science, or equivalent shall be required; and
2. Experience. One (1) year of acceptable operation of a water distribution system serving a population greater than or equal to 15,000 shall be required.

(3) Bottled water certification. The educational and experience qualifications for bottled water certifications shall be as follows:

- (a) Education. A High school diploma or general education development (GED) certificate shall be required; and
- (b) Experience. One (1) year of acceptable operation of a bottled water system shall be required.
- (4) The educational and experience qualifications for Operator in Training designations shall be as follows:
 - (a) Class IA-D Treatment, Class IB-D Treatment, Class ID Distribution, and Bottled Water certifications.
 - 1. Education. A high school diploma or general education development (GED) certificate shall be required; and
 - 2. Experience. Experience shall not be required.
 - (b) All other applicants for the classifications identified in subsections 1(1) and 1(2) of this administrative regulation;
 - 1. Shall have successfully qualified for and passed the certification exam of the same type classification at one (1) level lower than the Operator in Training designation being pursued; and
 - 2. Shall not have been subject to disciplinary action as provided by 401 KAR 11:050, Section 4.
- (5) Substitutions. The cabinet shall allow the following substitutions for the qualifications established in subsections (1), (2), and (3) of this section:
 - (a) Education in environmental engineering; environmental technology; and biological, physical, or chemical sciences shall be substituted if the substitution does not exceed fifty (50) percent of the required experience.
 - 1. An associate degree shall be considered equivalent to two (2) years of experience.
 - 2. A baccalaureate degree shall be considered equivalent to four (4) years of experience.
 - 3. Education that did not result in a degree in a related field may be substituted for the required experience as follows:
 - a. Ten (10) contact hours, one (1) Continuing Education Unit, or one (1) post-secondary education quarter hour with a passing grade shall be considered equivalent to 0.022 years of experience.
 - b. One (1) postsecondary education semester hour with a passing grade shall be considered equivalent to 0.033 years of experience.
 - 4. Education applied to the experience requirements specified in subsections (1) and (2) of this section shall not be applied to the education requirement.
 - (b) Experience may be substituted for the educational requirement as follows:
 - 1. One (1) year of operational experience at a water system may substitute for one (1) year of education.
 - 2. The cabinet may allow partial substitution of the education requirement by experience in maintenance, laboratory analysis, or other work related to the collection, treatment, or distribution of drinking water or wastewater. To establish how much experience shall be accepted, the cabinet shall determine the degree of technical knowledge needed to perform the work and the degree of responsibility the applicant had in the operation of the system.
 - 3. Experience applied to the education requirement specified in subsections (1) and (2) of this section shall not be applied to the experience requirement.
 - (c) Water treatment and distribution experience may be substituted as follows:
 - 1. Two (2) years of distribution system experience shall be considered equivalent to one (1) year of treatment experience.
 - 2. One (1) year of treatment experience shall be considered equivalent to one (1) year of distribution system experience. (36 Ky.R. 1055; 1457; eff. 2-5-2010.)

401 KAR 11:050. Operator certification.

RELATES TO: KRS 223.160-220, 224.10-420(2), 224.73-110, EO 2009-538

STATUTORY AUTHORITY: KRS 223.160-220, 224.10-100, 224.10-110, 224.73-110, EO 2009-538

NECESSITY, FUNCTION, AND CONFORMITY: KRS 224.10-110 authorizes the cabinet to promulgate administrative regulations concerning the certification of water and wastewater operators. EO 2009-538, effective June 12, 2009, establishes the new Energy and Environment Cabinet. This administrative regulation establishes application and examination procedures; provisions relating to certificate issuance, renewal, and termination; reciprocity; training; and disciplinary actions.

Section 1. Application and Examination for Certification. (1) An individual desiring to become a certified operator shall

first meet the qualifications established in 401 KAR 11:030 or 11:040 and then pass an examination administered by the cabinet.

(2)(a) An applicant for certification shall complete the Registration Form for Exams and Training and Education and Experience Documentation Form and shall submit them and the certification application fee to the cabinet.

(b) In addition to the requirements of paragraph (a) of this subsection, an applicant desiring to obtain an Operator in Training designation shall submit a signed letter for a certified operator located at the facility where the applicant will work. The letter shall include:

1. A statement from the certified operator indicating that the certified operator shall oversee the work of the applicant seeking an Operator in Training designation;

2. A commitment that the certified operator shall serve as a mentor to the applicant seeking an Operator in Training designation as long as the applicant is under the certified operator's direct responsible charge;

3. Verification that the certified operator is not currently the mentor for any other individuals with an Operator in Training designation; and

4. Confirmation that the certified operator has obtained a certification level that is equal to or greater than the certification level required to serve in primary responsibility of the facility.

(c) An application shall not be submitted to the cabinet unless the applicant has met the qualifications for examination.

(3)(a) After receipt of the application items established in subsection (2) of this section, the cabinet, considering the recommendation of the board, shall determine if the applicant meets the qualifications established in 401 KAR 11:030 or 11:040.

(b) If the applicant meets the qualifications, the cabinet shall approve the application and notify the applicant of the scheduled exam date.

(4)(a) Upon the applicant's completion of the examination, the cabinet shall notify the applicant of the applicant's examination score.

(b) A score of at least seventy (70) percent shall be required to pass the examination.

(5)(a) The cabinet shall issue a certificate and a wallet card to an applicant who successfully passes the certification examination.

(b) The certificate and wallet card shall designate the certification classification for which the operator has demonstrated competency.

(6) An applicant who fails to pass an examination may apply to take the examination again by resubmitting the Registration Form for Exams and Training and the application fee to the cabinet.

(7)(a) An examination shall not be returned to the applicant, but results may be reviewed by the applicant with a member of the cabinet.

(b) A request for a review shall be submitted to the cabinet in writing.

(8) A certificate shall be issued in a comparable classification, without examination, to a person who holds a valid certificate in a state, territory, or possession of the U.S. if:

(a) The requirements for certification under which the certificate was issued are not less stringent than the requirements for certification established in KRS 223.160-220, 224.73-110, and 401 KAR Chapter 11; and

(b) The applicant submits an Application for Reciprocity form and the reciprocity fee to the cabinet.

(9)(a) A certified operator who holds an Operator in Training designation may upgrade the certification by removing the Operator in Training Designation without examination if the operator:

1. Has satisfied the requirements of Section 3(1)(a) and (b) of this administrative regulation;

2. Has acquired the minimum experience required for the certification being pursued as required by 401 KAR 11:030 or 11:040; and

3. Submits a letter from the certified operator who has served as the applicant's mentor during the Operator in Training period that recommends the removal of the Operator in Training designation.

(b) A certified operator with an Operator in Training designation who is unable to comply with the requirements established in paragraph (a) of this subsection shall apply for and retake the certification exam to upgrade the operator's certification.

Section 2. Duration of Certification. (1)(a) Wastewater certifications shall expire on June 30 of an odd-numbered year unless suspended, revoked, or replaced by a higher classification certificate before that date.

(b) Wastewater certifications issued on or after January 1 and on or before June 30 of an odd-numbered year shall expire on June 30 of the next odd-numbered year.

(2)(a) Water certifications shall expire on June 30 of an even-numbered year unless suspended, revoked, or replaced by a higher classification certificate before that date.

(b) Water certifications issued on or after January 1 and on or before June 30 of an even-numbered year shall expire on June 30 of the next even-numbered year.

(3)(a) An expired certification shall continue in force pending the administrative processing of a renewal if the certified operator has complied with the renewal requirements of Section 3 of this administrative regulation.

(b) A certification continued in accordance with this subsection shall remain fully effective and enforceable.

(4) A certification shall terminate if not renewed on or before December 31 of the year the certification expired.

Section 3. Continuing Education and Certification Renewal. (1) A certified operator who is not designated an Operator in Training may renew a certification without examination if the operator has:

(a) Accumulated the training hours required in subsection (5) of this section; and

(b) Submitted a completed Application for Certification Renewal form and the renewal fee to the cabinet or has renewed the certification electronically on the cabinet's Web site.

(2)(a) A certified operator seeking to renew a certification with an Operator in Training designation shall apply for and retake the certification exam as provided in Section 1 of this administrative regulation.

(b) The cabinet shall not approve an operator to take an exam to renew a certification with Operator in Training designation unless the applicant has accumulated the training hours required in subsection (5) of this section.

(3) If the Application for Certification Renewal form and the renewal fee are not received by the cabinet or submitted electronically by June 30 of the year the certification expires, a late renewal fee as established in 401 KAR 8:050, Section 3 or 11:060, Section 1 shall be paid.

(4)(a) A terminated certification shall not be renewed.

(b) An operator whose certification is terminated and who wishes to become recertified shall reapply for and pass an examination in accordance with Section 1 of this administrative regulation.

(5)(a) Prior to applying for certification renewal, a certified operator shall complete the required number of cabinet-approved training hours.

(b) A certified operator holding multiple wastewater certifications issued in accordance with this administrative regulation shall complete the required number of cabinet-approved training hours for the highest certificate held in lieu of completing the required number of continuing education hours required for each certificate.

(c) A certified operator holding multiple water certifications issued in accordance with this administrative regulation shall complete the required number of cabinet-approved training hours for the highest certificate held in lieu of completing the required number of continuing education hours required for each certificate.

(d) Hours earned prior to initial certification shall not count toward certification renewal.

(e) Wastewater training hours shall expire two (2) years from the date earned.

(f) Water training hours shall be completed for each renewal during the two (2) year period immediately prior to the certificate expiration date.

1. Certified operators with a Bottled Water, Limited, Class I or II Treatment, Collection, or Distribution certification shall complete twelve (12) hours of approved training; or

2. Certified operators with a Class III or IV Treatment, Collection, or Distribution certification shall complete twenty-four (24) hours of approved training.

(6)(a) A training provider seeking approval of certified operator training shall submit to the cabinet a completed Application for Approval of Courses for Continuing Education Credit form.

(b) Upon completion of the approved training, the provider shall submit to the cabinet a completed Continuing Education Activity Report form.

(c) A certified operator who has attended training that has not been submitted to the cabinet for approval may apply for training approval as established in paragraph (a) of this subsection.

(d) A certified operator who provides approved training shall receive hour-for-hour credit for actual instruction time.

(7)(a) Cabinet approval of training shall expire two (2) years following the date of approval.

(b) The cabinet, in consultation with the board, shall extend the approval expiration date if:

1. The provider requests the extension in writing; and
2. The training has not changed from the previous approval.

Section 4. Disciplinary Action. (1) A certified operator shall be subject to disciplinary action if the cabinet, in consultation with the board, determines that the certified operator has not satisfactorily performed the operator's duties in accordance with 401 KAR 11:020.

(2)(a) A written complaint received by the board or cabinet regarding a certified operator, unless duplicitous or frivolous, and violations of 401 KAR 11:020 that are identified by the cabinet shall be evaluated by the board.

(b) The certified operator shall appear before the board if requested by the board

(3) The board shall make a recommendation to the cabinet regarding disciplinary action. The board may recommend that disciplinary action not taken or recommend that a disciplinary action be taken if the board determines that the certified operator has not satisfactorily performed operator duties in compliance with 401 KAR 11:020.

(4)(a) Upon receiving a recommendation from the board, the cabinet shall review the available evidence.

(b) After completing the review, the cabinet shall initiate the recommended disciplinary action or notify the board as to why an alternative disciplinary action was taken.

(5) A disciplinary action shall be commensurate with the severity, duration, and number of the violations. Disciplinary actions may include:

(a) Probation of the operator's certification for a specified period of time, not to exceed one (1) year;

(b) Suspension of the operator's certification for a specified period of time, not to exceed four (4) years, during which the certification shall be considered void;

(c) Revocation of the operator's certification;

(d) Civil or criminal penalties; or

(e) A combination of the disciplinary actions established in paragraphs (a) through (d) of this subsection.

(6) If disciplinary action is taken, the cabinet shall notify the certified operator and the operator's employer by certified mail of the action, the reasons outlined for the action, and the length of time for which the disciplinary action shall apply.

(7)(a) A certified operator whose certification has been suspended shall not have primary responsibility during the period that the suspension remains in effect.

(b) Experience gained during a suspension shall not be included toward meeting the requirements of 401 KAR 11:030 or 11:040.

(8) If a certification is revoked, the operator shall be ineligible for future certification.

(9) A certified operator who is aggrieved by a disciplinary action may file a petition for hearing with the cabinet pursuant to KRS 224.10-420(2).

Section 5. Incorporation by Reference. (1) The following material is incorporated by reference:

(a) "Registration Form for Exams and Training", August 2009;

(b) "Education and Experience Documentation Form", July 2009;

(c) "Application for Certification Renewal", August 2009;

(d) "Application for Approval of Courses for Continuing Education Credit", August 2009;

(e) "Continuing Education Activity Report", August 2009; and

(f) "Application for Reciprocity", July 2009.

(2) This material may be inspected, copied, or obtained, subject to applicable copyright law, at the Division of Compliance Assistance, 300 Fair Oaks Lane, Frankfort, Kentucky 40601, Monday through Friday, 8 a.m. to 4:30 p.m. (35 Ky.R. 479; Am. 1216; eff. 3-6-2009; eff. 2-5-2010.)

401 KAR 8:050. Drinking water program fees.

RELATES TO: KRS 223.220, 224.10-100, 224.10-110, 40 C.F.R. 142.10, Pub. L. 104-182, EO 2008-507, 2008-531

STATUTORY AUTHORITY: KRS 223.220, 224.10-100(20), 224.10-110

NECESSITY, FUNCTION, AND CONFORMITY: KRS Chapter 224.10-100(20) authorizes the cabinet to establish, by administrative regulation, a fee or schedule of fees for the cost of processing applications for permits authorized by KRS Chapter 224. EO 2008-507 and 2008-531, effective June 16, 2008, abolish the Environmental and Public Protection Cabinet and establish the new Energy and Environment Cabinet. This administrative regulation establishes fees for reviewing plans and specifications of public water systems, for operator certification, and for laboratory certification.

Section 3. (1) Fees for certification of water treatment plant and water distribution system operators.

(a) Certification application fee shall be \$100.

(b) A renewal application fee shall be:

1. Fifty (50) dollars if renewed through the cabinet Web site; or
2. \$100 if not renewed through the cabinet Web site.

(c) A renewal late fee shall be \$250.

(d) A reciprocity fee shall be \$500.

(2) Each year the cabinet, in consultation with the board, shall establish fees for operator training conducted by the cabinet.

(3)(a) The fees established in subsection (1) of this section shall be nonrefundable.

(b) Fifty (50) percent of the fees established in subsection (2) of this section shall be refundable if registration is canceled at least two (2) business days prior to the beginning of the training event.

(c) The fees in subsection (2) of this section shall be fully refunded if the training event is cancelled by the cabinet. (17 Ky.R. 599; eff. 11-15-1990; 35 Ky.R. 1847; eff. 7-6-2009.)

ACKNOWLEDGEMENTS

Lightmypump.com

The Pump Guy.com

Water Distribution System Operation and Maintenance – Office of Water Programs, CSU Sacramento
AWWARF #2661

Stop Backflow News, Case Histories and Solutions

EPA Cross-Connection Manual

AWWA Operational Techniques for Distribution Systems

The Drinking Water Handbook Frank R. Spellman & Joanne Drinan

AWWA Distribution Main Flushing and Cleaning

How Stuff Works.com

Center for Biofilm Engineering at MSU – Bozeman

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